

# EVOLVED STARS (II)

## LINE SURVEYS IN THE ALMA ERA

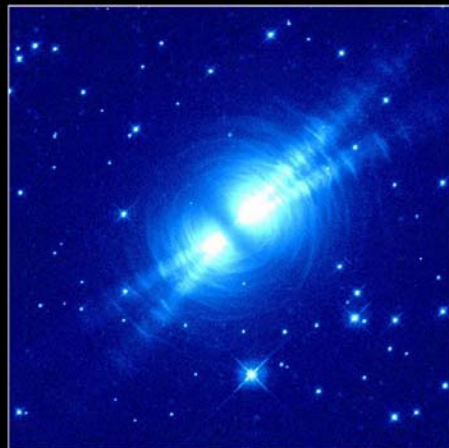
José Cernicharo

[jcernicharo@cab.inta-csic.es](mailto:jcernicharo@cab.inta-csic.es)

CAB (CSIC-INTA)

Dpt. Astrophysics

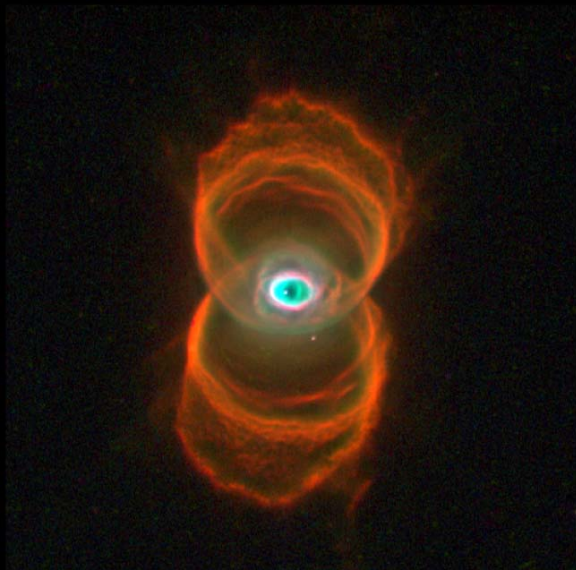
Madrid. Spain



Egg Nebula • CRL 2688

HST • WFPC2

PRC96-03 • ST ScI OPO • January 16, 1996  
R. Sahai and J. Trauger (JPL), the WFPC2 Science Team and NASA

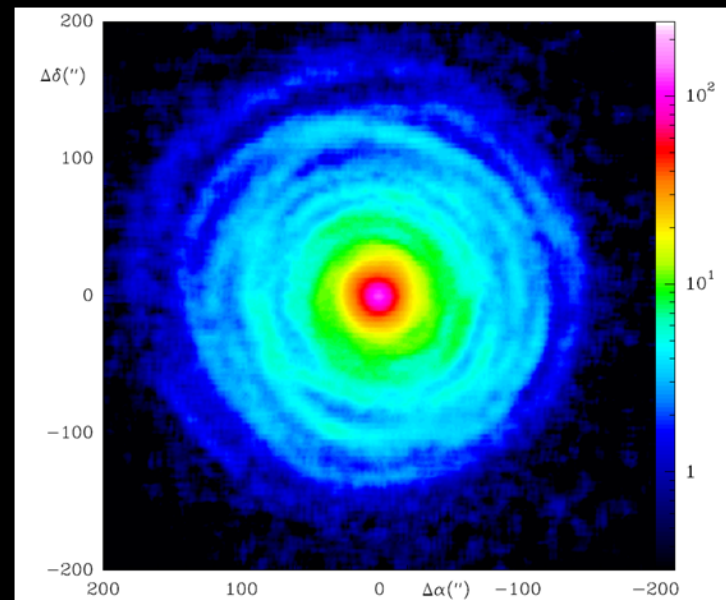
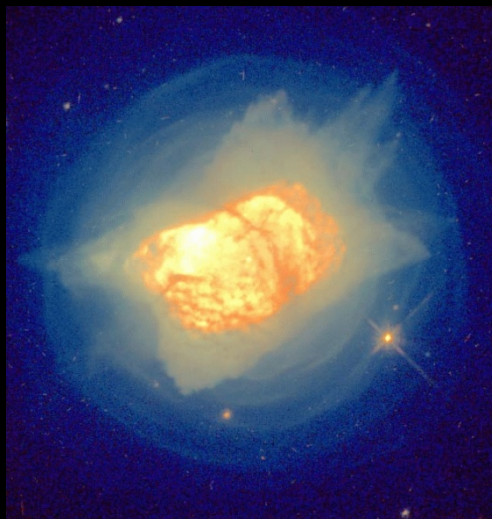


Hourglass Nebula • MyCn18

HST • WFPC2

PRC96-07 • ST ScI OPO • January 16, 1996

R. Sahai and J. Trauger (JPL), the WFPC2 Science Team and NASA



# The case of IRC+10216

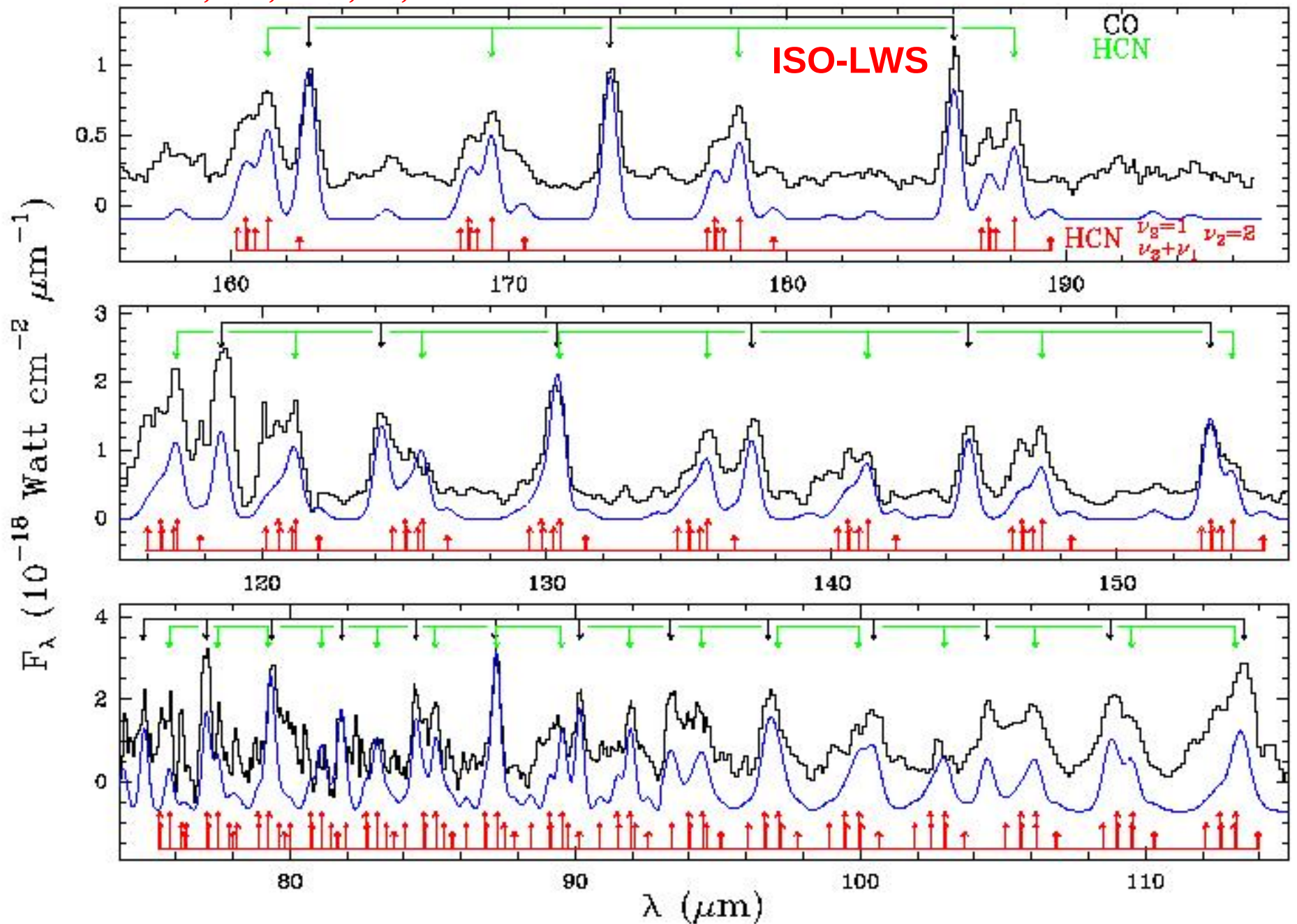
## ANATOMY OF AN AGB STAR

- THE DUST FORMATION ZONE
- THE INTERMEDIATE ENVELOPE
- THE EXTERNAL LAYERS
- PHOTOCHEMISTRY, SHOCKS AND  
THE ROLE OF BINARITY IN SHAPING  
THE CIRCUMSTELLAR ENVELOPE

# **The Physical and Chemical Conditions in the dust formation zone in IRC+10216**

*The first step to understand the  
chemistry of the whole envelope*

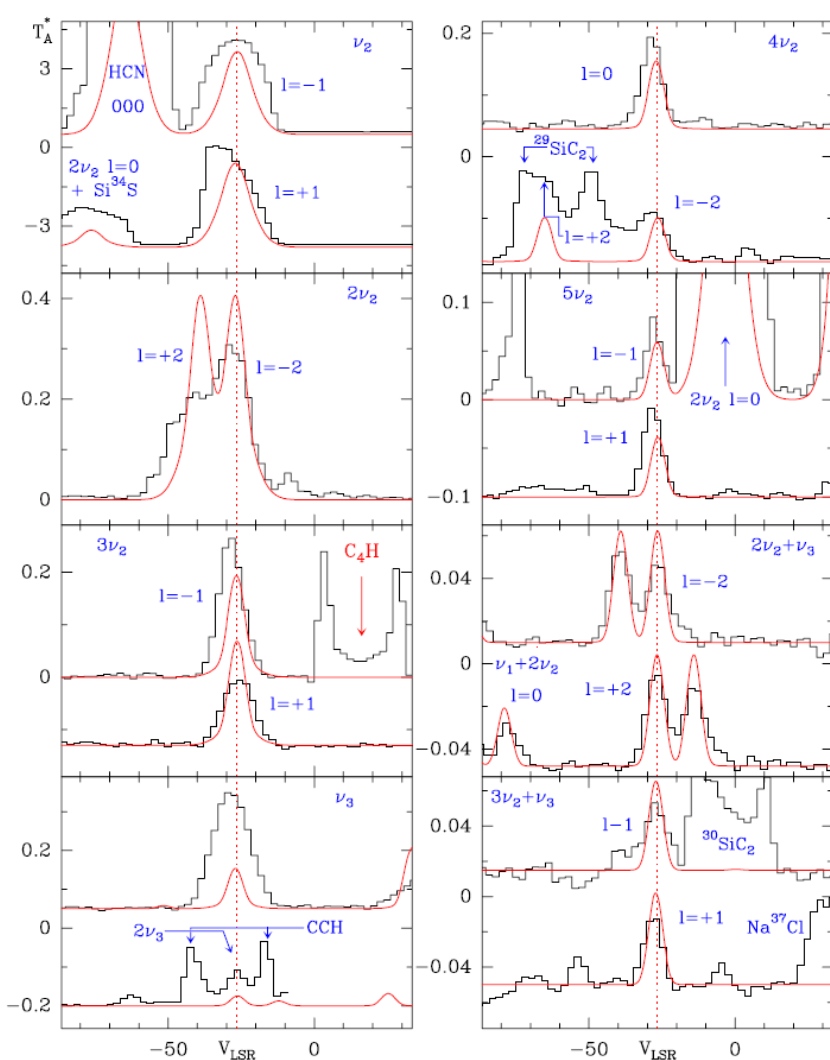
*Tools: Line Surveys in the Mid and  
Near Infrared with ISO & Ground  
Based Telescopes*



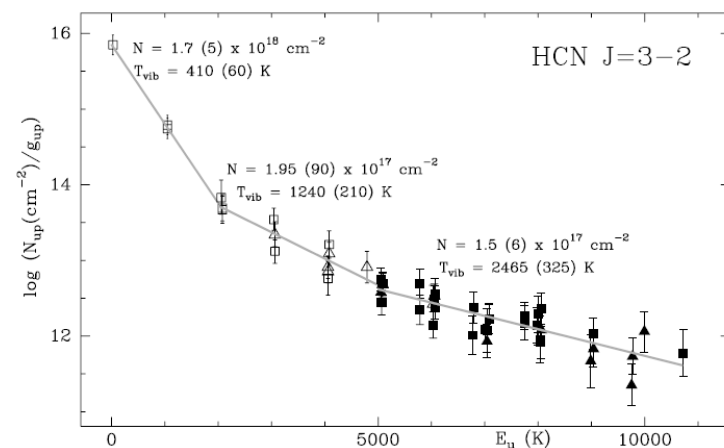
IRC+10216 a C-rich Circumstellar Envelope :: HCN (in all  $\nu$ ) + CO;  $\lambda/\Delta\lambda \sim 300$

## Probing the dust formation region in IRC +10216 with the high vibrational states of hydrogen cyanide<sup>★,★★</sup>

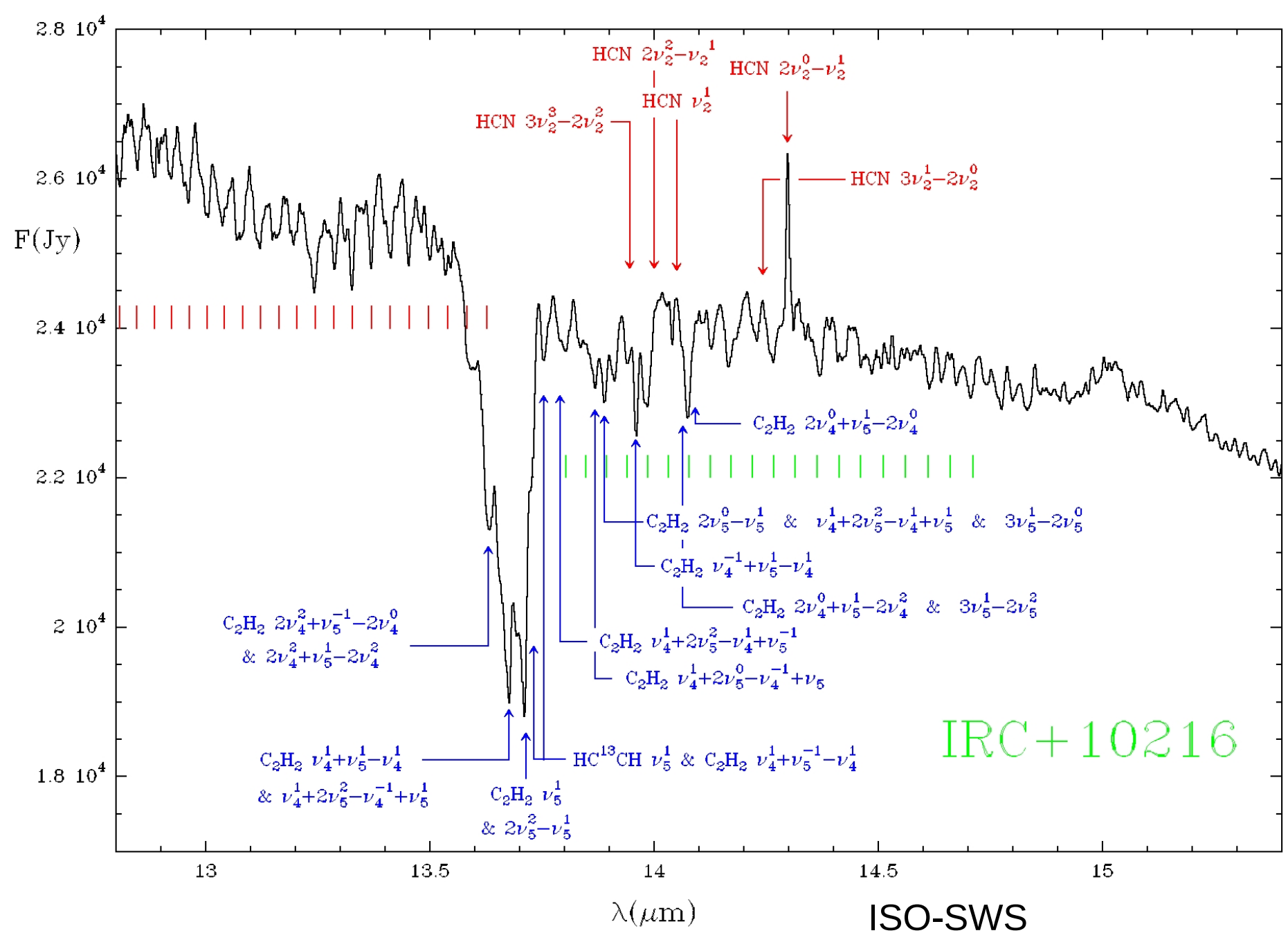
J. Cernicharo<sup>1</sup>, M. Agúndez<sup>2</sup>, C. Kahane<sup>3</sup>, M. Guélin<sup>4</sup>, J. R. Goicoechea<sup>1</sup>, N. Marcelino<sup>1</sup>, E. De Beck<sup>5</sup>, and L. Decin<sup>5</sup>

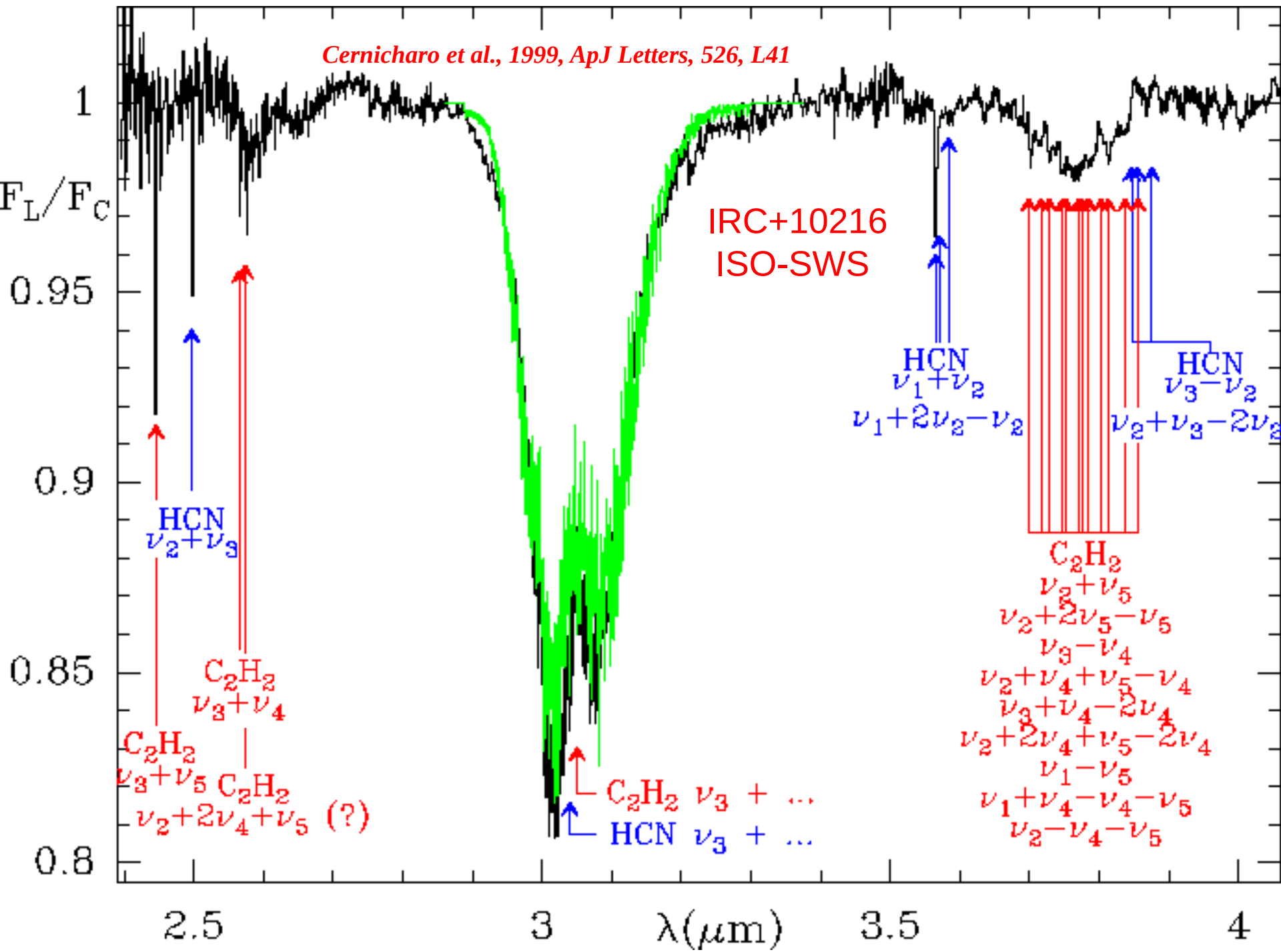


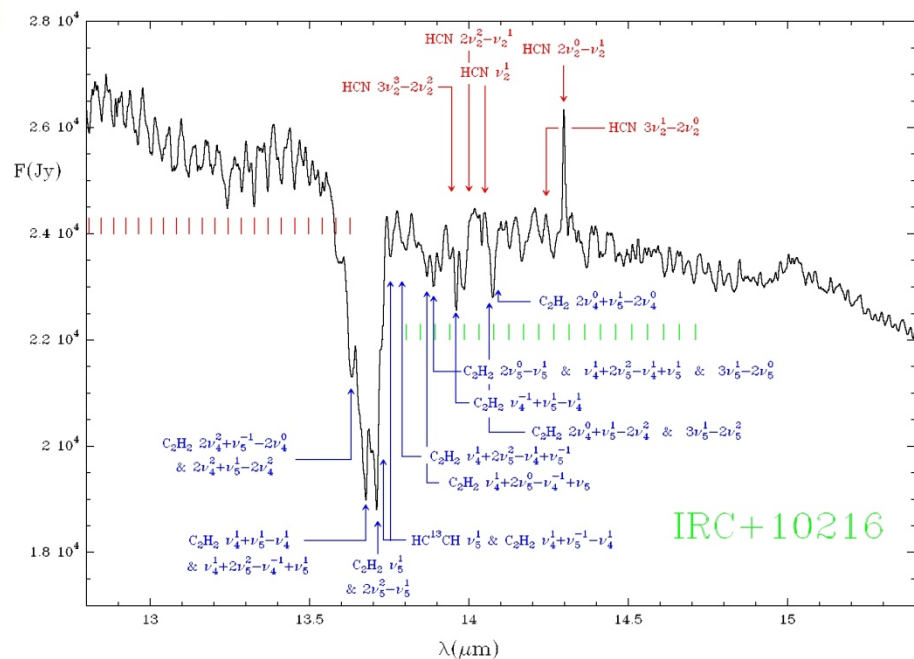
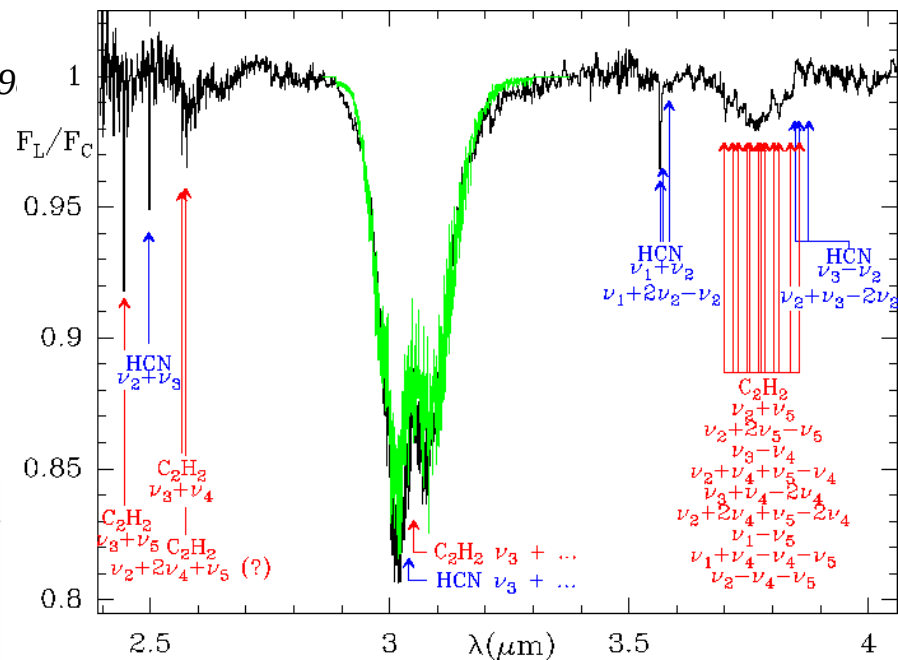
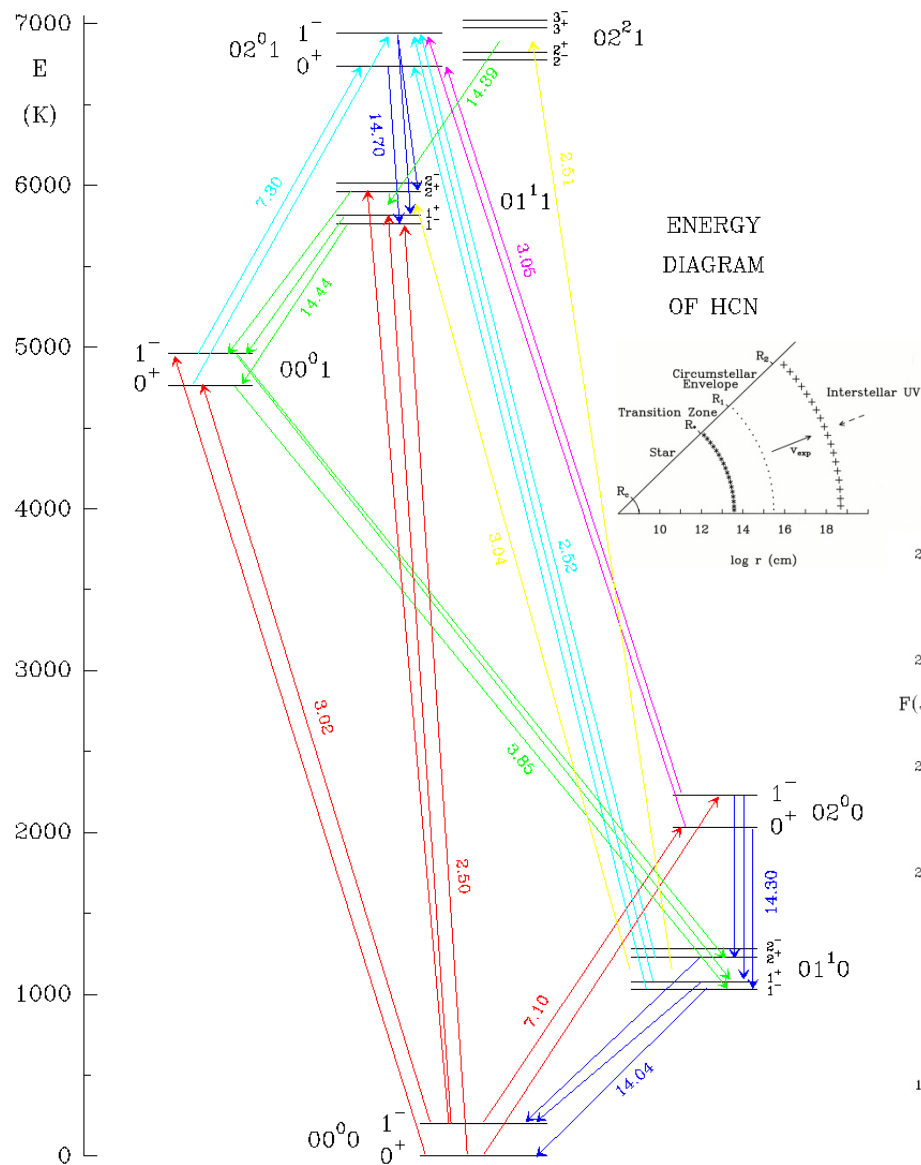
**Fig. 1.** Selected  $J = 3-2$  lines of HCN in various vibrational states (blue labels). The spectra observed with the IRAM 30-m telescope are indicated by black histograms. The line profiles calculated from our model (see text) are plotted in red. The line parameters are given in Table 1. Intensity scale is antenna temperature. Main beam antenna temperatures can be obtained by dividing the  $T_{A^*}$  scale of the 270 GHz data by a factor of 0.4. Velocity resolution is  $2.3 \text{ km s}^{-1}$ .



**Fig. 2.** Vibrational temperature diagram derived from the HCN  $J = 3-2$  lines given in Table 1. Upper level population vs. its energy above the ground state. The data is fitted by 3 line segments corresponding to regions with temperatures  $T_{\text{vib}} \sim 400 \text{ K}$ ,  $1200 \text{ K}$ , and  $2400 \text{ K}$ , respectively. HCN column densities are averaged over a source size of  $1''$  diameter. A conservative calibration error of 30% has been applied to all intensities. Values in parentheses correspond to  $3\sigma$  errors on the derived vibrational temperatures and column densities. Filled triangles correspond to lines with half-power widths  $\Delta v < 7 \text{ km s}^{-1}$ , filled squares correspond to lines with  $7 < \Delta v < 9 \text{ km s}^{-1}$ , empty triangles correspond to lines with  $9 < \Delta v < 12 \text{ km s}^{-1}$ , and empty squares to lines with  $\Delta v > 12 \text{ km s}^{-1}$ .







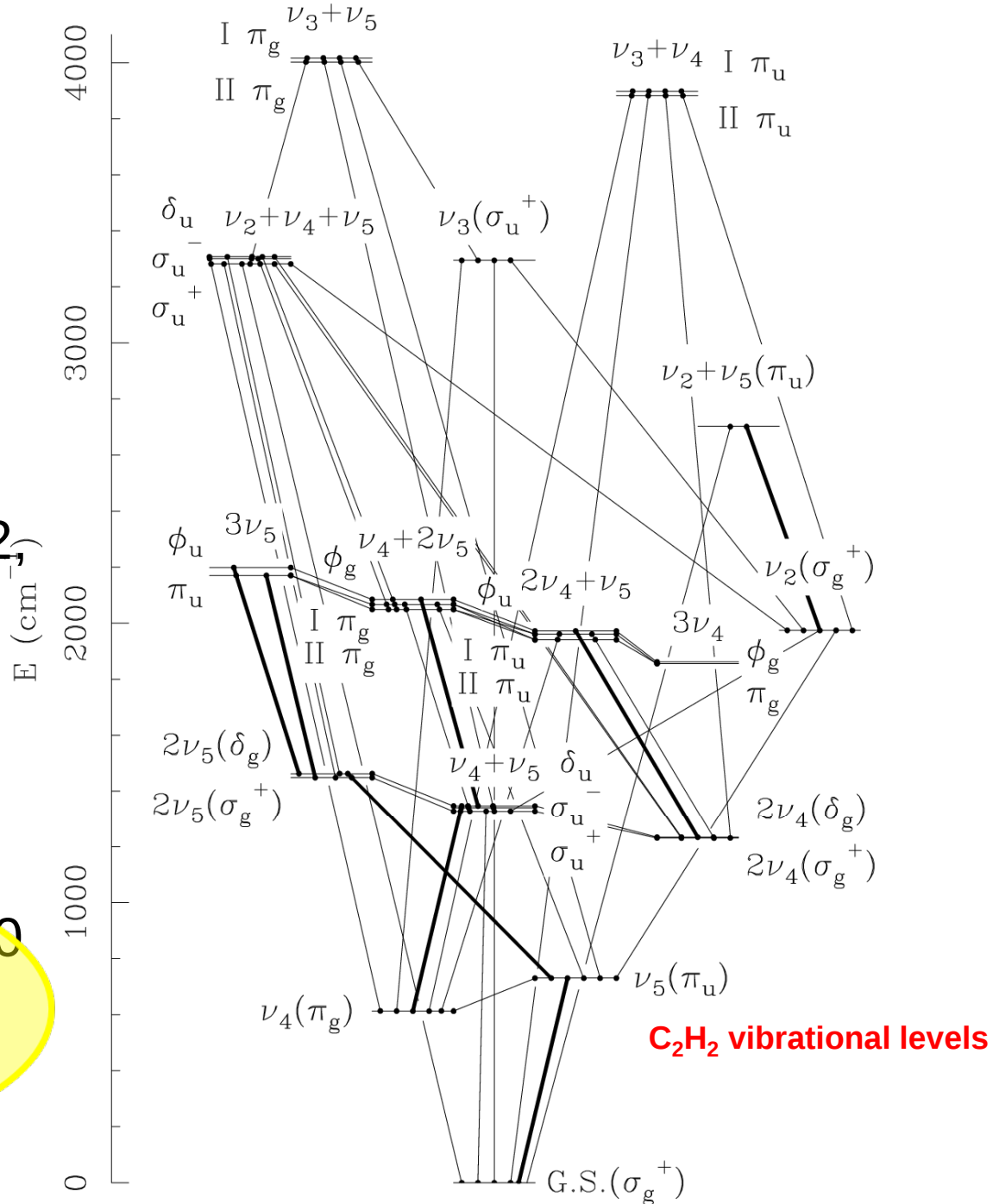
# Observations of $C_2H_2$ and HCN observations with TEXES at the IRTF

PhD of J.Pablo Fonfría  
(Fonfría et al., 2008, ApJ)

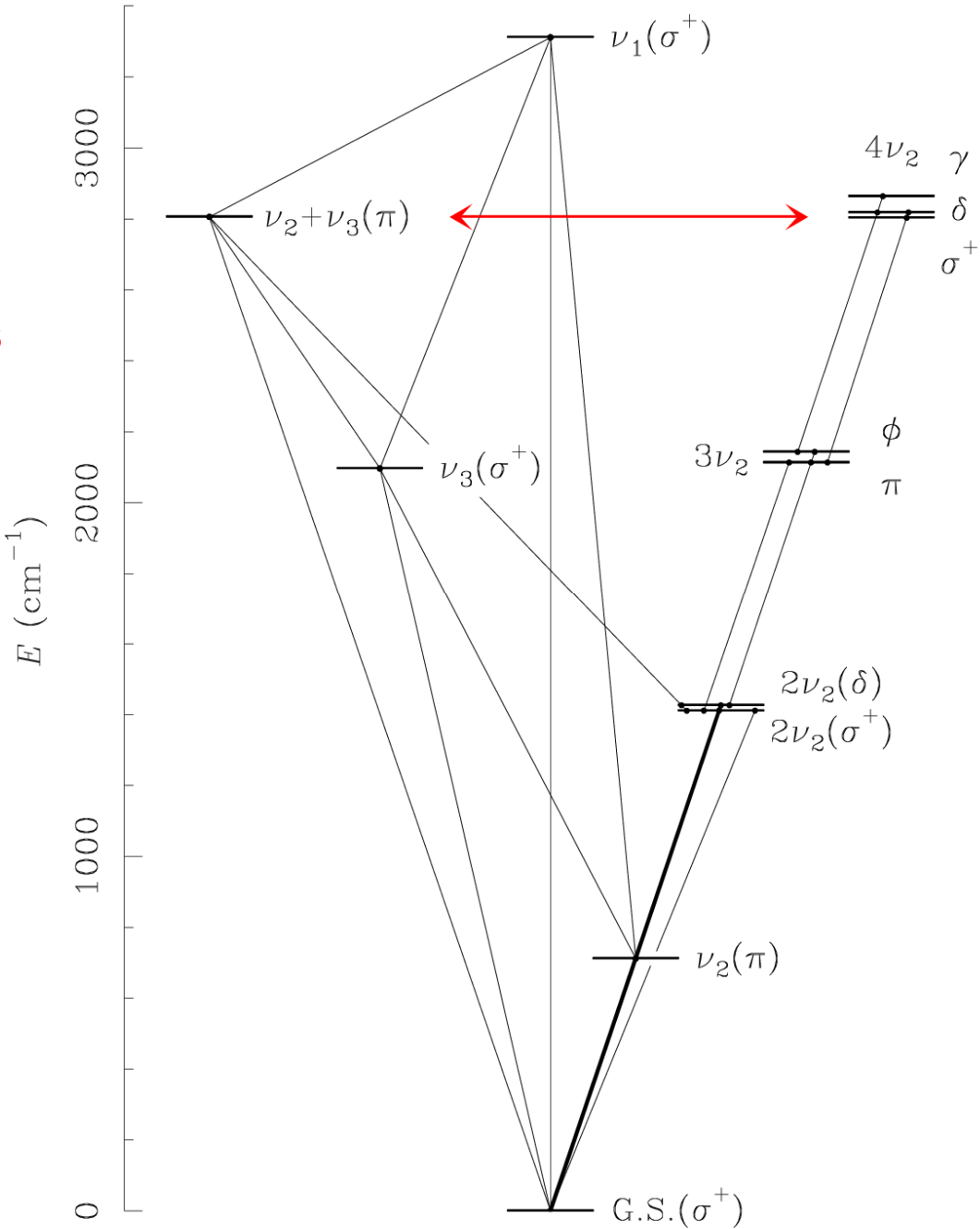
TEXES::Lacy et al., 2002,  
PASP, 114, 153

720-900  $cm^{-1}$   
11.1-13.9  $\mu m$

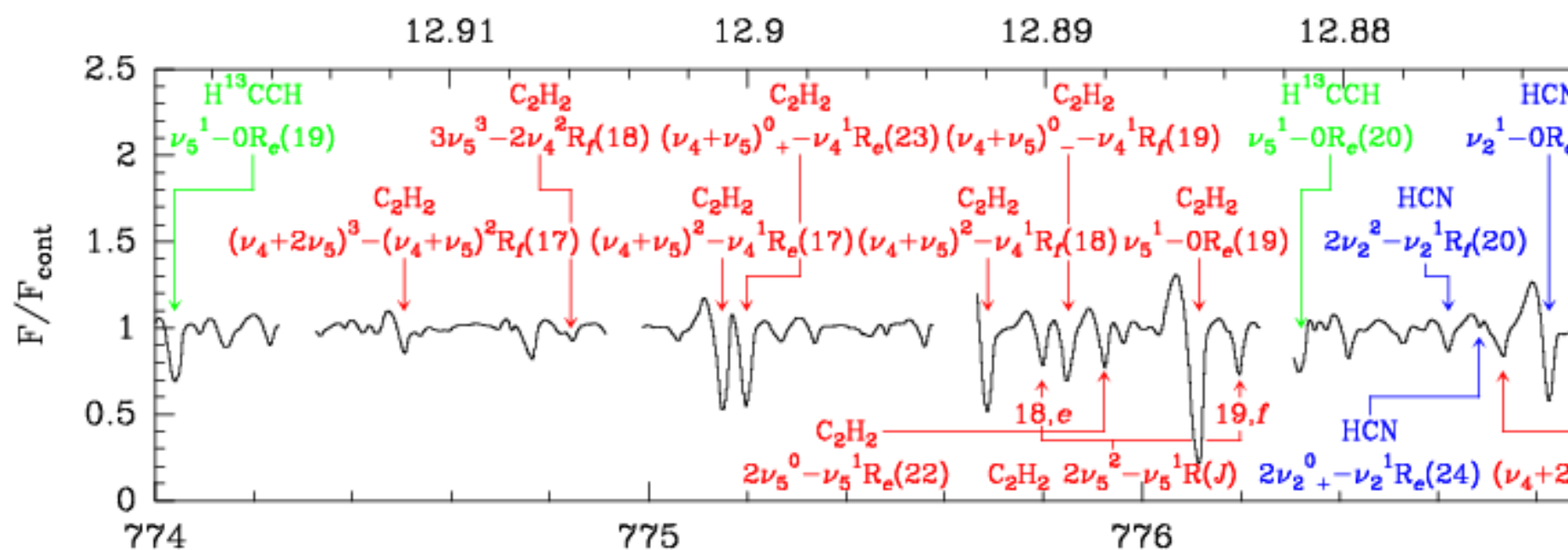
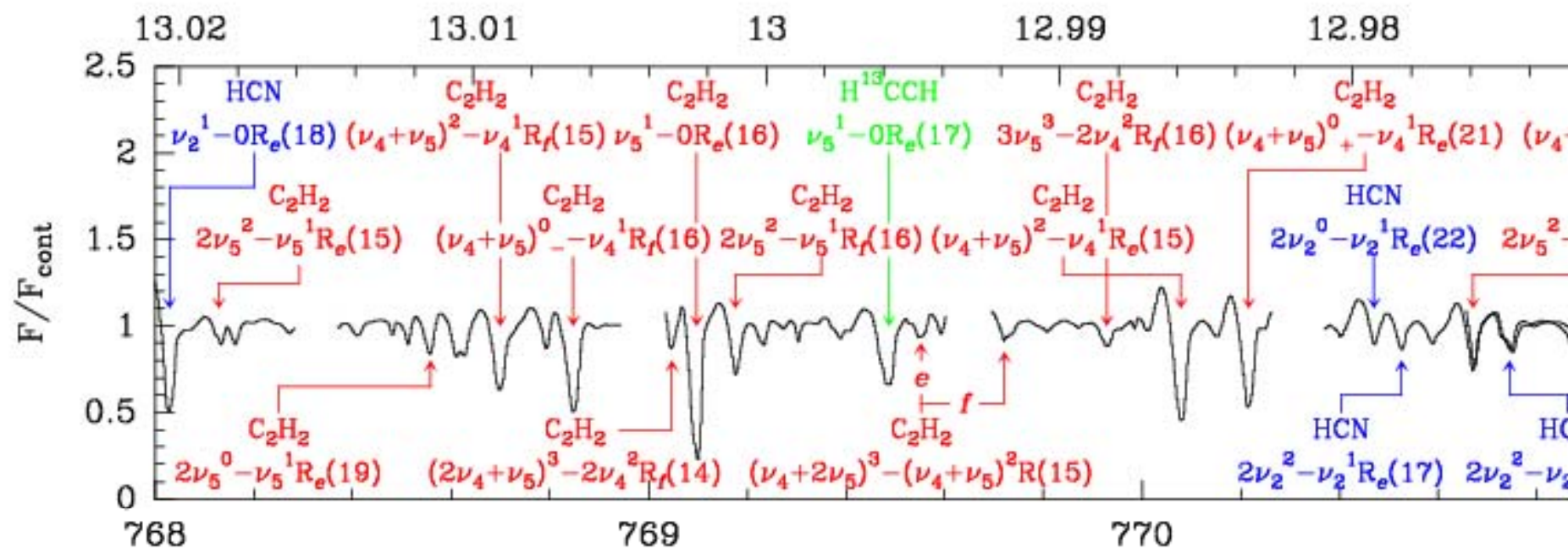
Spectral resolution 75000  
4 km/s

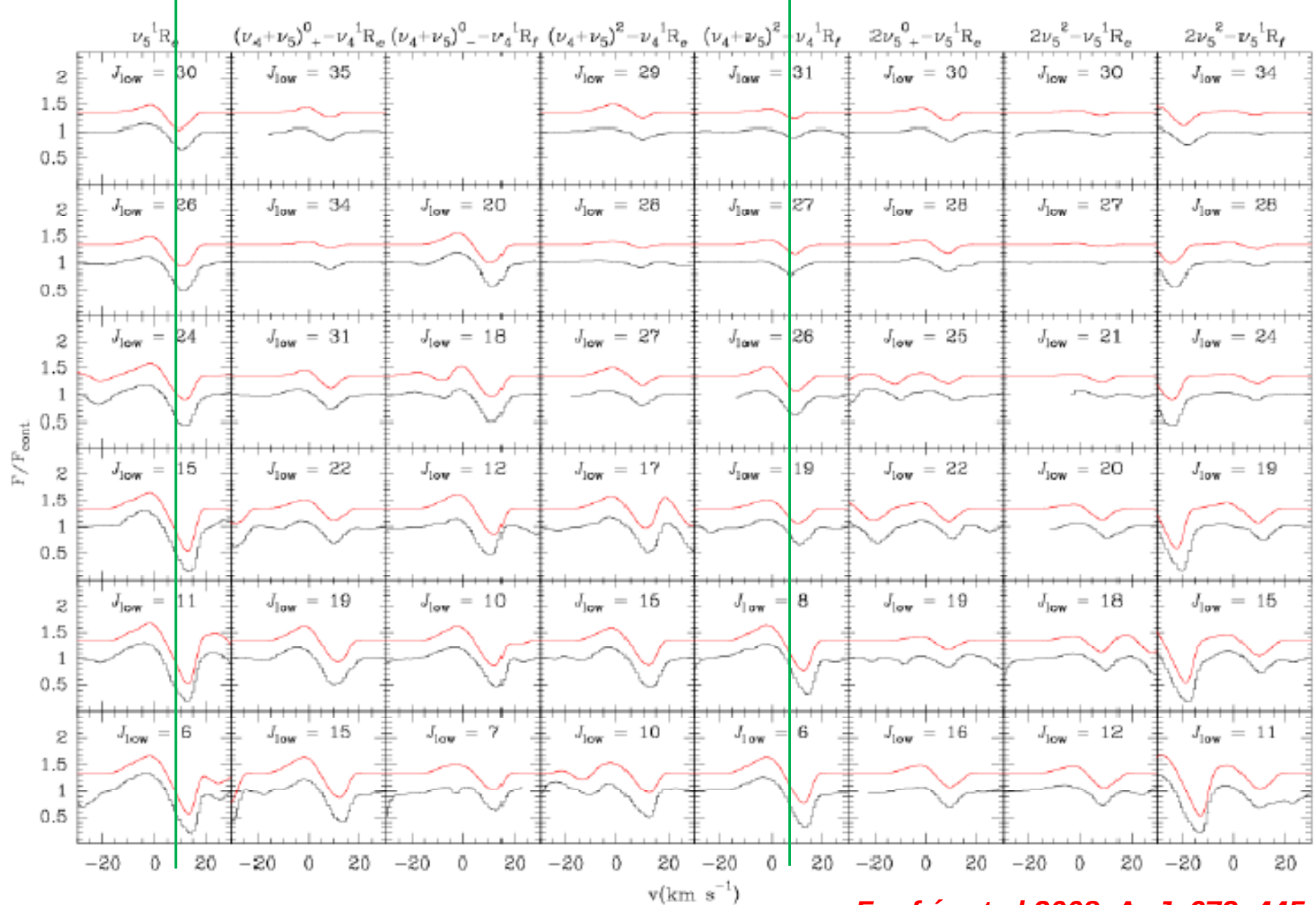


HCN vibrational levels









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DATA

—

MODEL  $\text{C}_2\text{H}_2 + \text{HCN} + \text{ISOTOPES}$

*Fonfría et al 2008, ApJ, 673, 445*

# RESULTS

## x(C<sub>2</sub>H<sub>2</sub>)

Z I 7.5 10<sup>-6</sup>

Z II 8.0 10<sup>-5</sup>

Z III 8.0 10<sup>-5</sup>

## x(HCN)

Z I 2.5 10<sup>-5</sup>

Z II 5.0 10<sup>-5</sup>

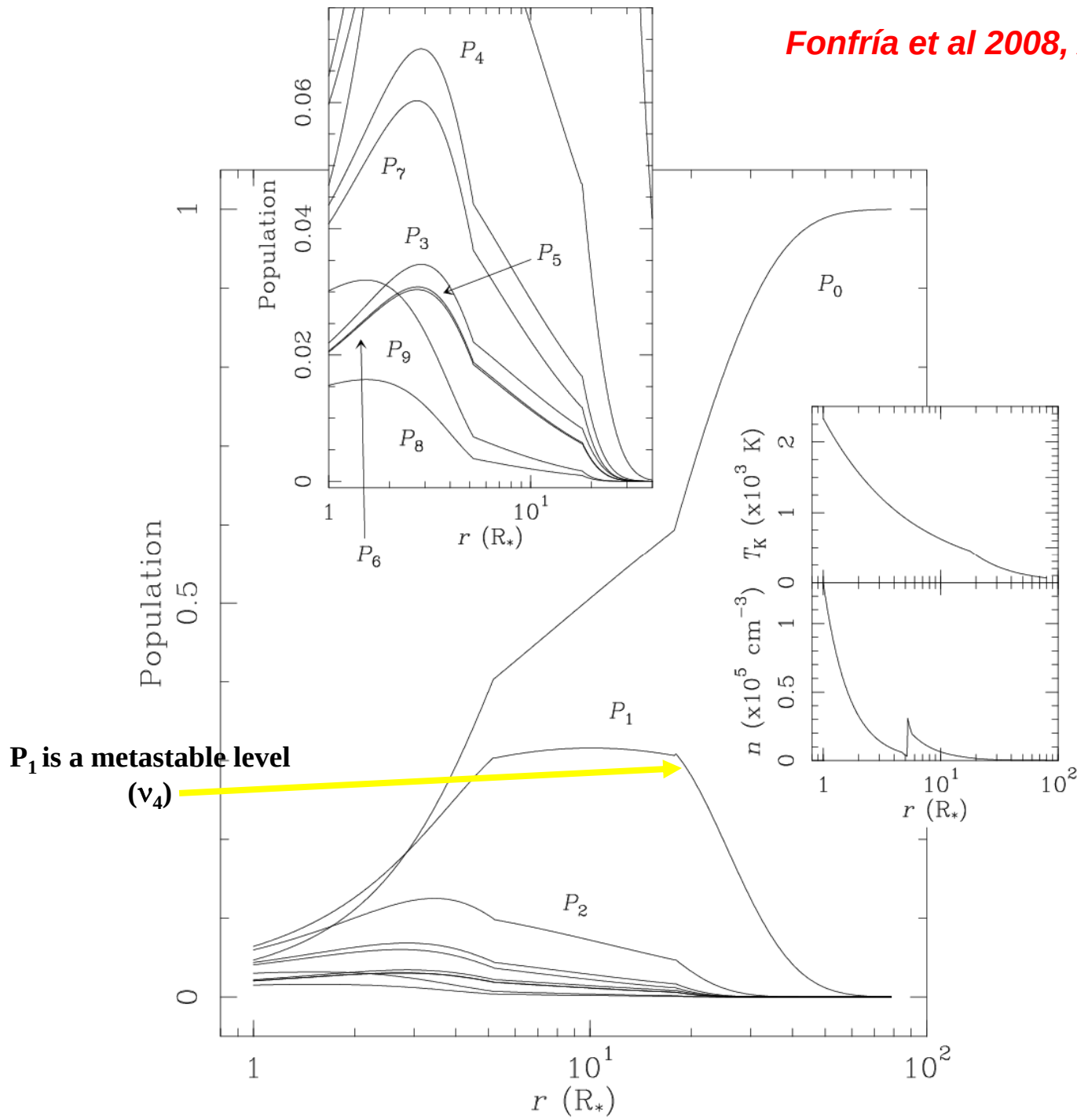
Z III 5.0 10<sup>-5</sup>

<sup>12</sup>C/<sup>13</sup>C=41

Z I 1- 5 R<sub>\*</sub>

Z II 5-20 R<sub>\*</sub>

Z III > 20 R<sub>\*</sub>

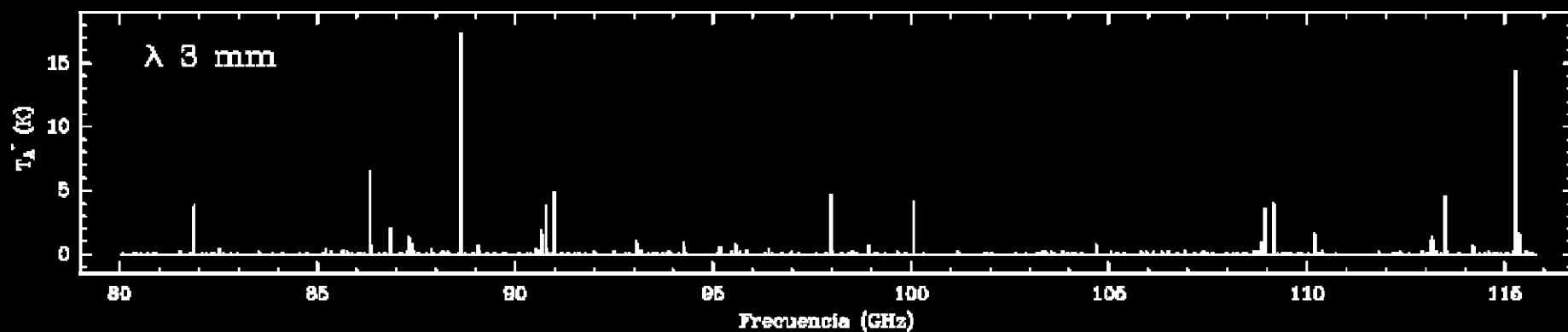
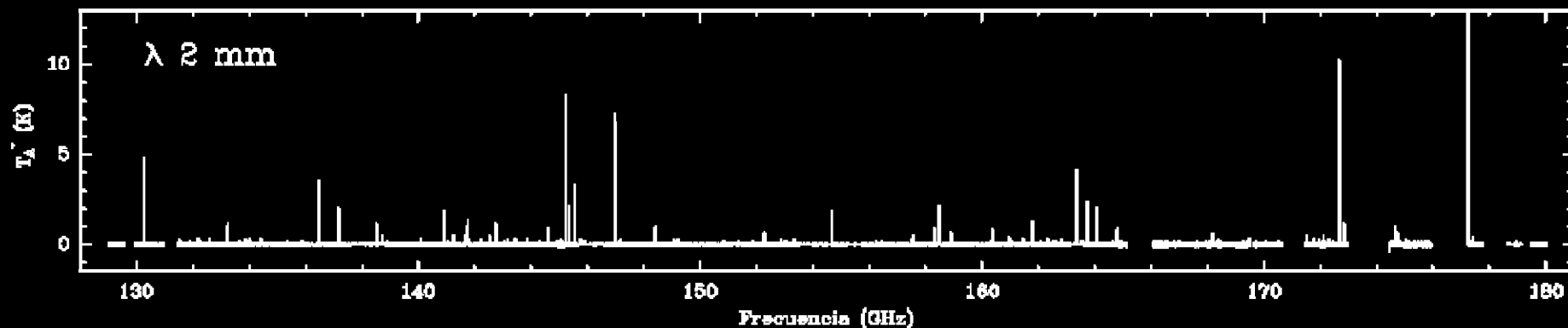
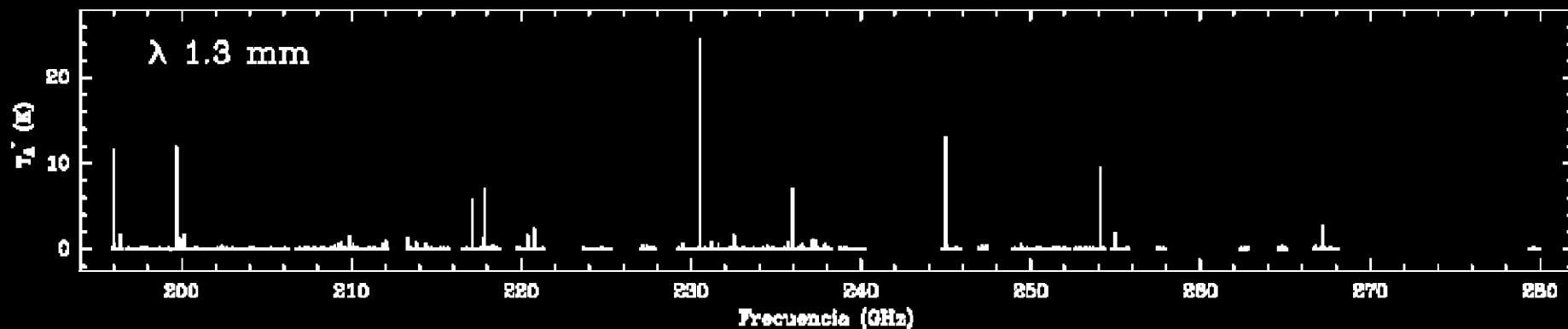


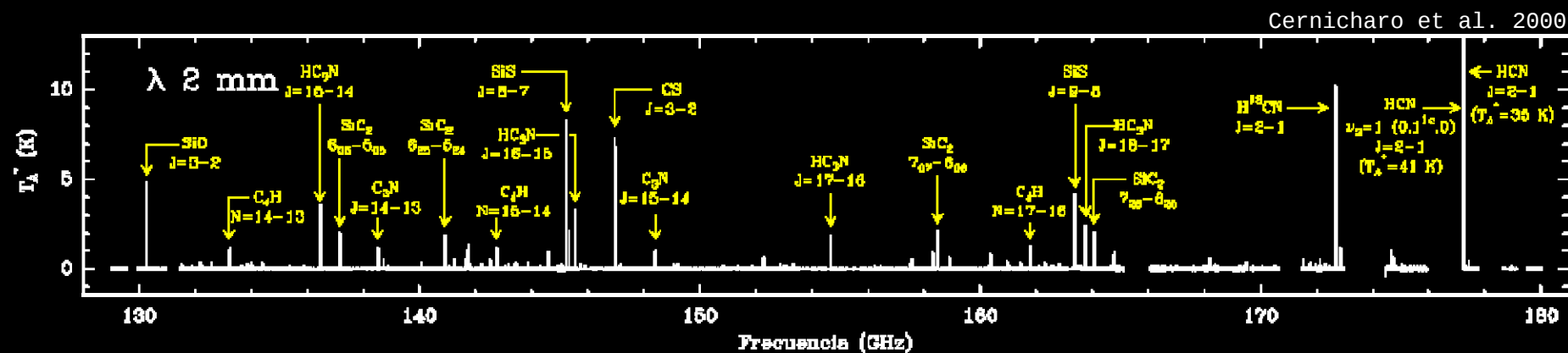
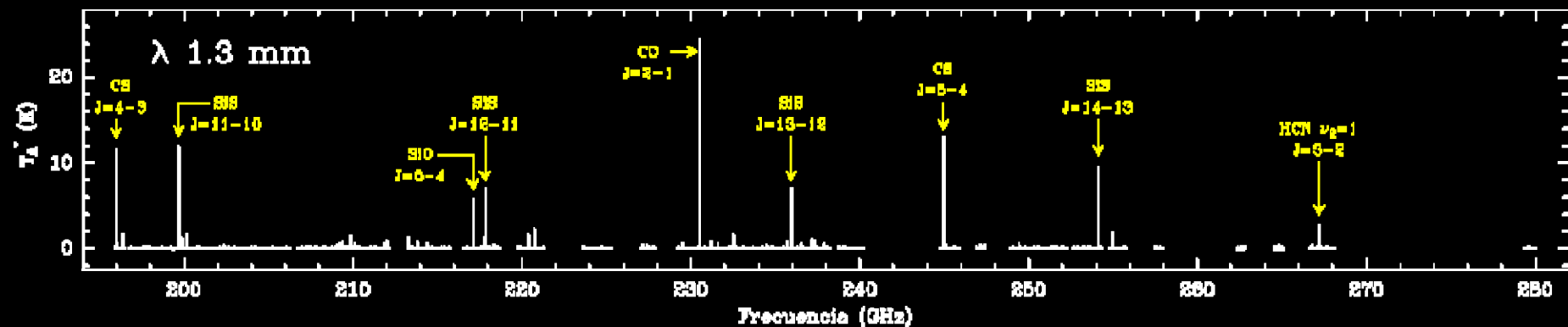
**The radio data**  
**The external shell (radicals and other exotic molecules)**

**Chemical Modelling : PhD of Marcelino Agúndez**

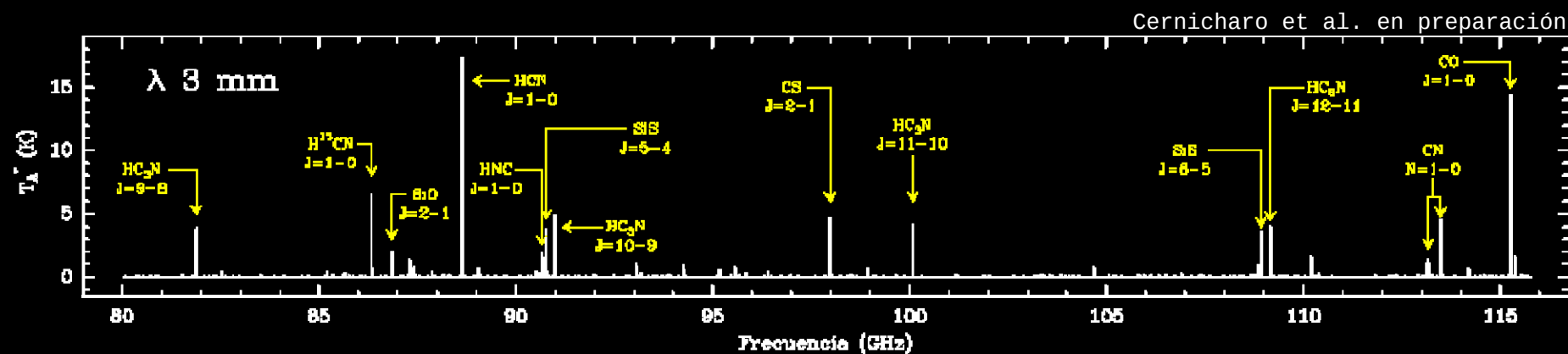
**Input for chemistry : C<sub>2</sub>H<sub>2</sub> & HCN in the inner shell**  
**PhD of J.Pablo Fonfría**

# LINE SURVEYS





Cernicharo et al. 2000

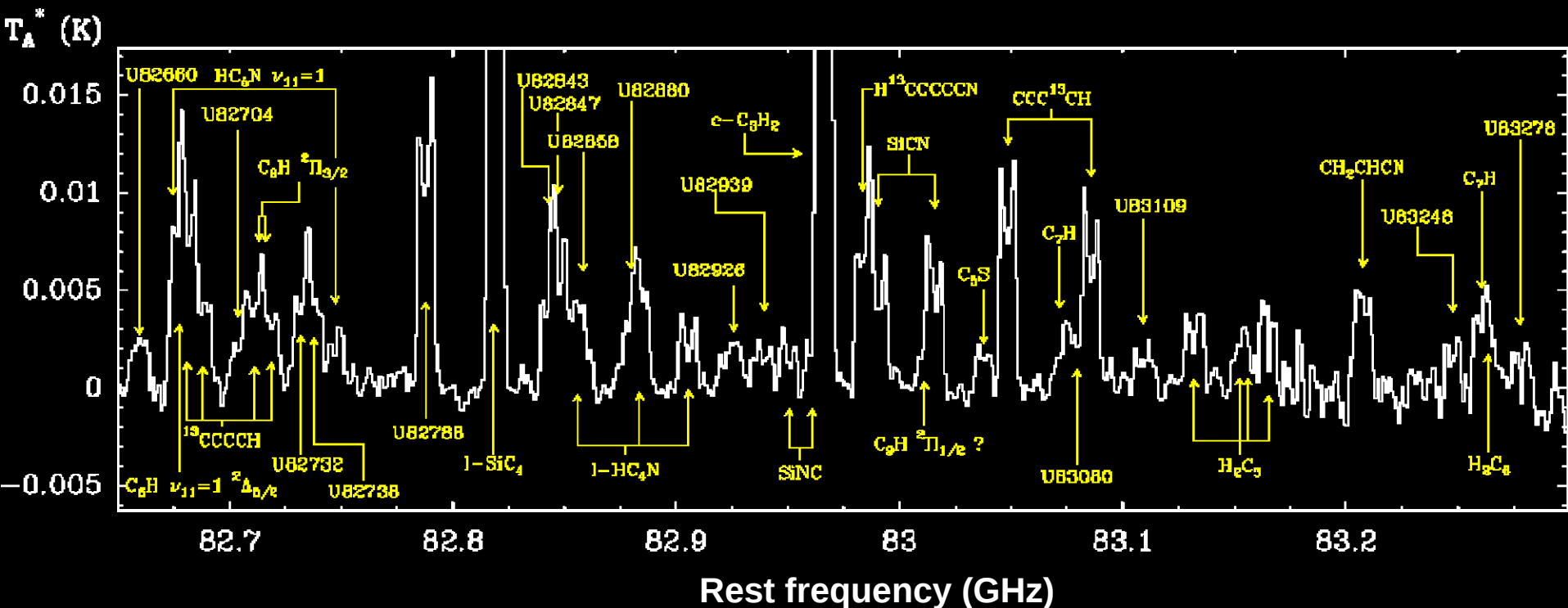


Cernicharo et al. en preparación

## METHODS: Radio astronomical observations

Line survey at  $\lambda = 3$  mm carried out with the IRAM 30-m radio telescope:

- 80.05-115.75 GHz
- 1339 emission lines ( $\sim 37$  lines/GHz)
- 886 assigned to rotational transitions of 60 molecules (+ different isotopologues and vibrationally excited states)
- 453 unassigned lines (only 31 with  $T_A^* > 10$  mK)
- high sensitivity:  $\text{rms}(T_A^*) < 1$  mK for most frequencies



Lines arise from a region where temperature and density vary very fast



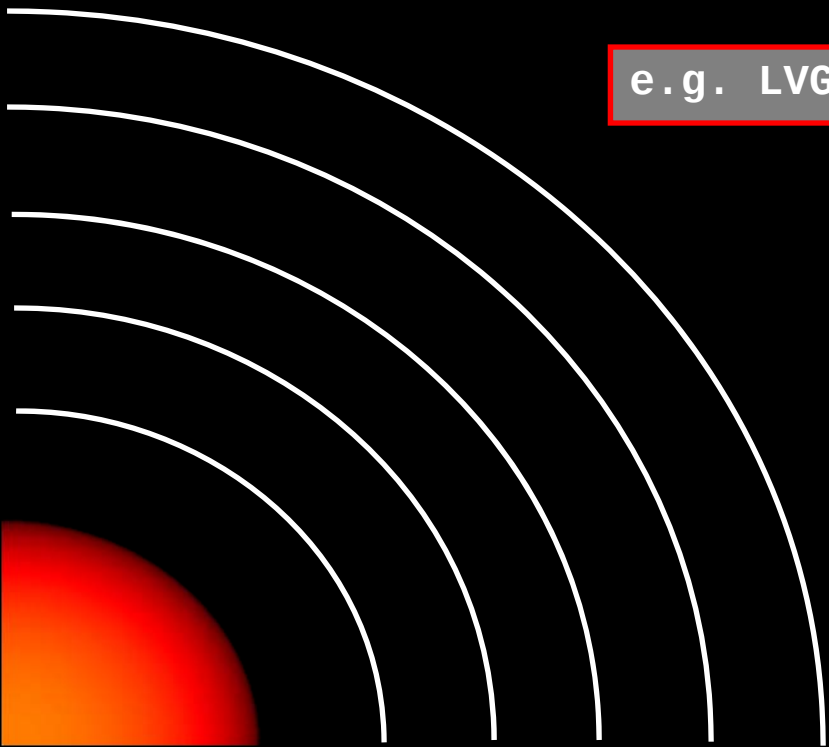
The medium is not homogeneous

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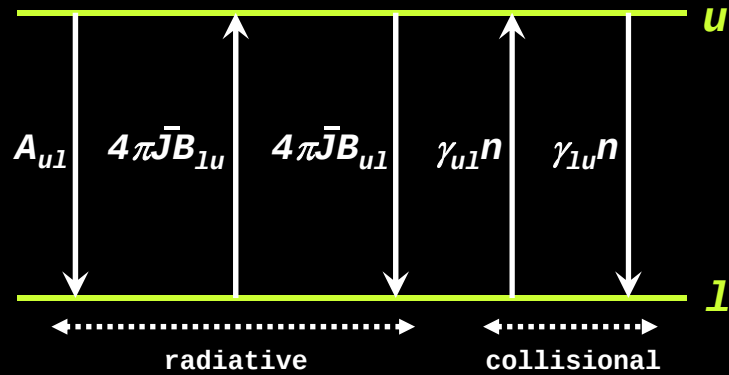
It is necessary to use more sophisticated methods



e.g. LVG multishell or non-local codes

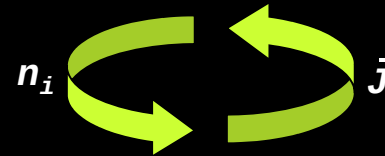


# METHODS: Non LTE radiative transfer (LVG and non Local Codes)



## Statistical Equilibrium

$$\frac{dn_i}{dt} = \sum_{j \neq i} n_j (4\pi\bar{J}B_{ji} + \gamma_{ji}n) + \sum_{j>i} n_j A_{ji} - n_i \sum_{j \neq i} (4\pi\bar{J}B_{ij} + \gamma_{ij}n) - n_i \sum_{j<i} A_{ij}$$



$$\bar{J} = (1 - \beta)S_v + \beta I_v^{bg}$$

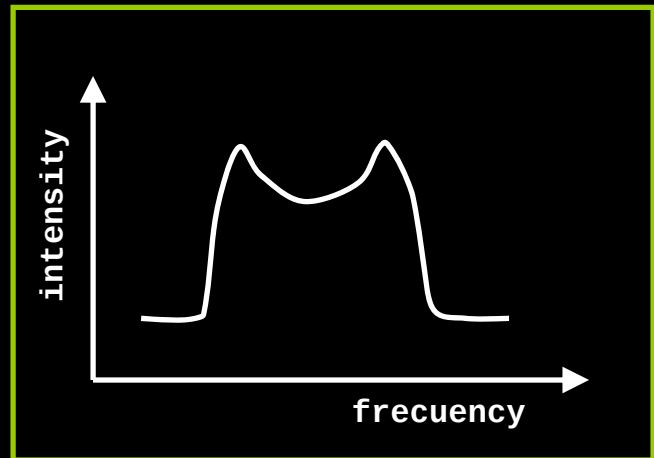
LVG codes



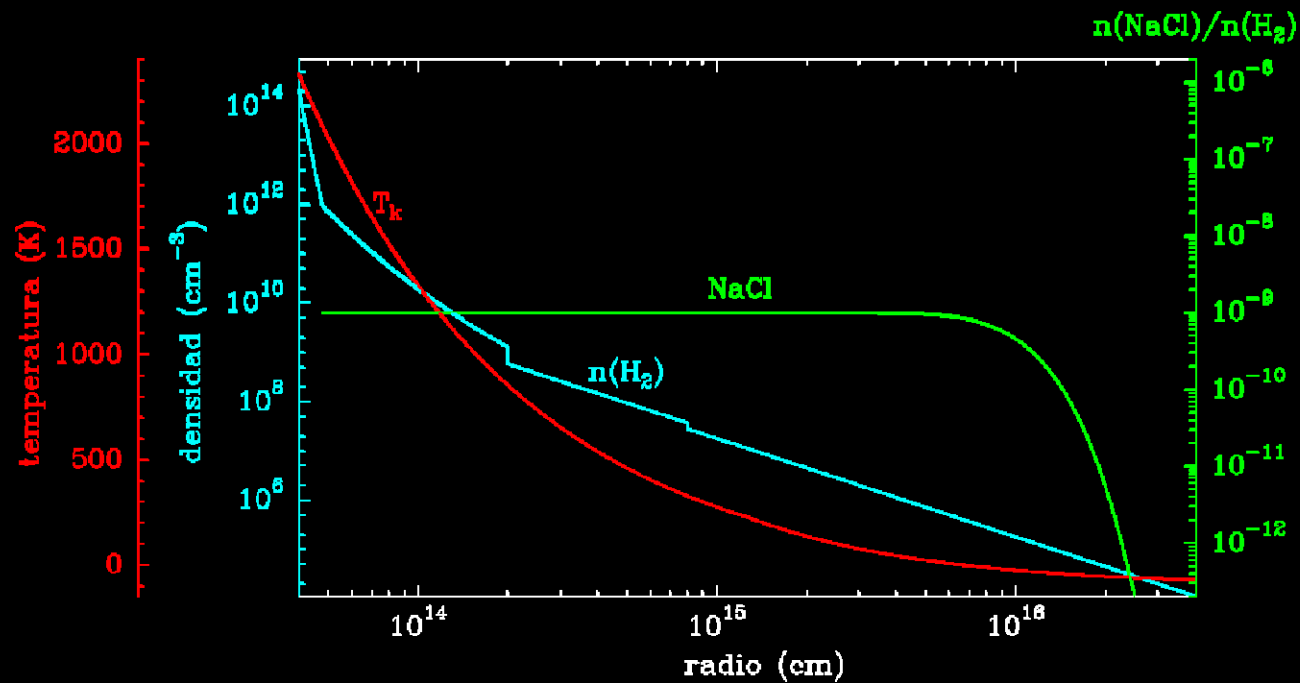
$n_i$

population

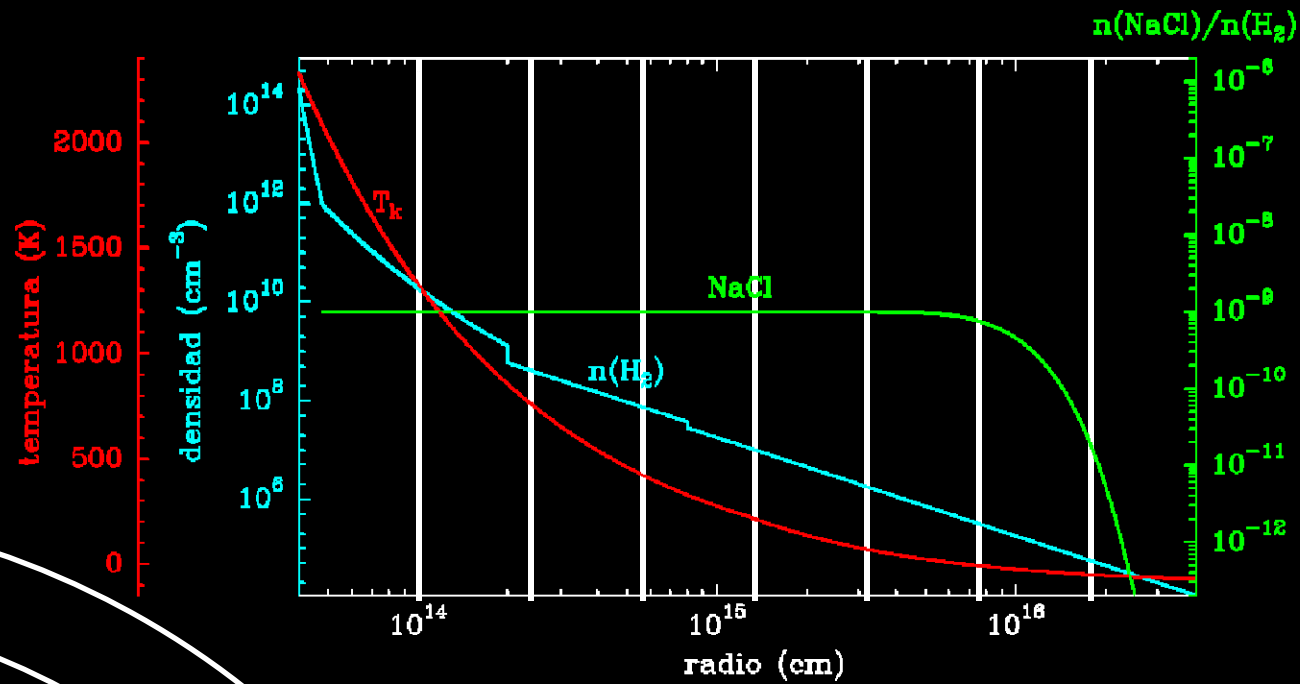
Ray tracing



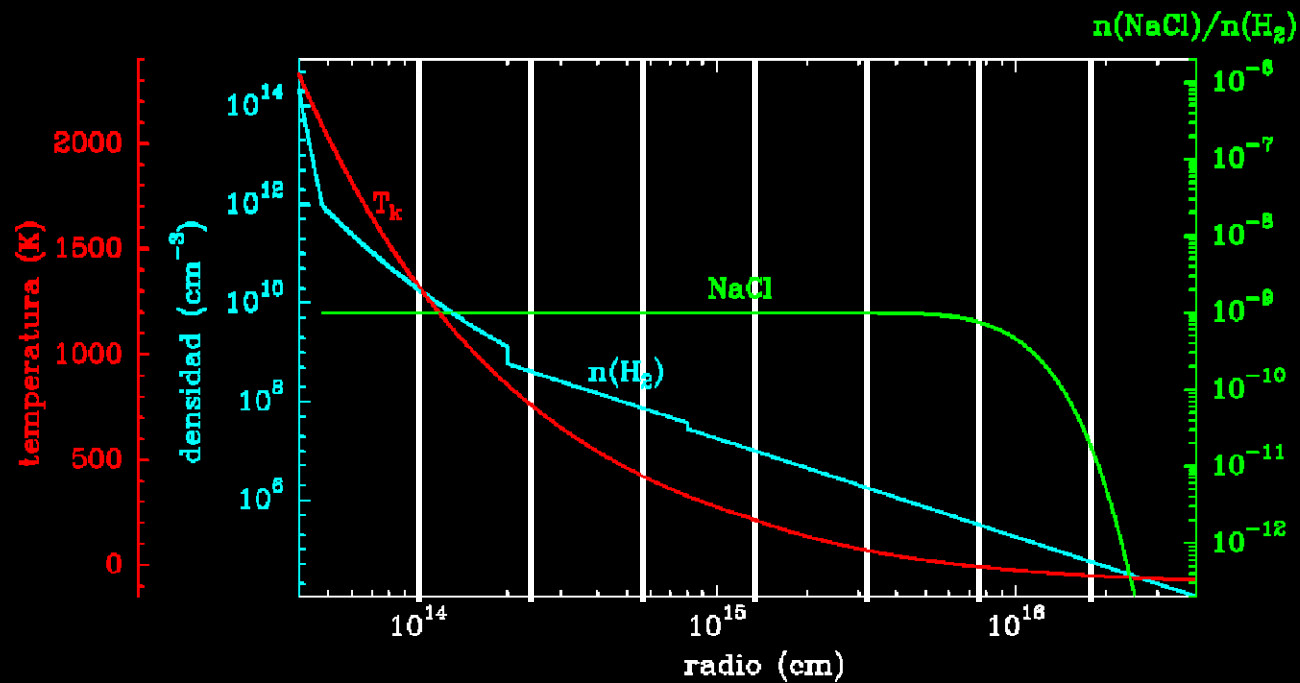
## LVG multishell models



## LVG multishell models



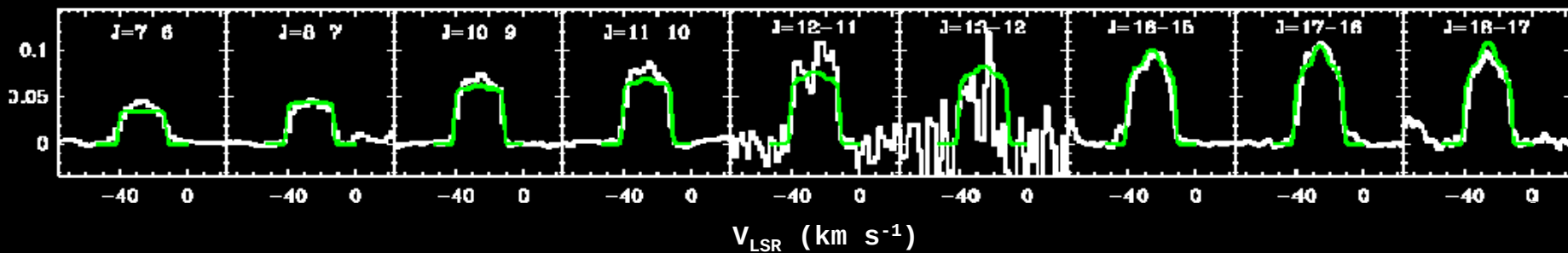
## LVG multishell models



Results

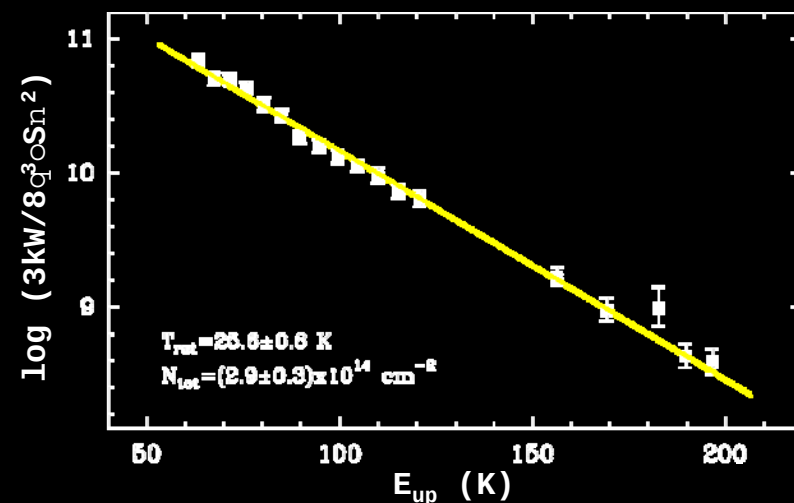
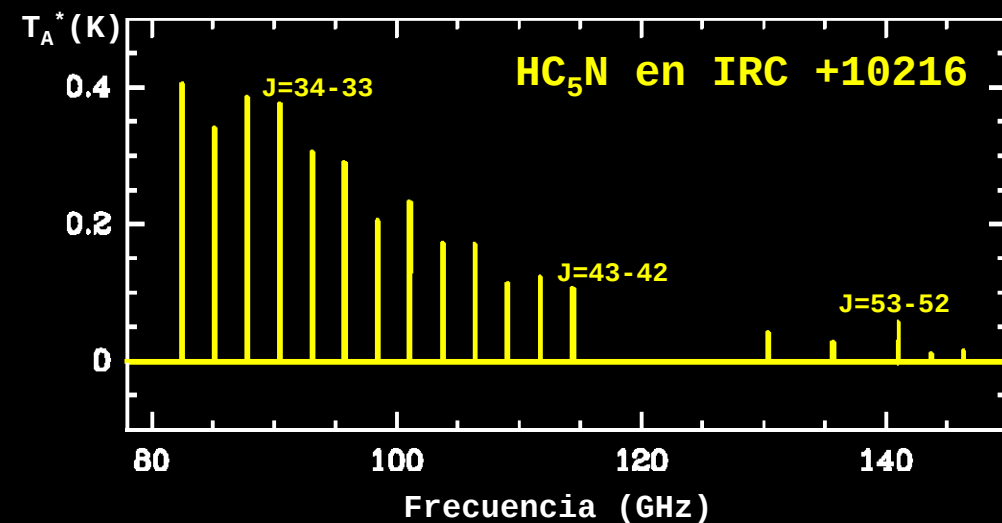
 $T_A^* (\text{K})$ 

NaCl



## Rotational temperature Diagram

$$\log\left(\frac{3kW}{8\pi^3\nu S\mu^2}\right) = \log\left(\frac{N}{Z_{\text{rot}}}\right) - \frac{\log e}{kT_{\text{rot}}} E_{\text{up}}$$



Integrated intensity of the lines

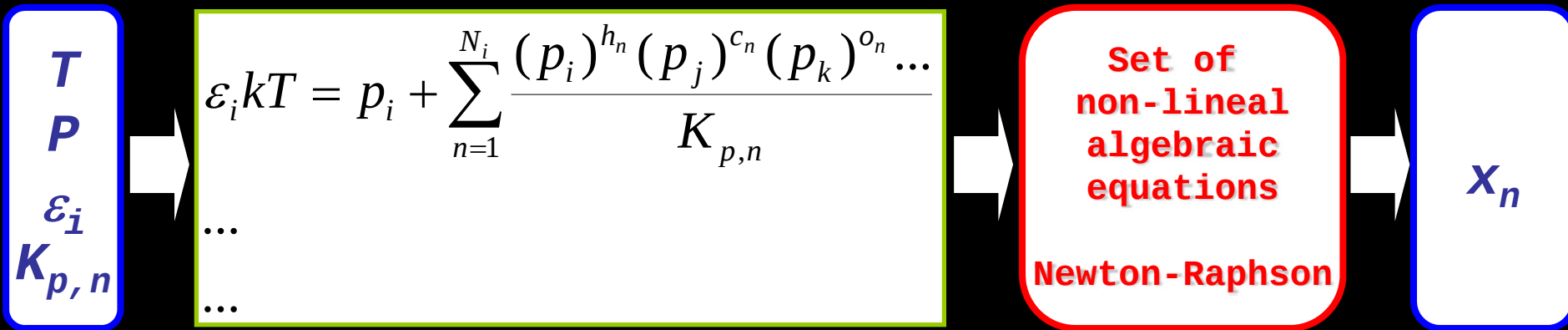
Rotational temperature ( $T_{\text{rot}}$ ) and column density ( $N$ )

### Very crude approach for most species!

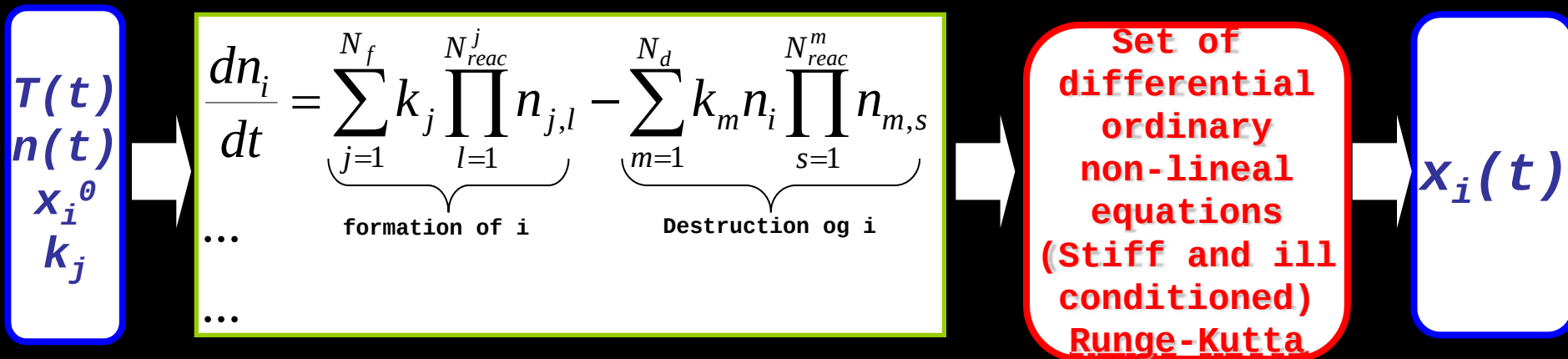
- 1) rotational levels in LTE at some  $T_{\text{rot}}$
- 2) the medium has to be homogeneous
- 3) optically thin lines
- 4) no infrared pumping
- 5)  $T_{\text{rot}} \gg T_{\text{CMB}}$

|                   |                   |         |                                  |         |  |  |                                  |         |                   |         |                    |         |                 |          |
|-------------------|-------------------|---------|----------------------------------|---------|--|--|----------------------------------|---------|-------------------|---------|--------------------|---------|-----------------|----------|
| 10 <sup>-3</sup>  | CO                | 1(-3)   |                                  |         |  |  |                                  |         |                   |         |                    |         |                 |          |
| 10 <sup>-4</sup>  |                   |         |                                  |         |  |  |                                  |         |                   |         |                    |         |                 |          |
|                   |                   |         | C <sub>2</sub> H <sub>2</sub>    | 8(-5)   |  |  | HCN                              | 2(-5)   |                   |         |                    |         |                 |          |
| 10 <sup>-5</sup>  |                   |         |                                  |         |  |  |                                  |         |                   |         |                    |         |                 |          |
|                   |                   |         | CH <sub>4</sub>                  | 3.5(-6) |  |  |                                  |         |                   |         |                    |         |                 |          |
|                   |                   |         | C <sub>2</sub> H                 | 3(-6)   |  |  | NH <sub>3</sub>                  | 2(-6)   |                   |         |                    |         |                 |          |
|                   |                   |         | C <sub>4</sub> H                 | 2.5(-6) |  |  | CN                               | 1.7(-6) |                   |         |                    |         |                 |          |
|                   |                   |         | C <sub>2</sub>                   | 1(-6)   |  |  | HC <sub>3</sub> N                | 1.4(-6) |                   |         | SiC <sub>2</sub>   | 1.2(-6) |                 |          |
| 10 <sup>-6</sup>  |                   |         | C <sub>3</sub>                   | 1(-6)   |  |  |                                  |         |                   |         | SiS                | 1(-6)   |                 |          |
|                   |                   |         |                                  |         |  |  | C <sub>3</sub> N                 | 4(-7)   | CS                | 5(-7)   |                    |         |                 |          |
|                   |                   |         |                                  |         |  |  |                                  |         |                   |         | SiH <sub>4</sub>   | 2.2(-7) |                 |          |
|                   |                   |         |                                  |         |  |  | HC <sub>5</sub> N                | 2(-7)   |                   |         | SiO                | 1.2(-7) |                 |          |
| 10 <sup>-7</sup>  | H <sub>2</sub> O  | 1(-7)   | C <sub>5</sub>                   | 1(-7)   |  |  | HNC                              | 1(-7)   |                   |         |                    |         |                 |          |
|                   |                   |         | 1-C <sub>3</sub> H               | 5(-8)   |  |  |                                  |         |                   |         |                    |         |                 |          |
|                   | OH                | 4(-8)   | C <sub>6</sub> H                 | 4(-8)   |  |  |                                  |         |                   |         | SiC                | 4(-8)   |                 |          |
|                   |                   |         | C <sub>5</sub> H                 | 3(-8)   |  |  | CH <sub>3</sub> CN               | 3(-8)   | C <sub>2</sub> S  | 3(-8)   |                    |         | AlCl            | 3.5(-8)  |
|                   |                   |         | c-C <sub>3</sub> H <sub>2</sub>  | 3(-8)   |  |  |                                  |         |                   |         |                    |         |                 |          |
|                   |                   |         | CH <sub>3</sub> C <sub>2</sub> H | 3(-8)   |  |  |                                  |         |                   |         |                    |         |                 |          |
|                   |                   |         | c-C <sub>3</sub> H               | 2(-8)   |  |  | HC <sub>7</sub> N                | 2(-8)   |                   |         |                    | HCP     | 2.5(-8)         |          |
|                   |                   |         | C <sub>2</sub> H <sub>4</sub>    | 2(-8)   |  |  |                                  |         |                   |         |                    |         | NaCN            |          |
|                   | H <sub>2</sub> CO | 1.3(-8) | H <sub>2</sub> C <sub>4</sub>    | 1.4(-8) |  |  |                                  |         | C <sub>3</sub> S  | 1.2(-8) |                    |         |                 |          |
| 10 <sup>-8</sup>  |                   |         |                                  |         |  |  |                                  |         |                   |         |                    |         |                 |          |
|                   |                   |         | C <sub>8</sub> H                 | 8(-9)   |  |  | HC <sub>9</sub> N                | 8(-9)   | H <sub>2</sub> CS | 7(-9)   | SiN                | 8(-9)   | PH <sub>3</sub> | 8(-9)    |
|                   |                   |         |                                  |         |  |  | CH <sub>2</sub> CN               | 7(-9)   |                   |         |                    |         | MgNC            | 8(-9)    |
|                   |                   |         |                                  |         |  |  | HC <sub>2</sub> N                | 6(-9)   |                   |         |                    |         | AlF             | 7.5(-9)  |
|                   |                   |         |                                  |         |  |  | C <sub>5</sub> N                 | 4(-9)   |                   |         |                    |         |                 |          |
|                   |                   |         | C <sub>7</sub> H                 | 3(-9)   |  |  | HCCNC                            | 4(-9)   |                   |         |                    |         |                 |          |
|                   |                   |         | H <sub>2</sub> C <sub>6</sub>    | 3(-9)   |  |  | C <sub>2</sub> H <sub>3</sub> CN | 4(-9)   | H <sub>2</sub> S  | 4(-9)   | c-SiC <sub>3</sub> | 4(-9)   |                 |          |
|                   |                   |         | C <sub>6</sub> H <sup>-</sup>    | 3(-9)   |  |  | C <sub>5</sub> N <sup>-</sup>    | 2.3(-9) |                   |         | SiC <sub>4</sub>   | 3(-9)   |                 |          |
|                   | C <sub>3</sub> O  | 2(-9)   | C <sub>8</sub> H <sup>-</sup>    | 1.5(-9) |  |  | HC <sub>4</sub> N                | 2(-9)   |                   |         | SiCN               | 2(-9)   |                 |          |
|                   |                   |         | H <sub>2</sub> C <sub>3</sub>    | 1.5(-9) |  |  | C <sub>3</sub> N <sup>-</sup>    | 1.1(-9) | C <sub>5</sub> S  | 1.2(-9) | SiNC               | 1.1(-9) | PN              | 1(-9)    |
| 10 <sup>-9</sup>  |                   |         |                                  |         |  |  |                                  |         |                   |         |                    |         |                 |          |
|                   | HCO <sup>+</sup>  | 7(-10)  |                                  |         |  |  | HNCCC                            | 5(-10)  |                   |         |                    |         | NaCl            | 1(-9)    |
|                   |                   |         | C <sub>4</sub> H <sup>-</sup>    | 3(-10)  |  |  |                                  |         |                   |         | C <sub>2</sub> P   | 1(-9)   | AlNC            | 1(-9)    |
| 10 <sup>-10</sup> |                   |         |                                  |         |  |  |                                  |         |                   |         |                    |         |                 |          |
|                   |                   |         |                                  |         |  |  |                                  |         |                   |         |                    |         | MgCN            | 5(-10)   |
|                   |                   |         |                                  |         |  |  |                                  |         |                   |         |                    |         | KCl             | 2.5(-10) |

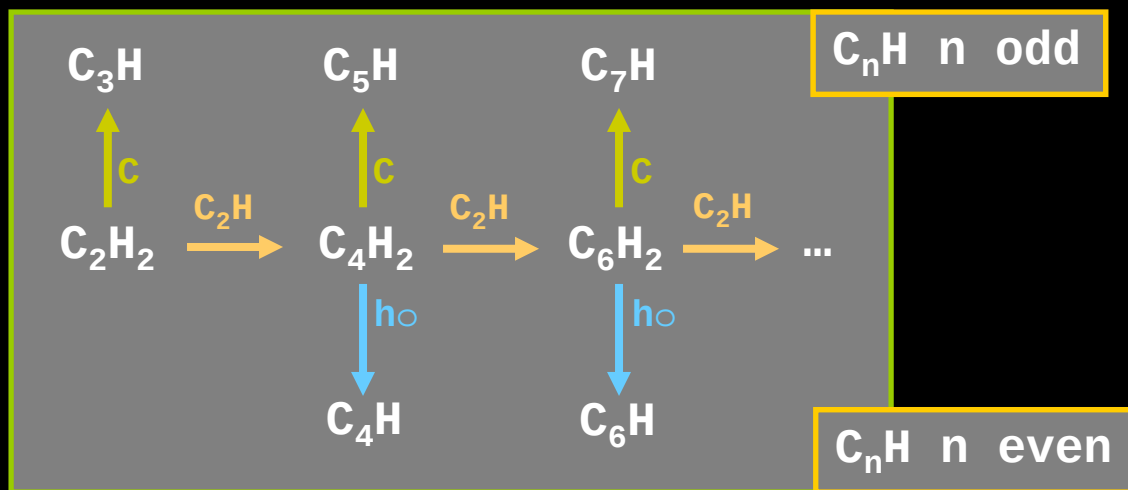
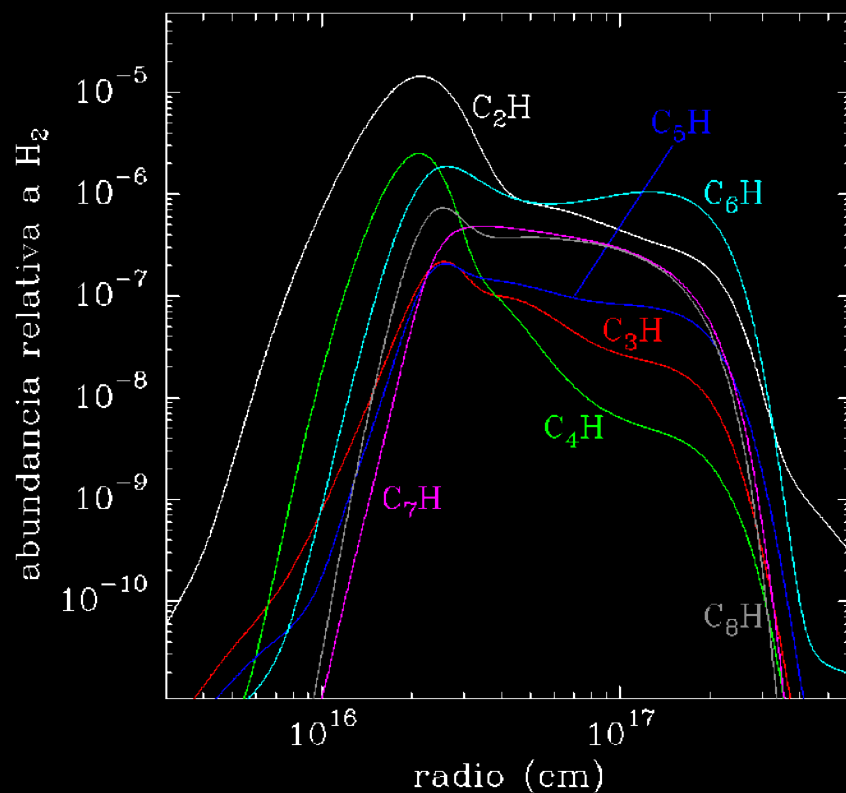
## Thermodynamical Chemical Equilibrium : **Dissociation energies and partition functions are needed**



## Chemical Kinetics: **Reaction rates are required**

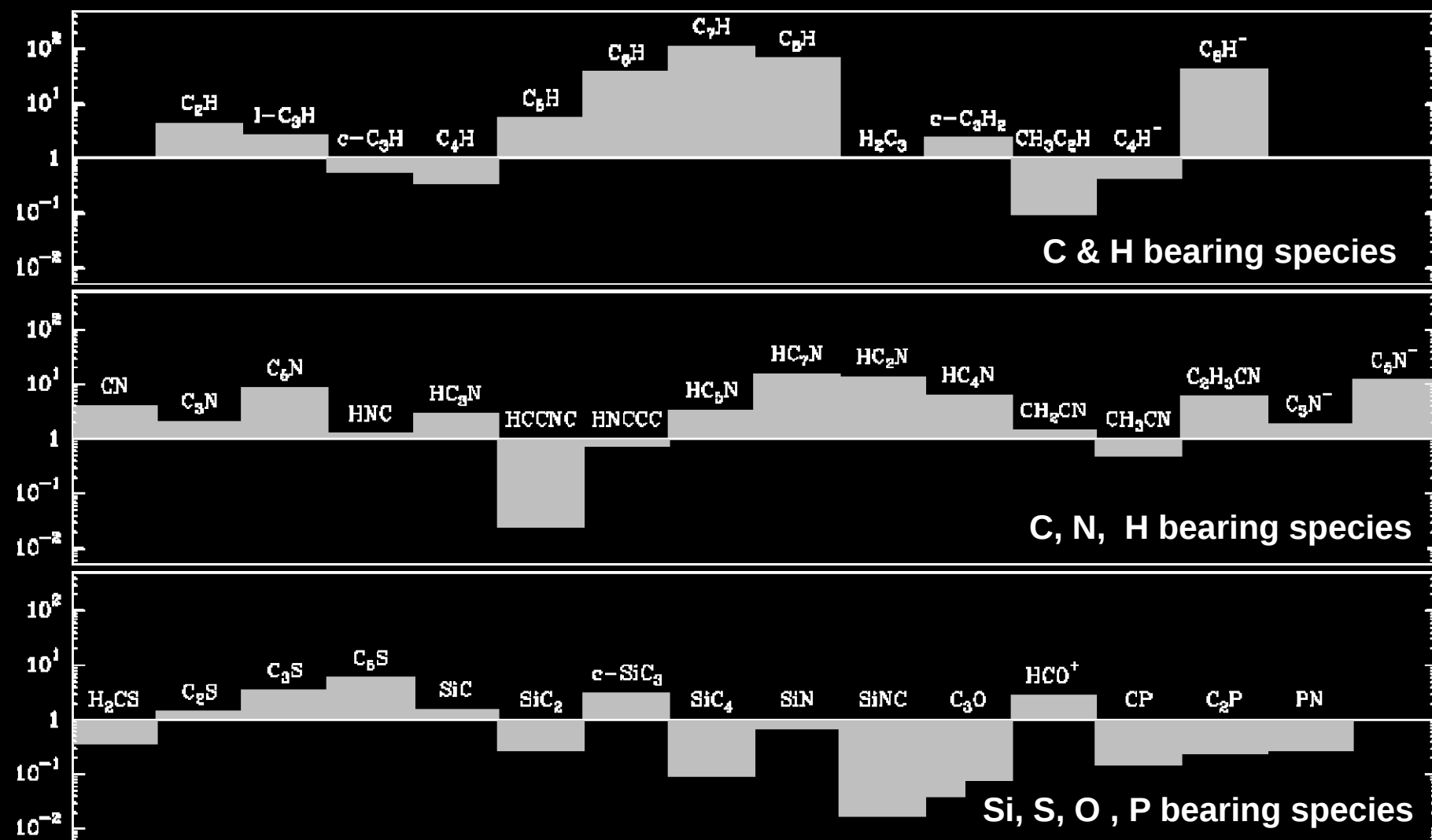


# RESULTS: THE OUTER ENVELOPE --- Chemical kinetic models



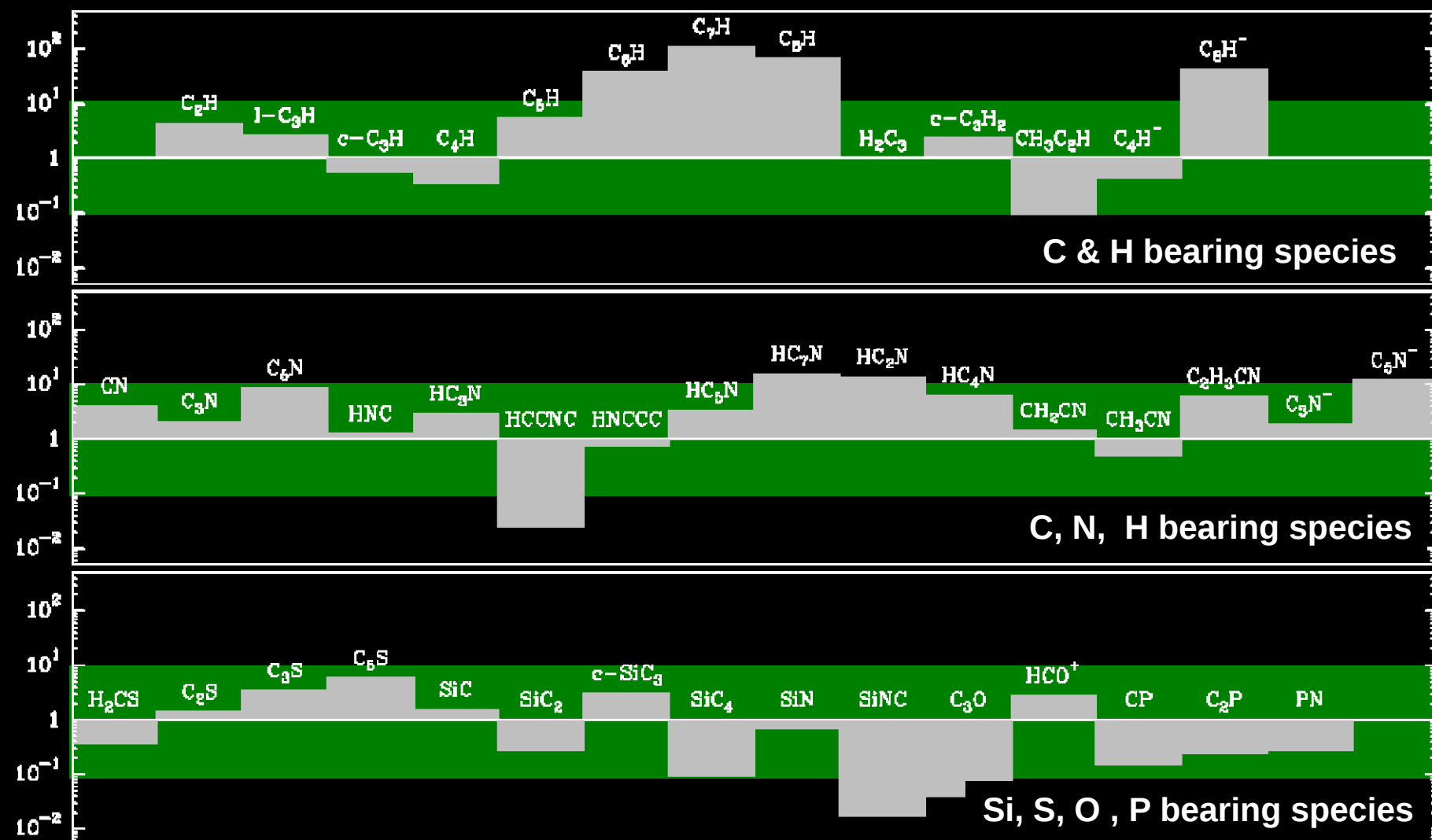
# RESULTS: THE OUTER ENVELOPE --- Chemical kinetic models

$N_{\text{calc}}/N_{\text{obs}}$



# RESULTS: THE OUTER ENVELOPE --- Chemical kinetic models

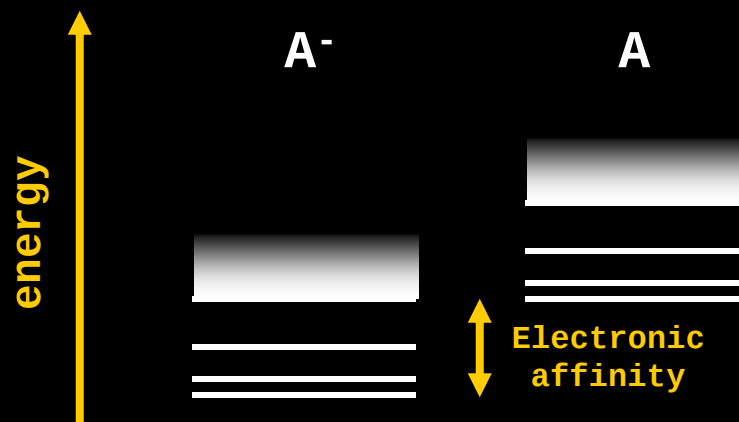
$N_{\text{calc}}/N_{\text{obs}}$



i First detection of molecular anions in IRC+10216  
 $C_6H^-$ ,  $C_4H^-$ ,  $C_8H^-$ ,  $C_3N^-$ ,  $C_5N^-$

### Thermodynamics

The formation of the anion is favoured



### Kinetics

The formation of the anion is NOT favoured



Except for species with a high electronic affinity and a large size  
e.g.  $C_4H$ ,  $C_6H$ ,  $C_8H$ ,  $C_3N$ ,  $C_5N$ , ...

*Nature Vol. 289 19 February 1981*

## **Can negative molecular ions be detected in dense interstellar clouds?**

**Eric Herbst**

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**The recent laboratory measurement<sup>1-3</sup> of rapid radiative electron attachment processes**



where A is a molecular species has renewed speculation on whether negative molecular ions can be synthesized efficiently in dense interstellar clouds. We argue here that for certain interstellar species A, the abundance ratio  $[A^-]/[A]$  may be as high as 0.01–0.1 in commonly assumed physical conditions. If this abundance ratio were correct, negative molecular ions might be detectable in dense interstellar clouds if their microwave spectral frequencies had been determined in the laboratory. It will be shown, however, that this is currently an unlikely prospect.

## History:

2006  $\text{C}_6\text{H}^-$  in IRC +10216 and TMC-1 (McCarthy et al.)

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**2006**  $\text{C}_6\text{H}^-$  in IRC +10216 & TMC-1 (McCarthy et al.)

**2007**  $\text{C}_4\text{H}^-$  in IRC +10216 (Cernicharo et al.)

**2007**  $\text{C}_8\text{H}^-$  in IRC +10216 & TMC-1 (Remijan et al.; Brünken et al.)

**2008**  $\text{C}_3\text{N}^-$  in IRC +10216 (Thaddeus et al.)

**2008**  $\text{C}_5\text{N}^-$  in IRC +10216 (Cernicharo et al.)

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**2008**  $\text{C}_3\text{N}^-$  in IRC +10216 (Thaddeus et al.)

**2008**  $\text{C}_5\text{N}^-$  in IRC +10216 (Cernicharo et al.)

**2010**  $\text{CN}^-$  in IRC+10216 (Agúndez et al.)

## Additional detections:

$\text{C}_6\text{H}^-$  in L1527 (Sakai et al. 2007)

$\text{C}_4\text{H}^-$  in L1527 (Agúndez et al. 2008)

$\text{C}_6\text{H}^-$  in L1544 y L1521F (Gupta et al. 2009)

## History:

2006  $C_6H^-$  in IRC +10216 & TMC-1 (McCarthy et al.)

2007  $C_4H^-$  in IRC +10216 (Cernicharo et al.)

2007  $C_8H^-$  in IRC +10216 & TMC-1 (Remijan et al.)

2008  $C_3N^-$  in IRC +10216 (Thornley et al.)

2008  $C_5N^-$  in IRC +10216 (Thornley et al.)

2

IRC +10216  
Is the only source where all anions have been observed  
 $CN^-$ ,  $C_4H^-$ ,  $C_6H^-$ ,  $C_8H^-$ ,  $C_3N^-$ ,  $C_5N^-$

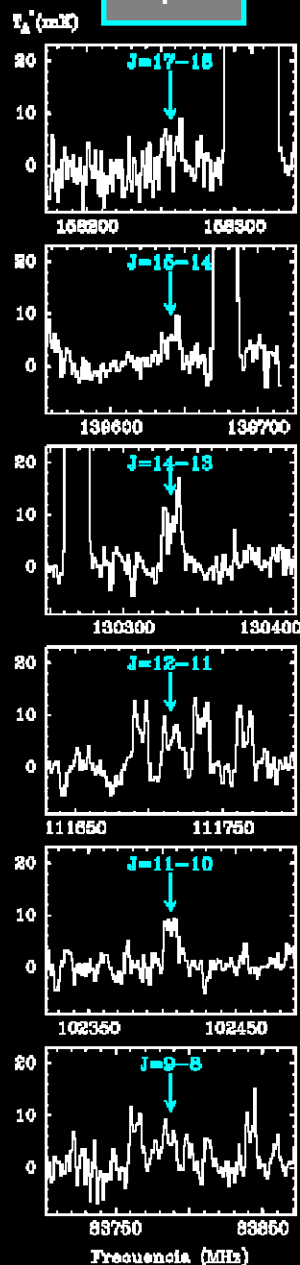
2007  $C_4H^-$  in L1527 (Sakai et al. 2007)

$C_4H^-$  in L1527 (Agúndez et al. 2008)

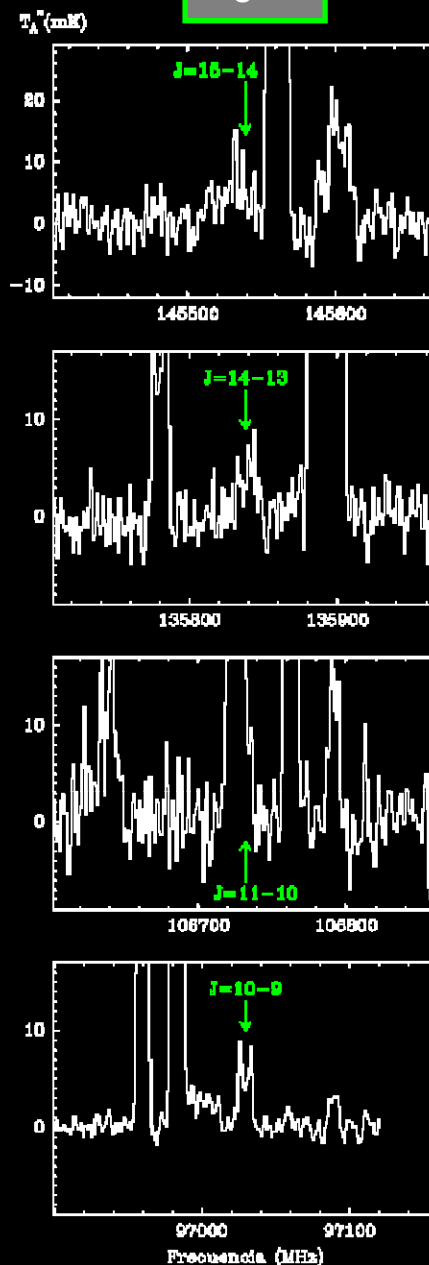
$C_6H^-$  in L1544 & L1521F (Gupta et al. 2009)

# RESULTS: SPECIFIC RESULTS : ANIONS

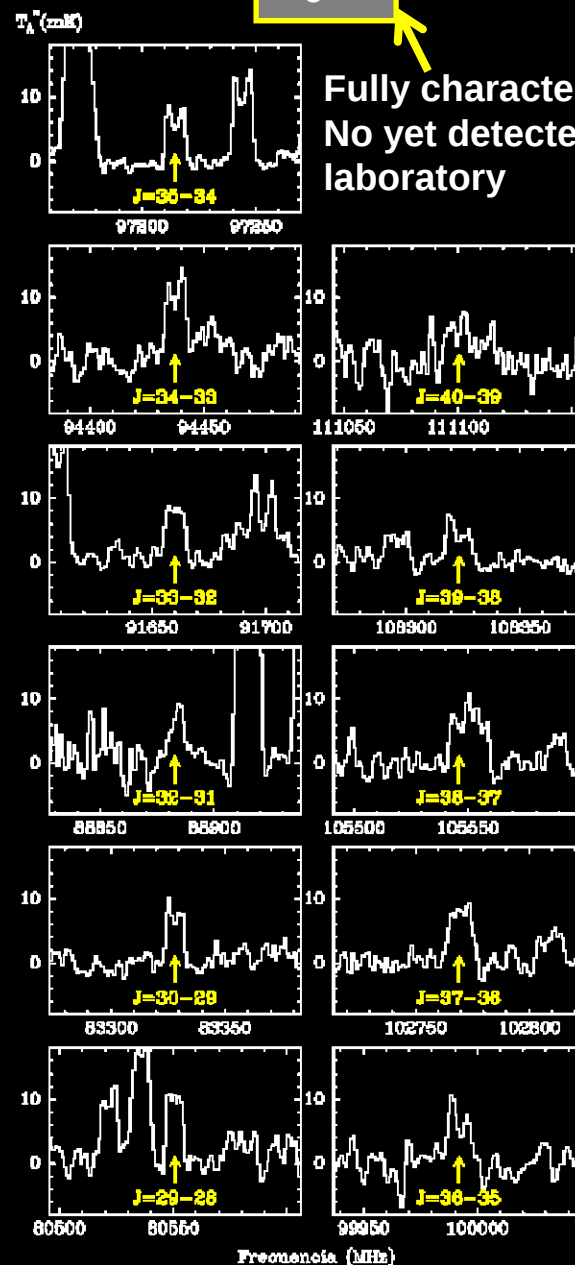
$C_4H^-$



$C_3N^-$



$C_5N^-$



Fully characterized in space  
No yet detected in the laboratory

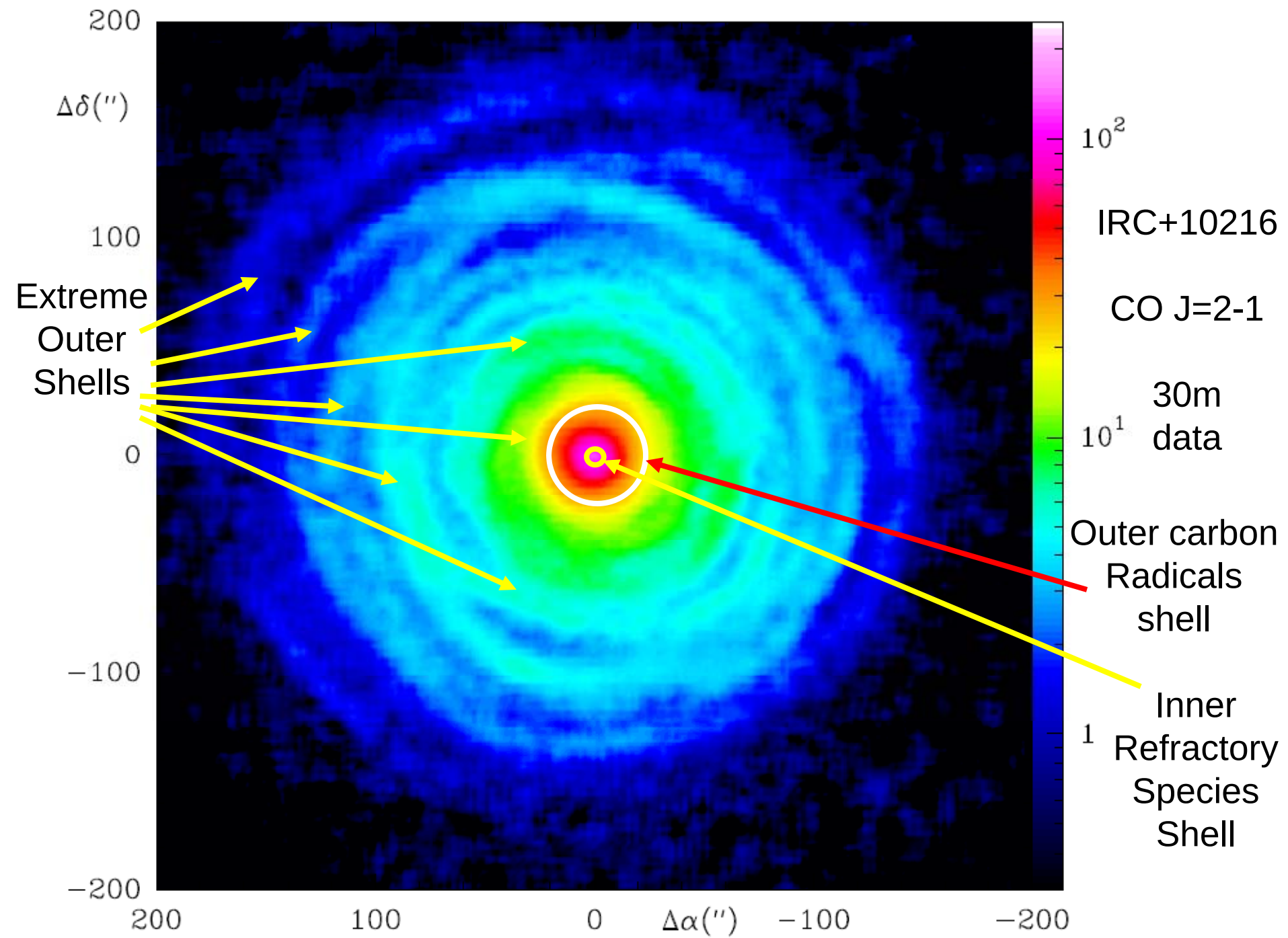
| reaction                                                                         | $k_{ra}$ (300 K)<br>IRC +10216 | $k_{ra}$ (300 K)<br>theory <sup>(a)</sup> |
|----------------------------------------------------------------------------------|--------------------------------|-------------------------------------------|
| $\text{C}_4\text{H} + \text{e}^- \rightleftharpoons \text{C}_4\text{H}^- + h\nu$ | $2-7 \times 10^{-11}$          | $1.1 \times 10^{-8}$                      |
| $\text{C}_6\text{H} + \text{e}^- \rightleftharpoons \text{C}_6\text{H}^- + h\nu$ | $3.0 \times 10^{-8}$           | $6.2 \times 10^{-8}$                      |
| $\text{C}_8\text{H} + \text{e}^- \rightleftharpoons \text{C}_8\text{H}^- + h\nu$ | $1.5 \times 10^{-7}$           | $6.2 \times 10^{-8}$                      |
| $\text{C}_3\text{N} + \text{e}^- \rightleftharpoons \text{C}_3\text{N}^- + h\nu$ | $2-5 \times 10^{-9}$           | $2.0 \times 10^{-10}$                     |
| $\text{C}_5\text{N} + \text{e}^- \rightleftharpoons \text{C}_5\text{N}^- + h\nu$ | $5.0 \times 10^{-7}$           | -                                         |

$k_{ra}$  with units of  $\text{cm}^3 \text{ s}^{-1}$

(a) Herbst & Osamura 2008; Petrie & Herbst 1997.

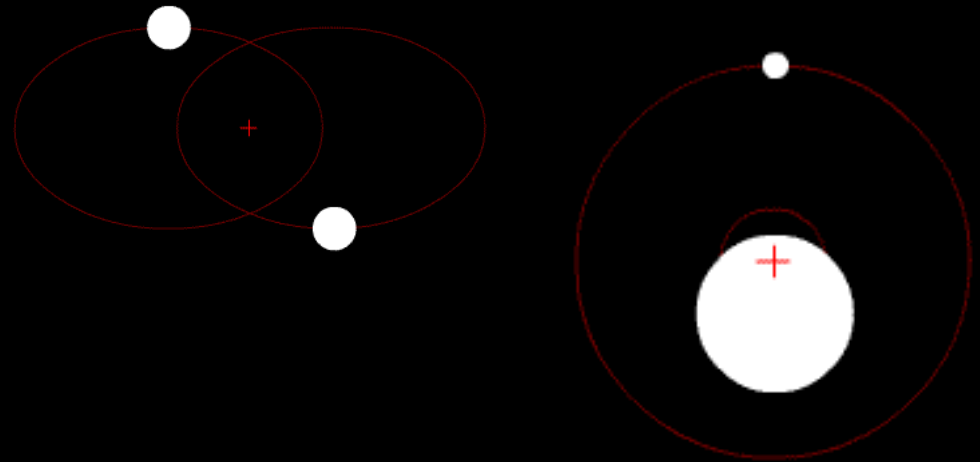
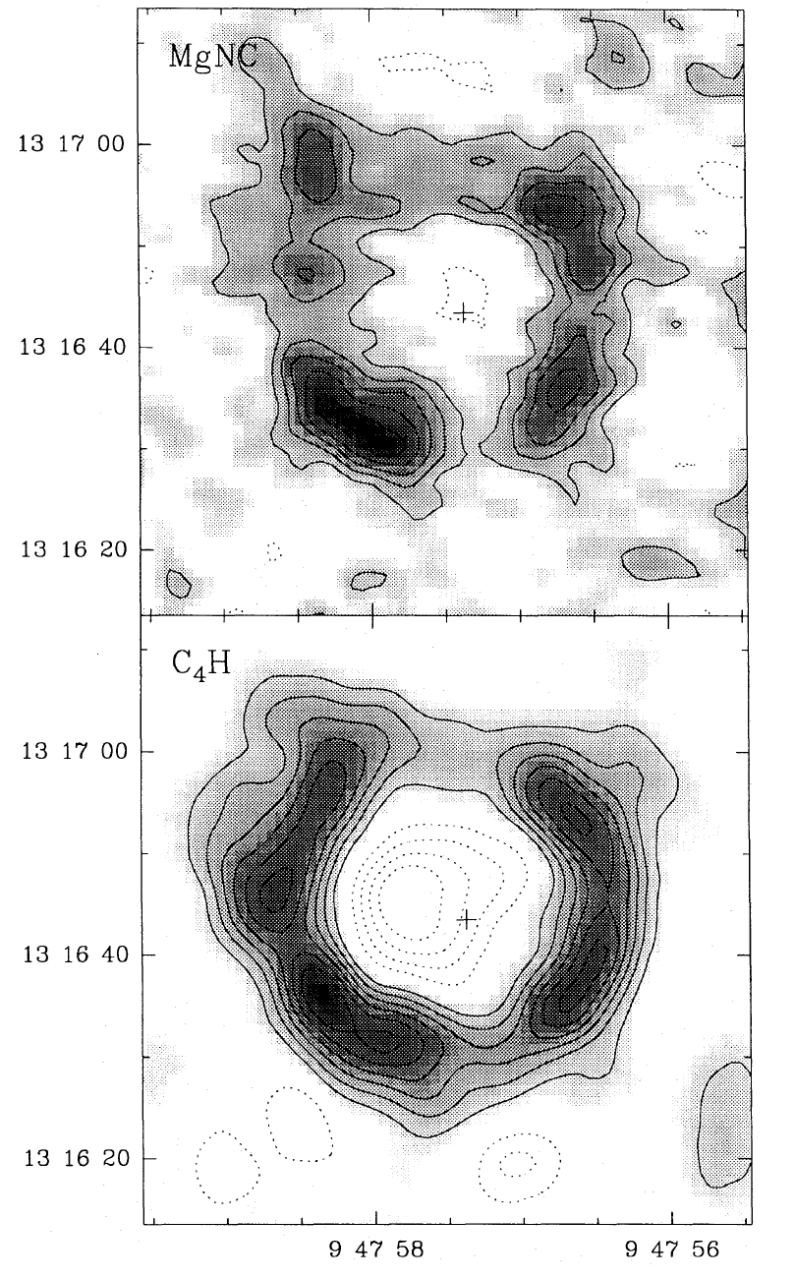
# THE FINAL PRODUCT OF A DETAILED STUDY

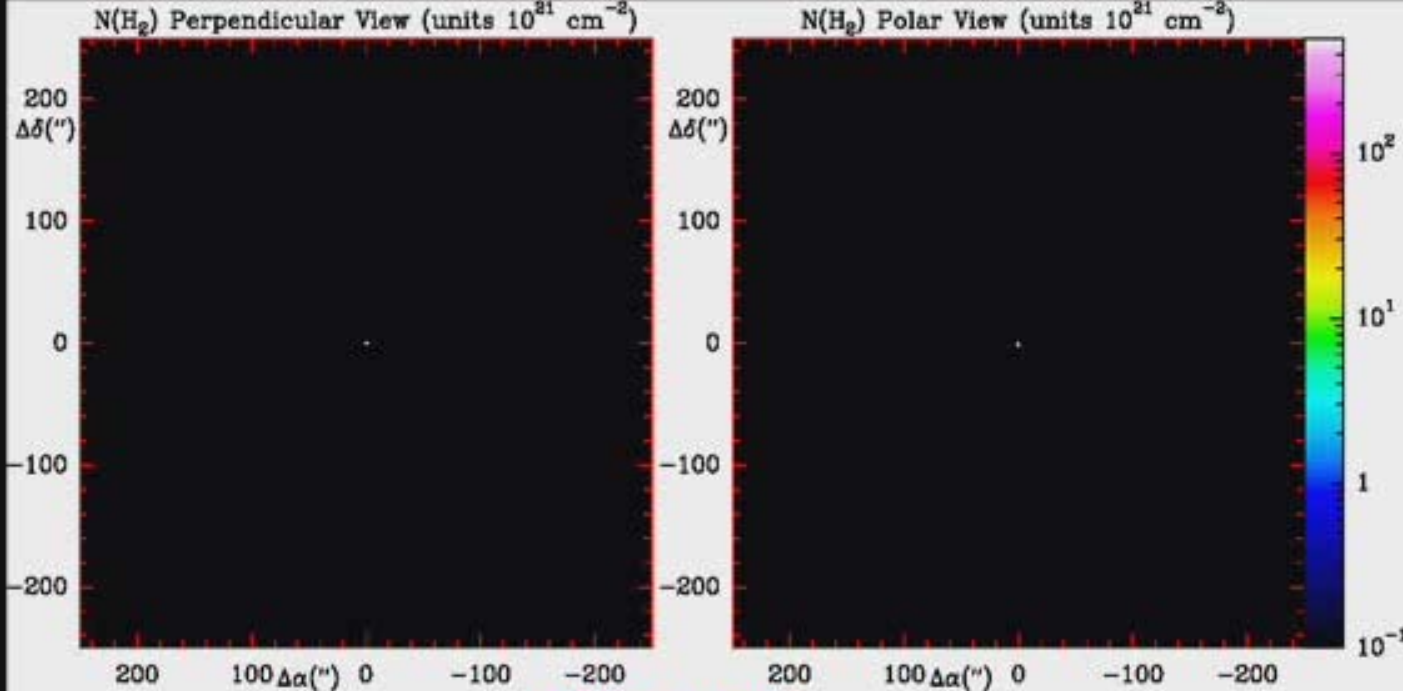
- New Molecules
- Abundances for all species
- Isotopic abundances (nuclear evolution)
- Clear differentiation of the different layers of the CSE
- Chemistry of exotic species (anions)
- A fine study of the missing reactions of the actual chemical networks



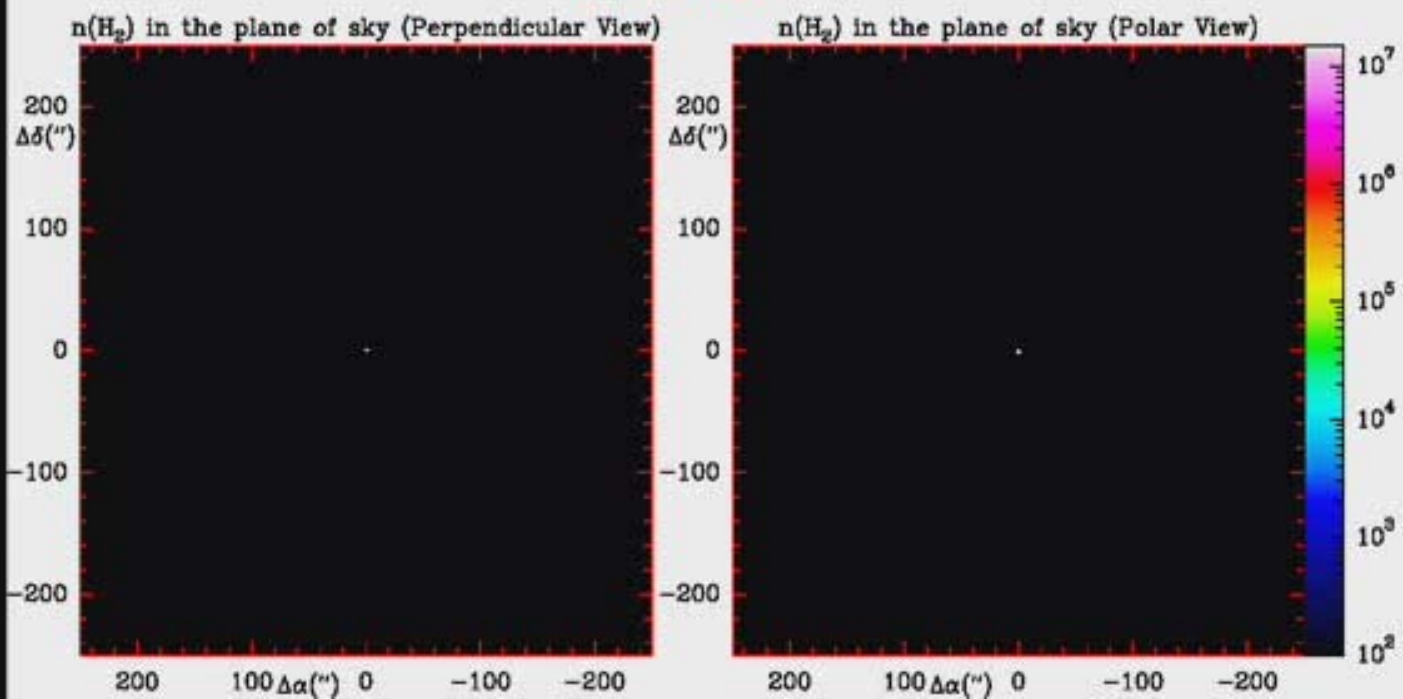
**What structure could we expect  
for the circumstellar envelope  
of an AGB binary star ?**

**Is the AGB loosing mass  
uniformely with time or it is  
having episodic events of  
high mass loss ?**





**TIME=40 yr**

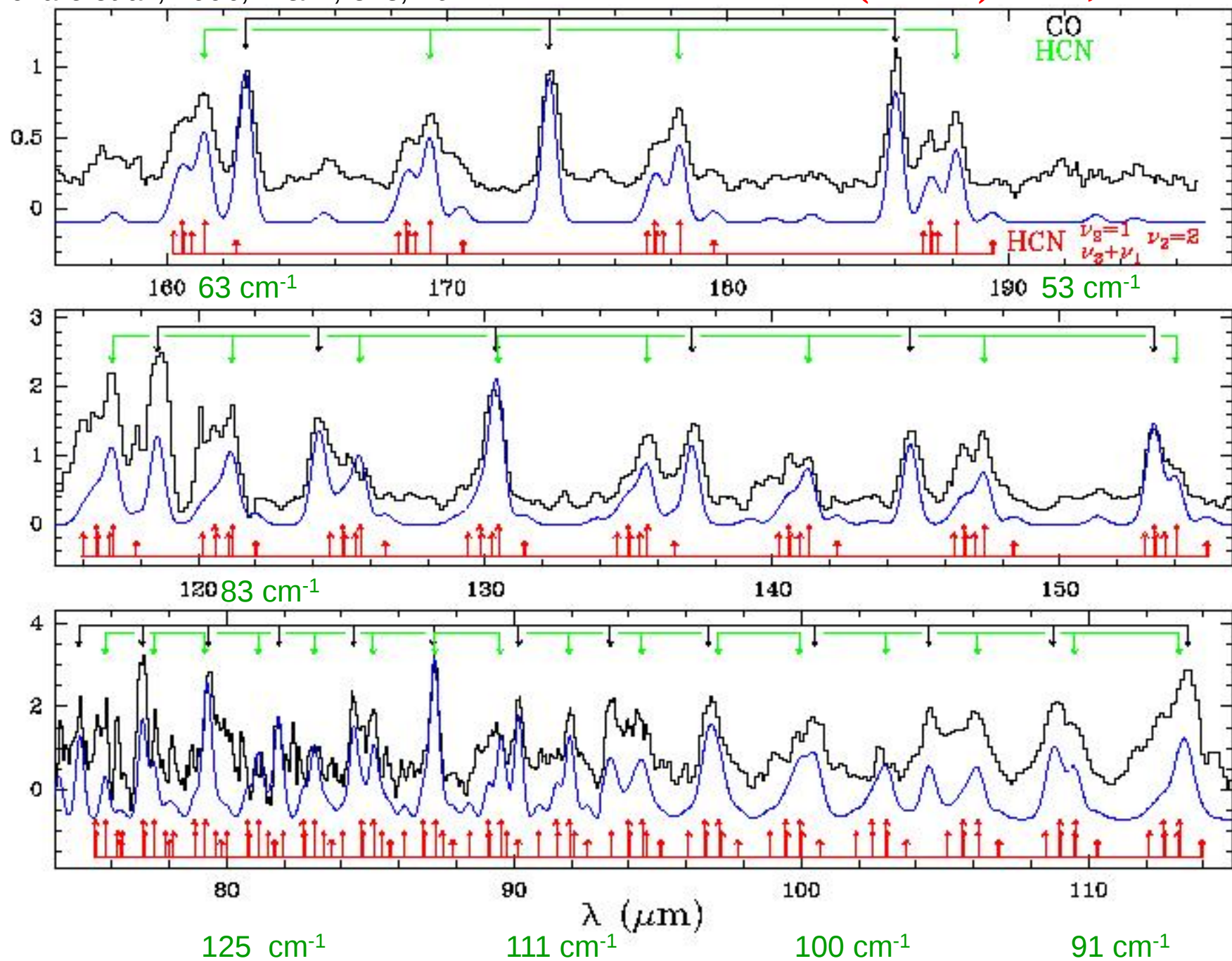


Mass loss increased when the stars are at the closest distance.

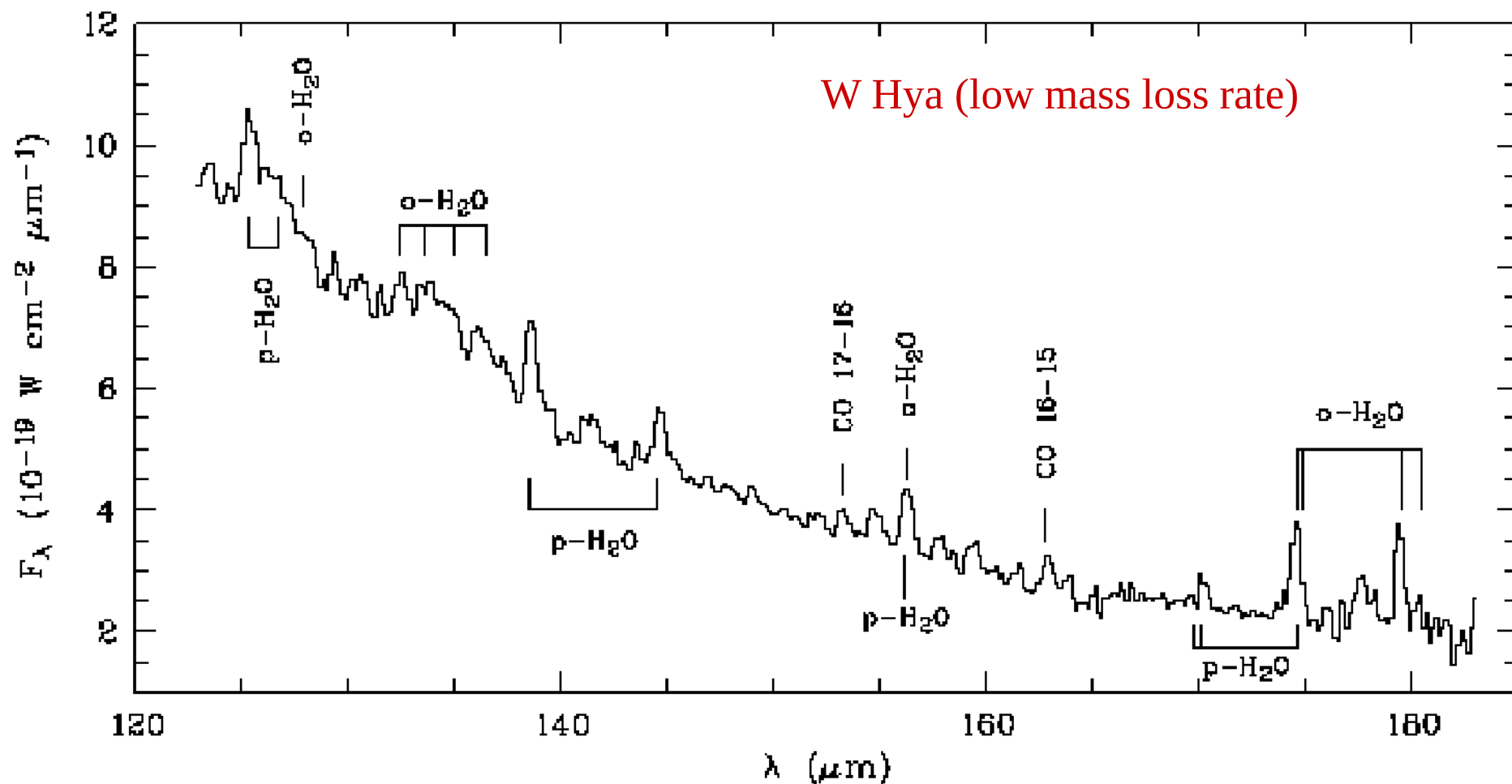
Ad hoc mass loss rate increased by a factor 10

**WATER IN C-rich and O-rich AGBs**

**ALMA can do it !**

$F_{\lambda} (10^{-18} \text{ Watt cm}^{-2} \mu\text{m}^{-1})$ 


# Water in O-rich AGB stars



Barlow et al., 1996; Neufeld et al., 1996

## VY CMa

All features are real and belong  
(95%) to H<sub>2</sub>O (5% from CO)

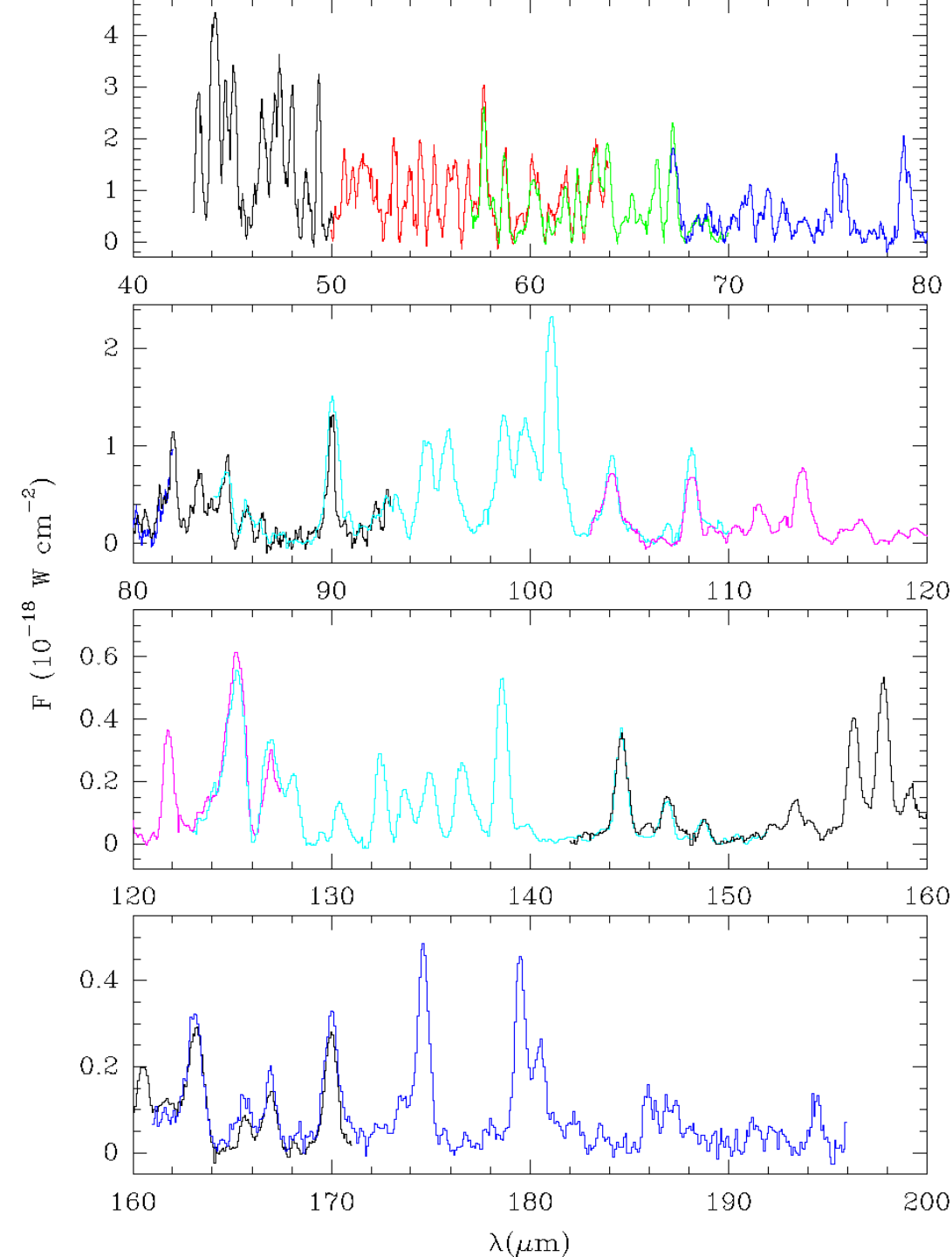
All pure rotational lines of water  
vapour with  $\lambda > 43$   $\mu\text{m}$  detected

Some lines from the  $v_2=1$   
bending level also detected

Modelling requires collisional  
rates for  $T_k=20\text{-}2000$  K including  
ro-vibrational collisions

ISO DATA 1998

VyCMa is a high mass loss rate  
object





**If C-rich :** CO, C<sub>2</sub>H<sub>2</sub>, HCN, SiO, SiS and a lot of species including H<sub>2</sub>O.

When AGBs move towards the PPN phase => warm photochemistry using the UV photons from the central star. Significant amounts of water, and other O-bearing species are produced

**If O-rich:** CO, H<sub>2</sub>O, OH, SiO, SiS, HCN, ...

**GROUND BASED OBSERVATIONS OF WATER**

**The problem of the observation of water vapor :**

**Our atmosphere is full of water vapor.**

**Water vapor was detected in the ISM and CSM in 1969 from observations at a 22 GHz (Cheung et al.).**

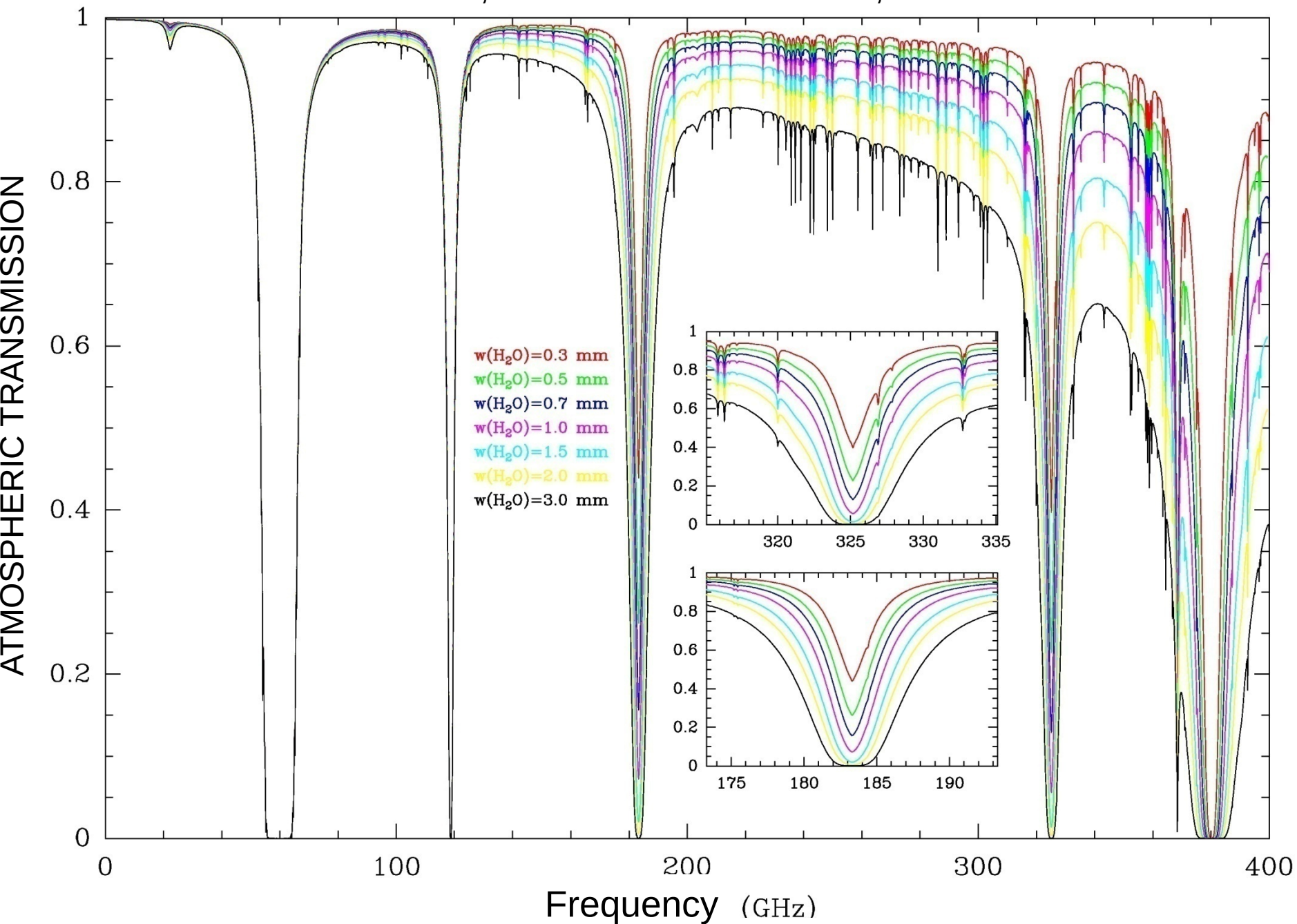
**Our atmosphere is transparent at this frequency as the transition involves two levels around 700 K and the line strength is rather low.**

**Water detected only in small regions of the ISM and CSM**

**To observe H<sub>2</sub>O we have to go to space !!!**

**But this is not completely true !!!!!**

ALTITUDE = 4600 m; PRESSURE = 560 mbars; TEMPERATURE = -5 C



## Letter to the Editor

# Detection of 183 GHz water vapor maser emission from interstellar and circumstellar sources

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<sup>2</sup> IRAM, Av. Divina Pastora N7 NC, E-18012 Granada, Spain

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Received January 12, accepted February 13, 1990

**SUMMARY:** We have observed, with the IRAM 30-m telescope, the emission of the 183 GHz  $3_{13-2_{20}}$  rotational transition of water from a selected sample of giant molecular clouds, low mass star forming regions and evolved stars. The zenith atmospheric transmission was 25-30% during our observations. The water emission shows strong and narrow masing lines (flux  $\approx 1000$ -15000 Jy) toward the hot molecular clouds with strong 22 GHz water masers. Broad line emission, perhaps of thermal origin, is found in the sources associated with molecular outflows. Double peaked profiles, corresponding to masing lines (flux  $\approx 100$ -600 Jy), are observed toward the evolved stars. Finally, 183 GHz maser emission is also observed in the more quiescent clouds, but at velocities different from those of the associated 22 GHz masers.

**Keywords:** *Interstellar medium : molecules : clouds. Masers. Radio lines : molecular. Earth : atmosphere. Stars : Circumstellar matter.*

## 1) INTRODUCTION:

Chemical models predict water vapor to be one the most abundant molecules in the interstellar medium (ISM) and in the molecular envelopes of oxygen-rich stars. That  $H_2O$  is present in the ISM is known since the detection of maser emission from its  $6_{16-5_{23}}$  22 GHz line (Cheung et al., 1969). At this frequency the terrestrial atmosphere is fairly transparent and the 22 GHz maser emission has been detected toward a large number of star-forming regions, evolved stars, and extragalactic objects. In spite of the large amount of observations of the masing  $6_{16-5_{23}}$  line, little is known about the abundance of this molecule and its spatial distribution in interstellar molecular clouds. The problems with observing other rotational transitions of this molecule arise from the terrestrial atmosphere, which absorbs the extraterrestrial radiation precluding any observation from low altitude sites.

Two submillimeter  $H_2O$  lines have been observed toward Ori-A (KL) with the Kuiper Airborne Observatory (KAO) : the  $3_{13-2_{20}}$  line at 183 GHz (Waters et al. 1980, and Kuiper et al. 1984), and the  $4_{14-3_{21}}$  line at 380 GHz (Phillips et al. 1980). These observations indicate a high  $H_2O$

In this Letter we report the discovery of strong maser and thermal emission from the  $3_{13-2_{20}}$   $H_2O$  line toward molecular clouds, and discuss briefly its spatial distribution and origin. Further analysis of our data will be presented elsewhere (Cernicharo et al., 1990).

TABLE 1

**ZENITH ATMOSPHERIC TRANSMISSION AT PICO VELETA  
FOR WATER LINES BELOW 500 GHz**

|                    |                    | TRANSMISSION                |                             |                              |                             |
|--------------------|--------------------|-----------------------------|-----------------------------|------------------------------|-----------------------------|
| LINE               | FREQUENCY<br>(GHz) | T=-5°C<br>RH=30%<br>W=2.6mm | T=-5°C<br>RH=10%<br>W=0.9mm | T=-15°C<br>RH=10%<br>W=0.4mm | T=-20°C<br>RH=5%<br>W=0.1mm |
|                    |                    |                             |                             |                              |                             |
| $3_{13}-2_{20}$    | 183.31012          | 4%                          | 31%                         | 52%                          | 73%                         |
|                    | 183.3±0.1          | 4%                          | 34%                         | 58%                          | 80%                         |
| $10_{29}-9_{36}$   | 321.22564          | 50%                         | 78%                         | 88%                          | 94%                         |
| $5_{15}-4_{22}$    | 325.15292          | 3%                          | 27%                         | 50%                          | 72%                         |
| $4_{14}-3_{21}$    | 380.19737          |                             |                             |                              | 5%                          |
| $11_{210}-10_{37}$ | 390.13674          | 29%                         | 65%                         | 78%                          | 89%                         |
| $7_{53}-6_{60}$    | 437.34667          | 12%                         | 47%                         | 66%                          | 80%                         |
| $6_{43}-5_{50}$    | 439.15081          | 1%                          | 19%                         | 43%                          | 68%                         |
| $7_{52}-6_{61}$    | 443.01830          | 2%                          | 25%                         | 49%                          | 73%                         |
| $4_{23}-3_{30}$    | 448.00108          |                             |                             |                              | 3%                          |
| $6_{42}-5_{51}$    | 470.88895          | 8%                          | 40%                         | 61%                          | 79%                         |
| $5_{33}-4_{40}$    | 474.68913          |                             | 11%                         | 32%                          | 59%                         |
| $6_{24}-7_{17}$    | 488.49113          | 3%                          | 12%                         | 15%                          | 17%                         |

In all calculations we have assumed a pressure of 704 mbar, and a water scale height of 2.5km. T is ambient temperature and RH is relative humidity. W is the amount of water vapor above the observatory site. Frequencies are from De Lucia (1972).

## **GROUND BASED OBSERVATIONS of H<sub>2</sub>O:**

Maser lines at 22, 183, 321 and 325 GHz + some observations of  $v_2=1$

## **INFRARED SPACE OBSERVATORY (ISO) :**

Observation of a large number of pure rotational and ro-vibrational lines. Most observations with reduced spectral resolution.

## **SWAS :**

Large scale maps with high spectral resolution of interstellar clouds. Pointed observations of evolved stars. Water in C-rich objects

## **Herschel (operating since May 2009):**

Largest telescope in space (3.5 meters)

Three instruments :

SPIRE (15-52 cm<sup>-1</sup>). Spectral resolution = 2.5 GHz (FTS)

PACS (52-190 cm<sup>-1</sup>). Spectral resolution = 1500-4000 (grating)

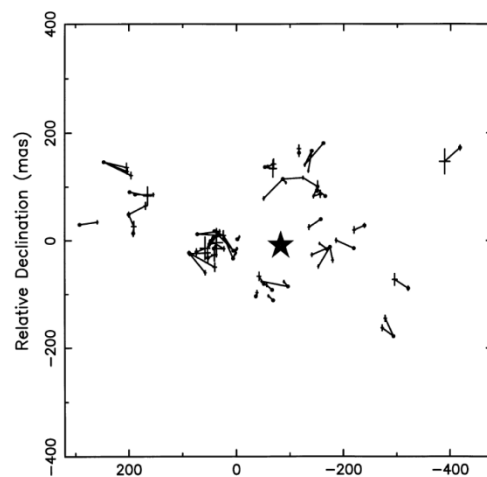
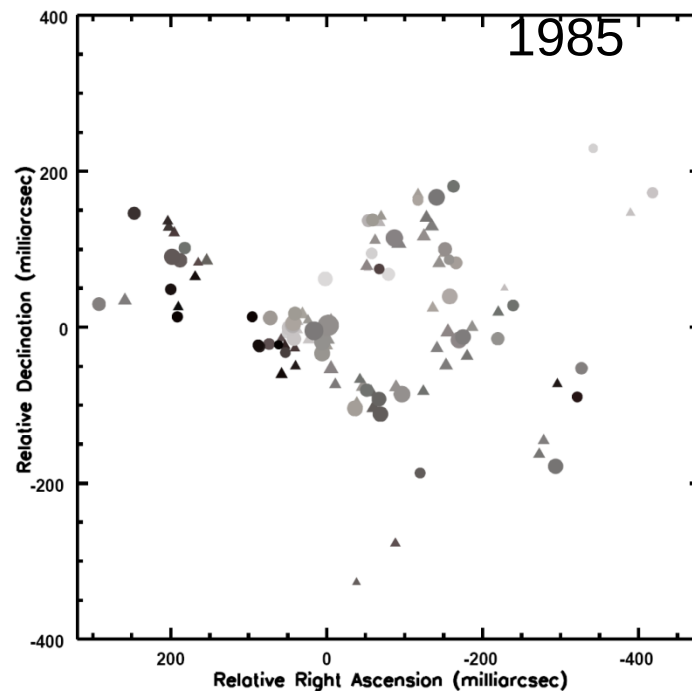
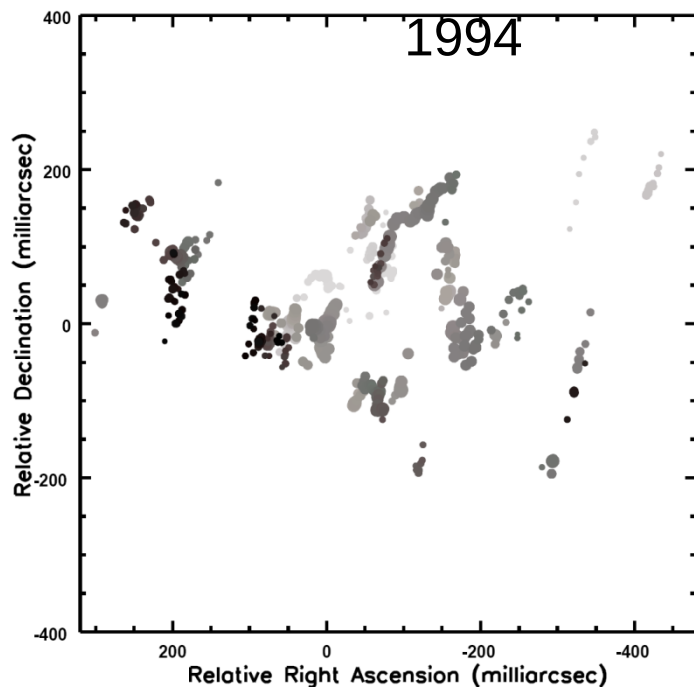
HIFI (15-65 cm<sup>-1</sup>). Spectral resolution = 10<sup>6</sup> (Heterodyne)

ALMA can observe at 325 GHz and in the next future at 183.3 GHz

## **WHAT IS KNOWN ON WATER VAPOUR IN THE ISM & CSM ?**

# 22 GHz observations of evolved stars (Vy CMa)

*A. M. S. Richards, J. A. Yates and R. J. Cohen* Mon. Not. R. Astron. Soc. **299**, 319–331 (1998)



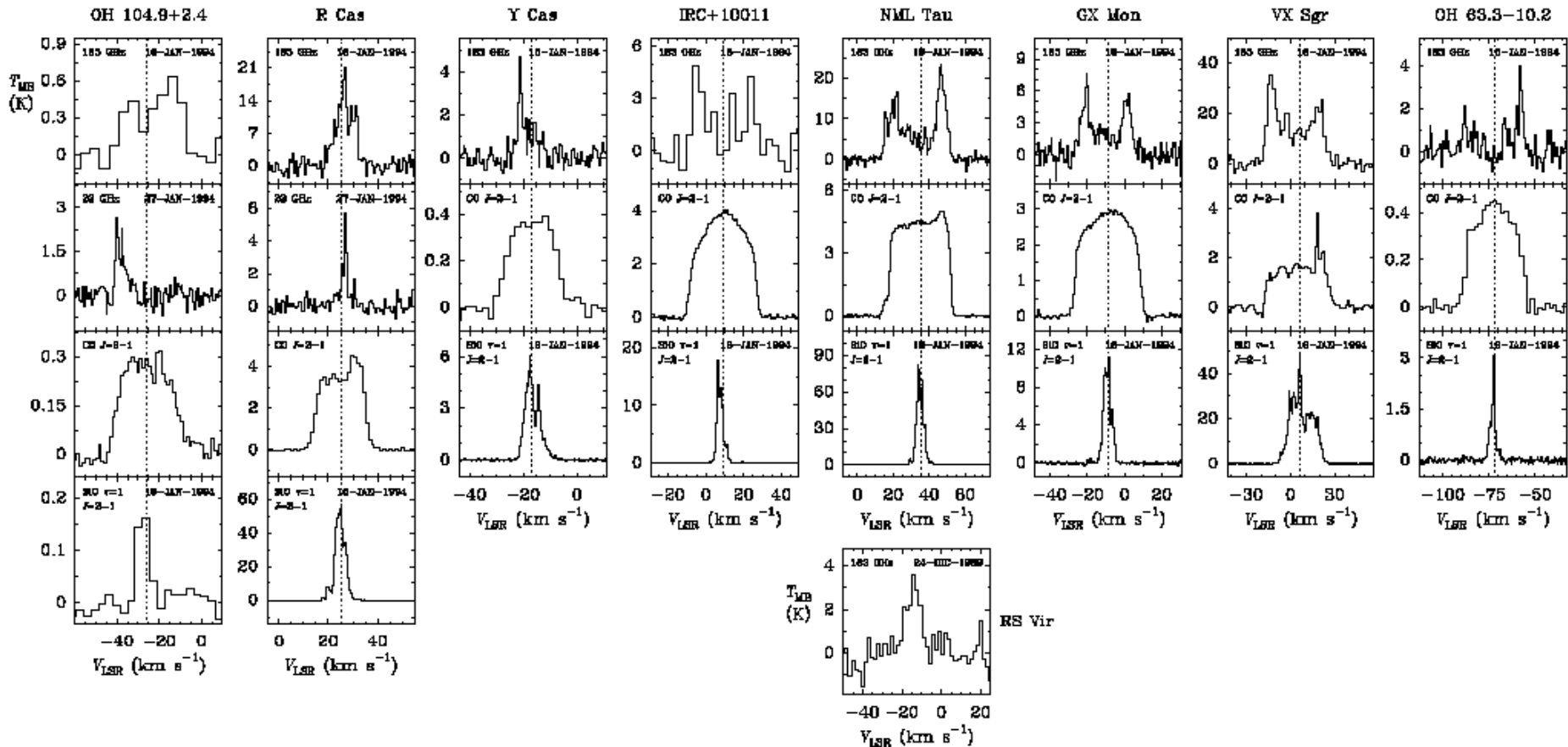
Water vapour in circumstellar envelopes

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<sup>2</sup> Observatorio Astronómico Nacional (OAN), E-28800 Alcalá de Henares, Madrid, Spain

<sup>3</sup> CSIC, IEM, Dpto. Física Molecular, Serrano 123, E-28006 Madrid, Spain



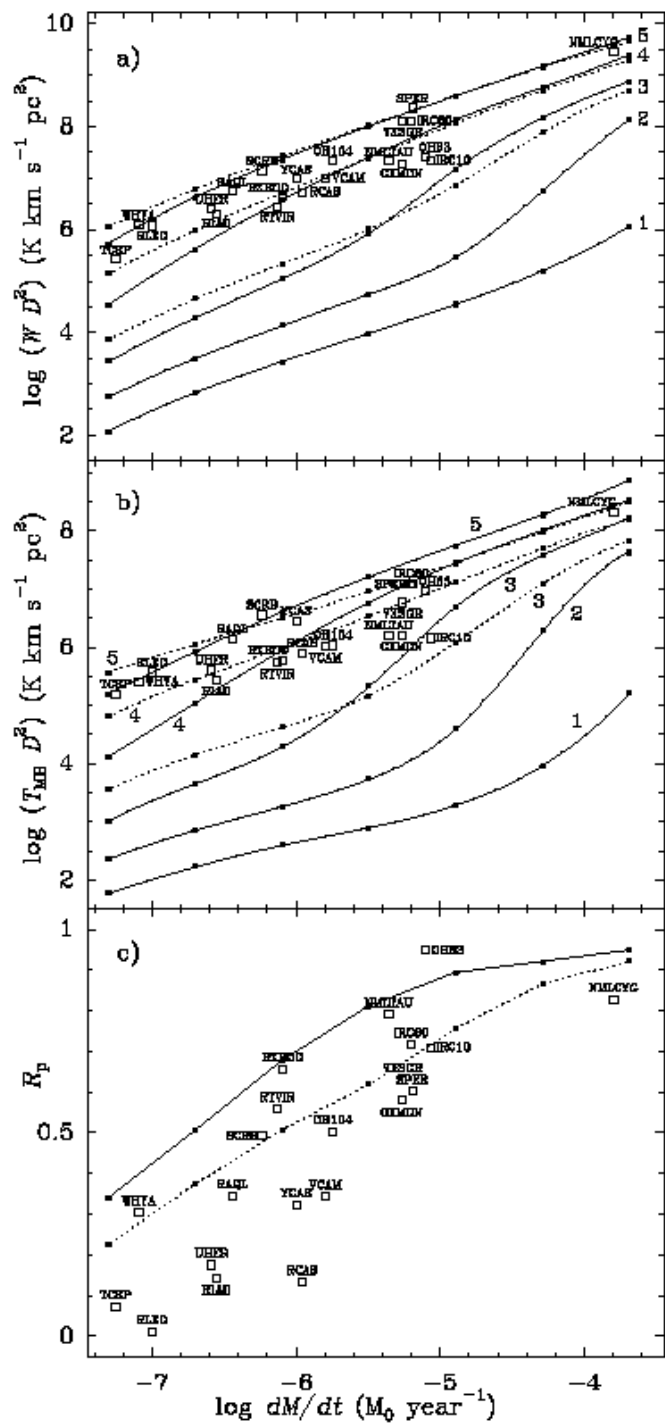


FIG. 12.— $W$ ,  $T_p$ , and  $R_p$  for models that differ in  $\dot{M}$ ,  $x(\text{p-H}_2\text{O})$  and  $v_i$ . Solid lines correspond to models with  $F_i = 10$ , while dashed lines show model results for  $F_i = 20$ .  $v_i$  is determined from  $v_i = 1.9 \times 10^2 \dot{M}^{0.2}$  (Paper I). (a and b) Each curve corresponds to a given value of  $x(\text{p-H}_2\text{O})$ : (1)  $10^{-6}$ ; (2)  $3.3 \times 10^{-6}$ ; (3)  $10^{-5}$ ; (4)  $3.3 \times 10^{-5}$ ; (5)  $10^{-4}$ . (c) Results of calculator with  $x(\text{p-H}_2\text{O}) = 3.3 \times 10^{-5}$  only. The observational values are also shown.

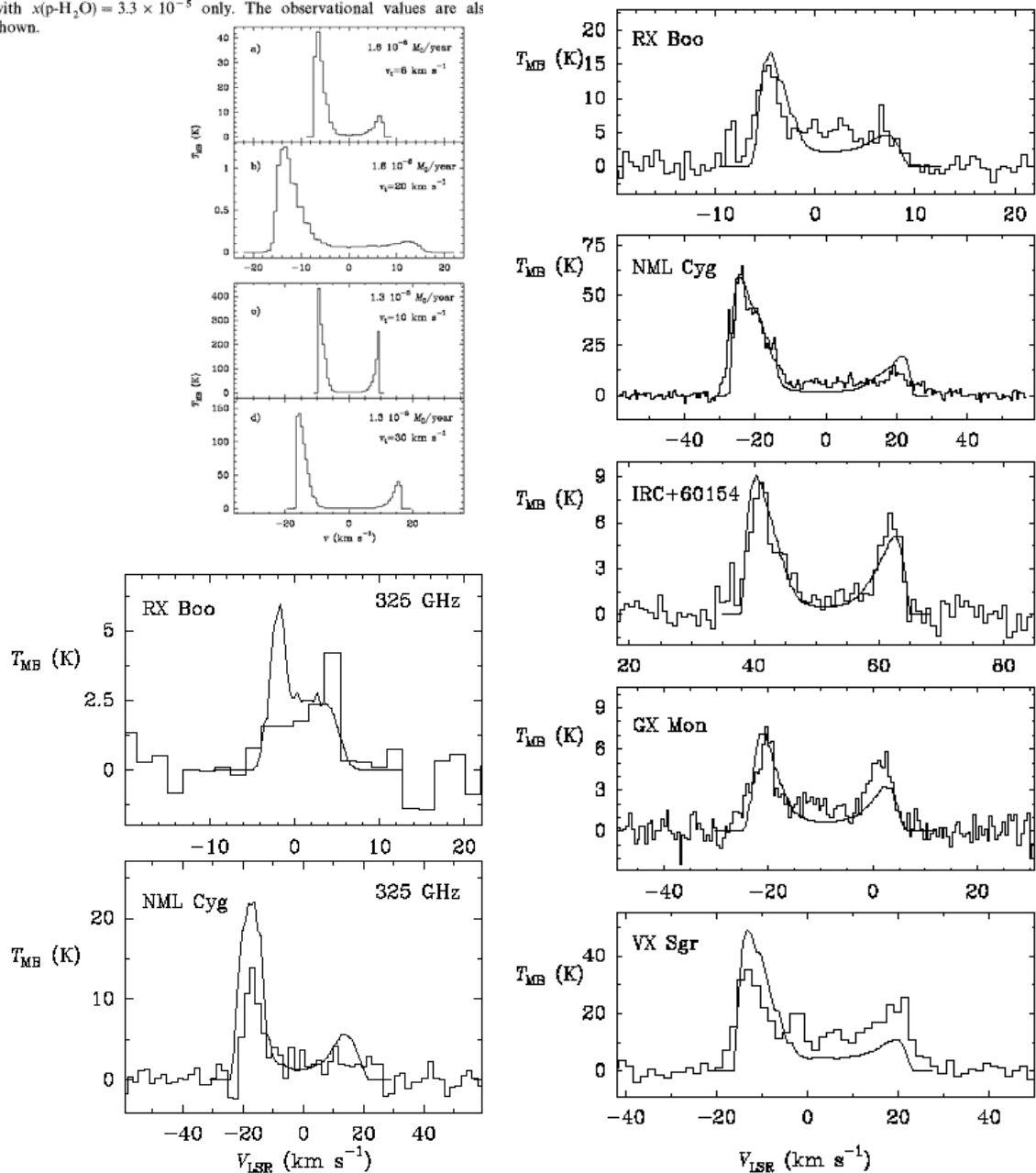
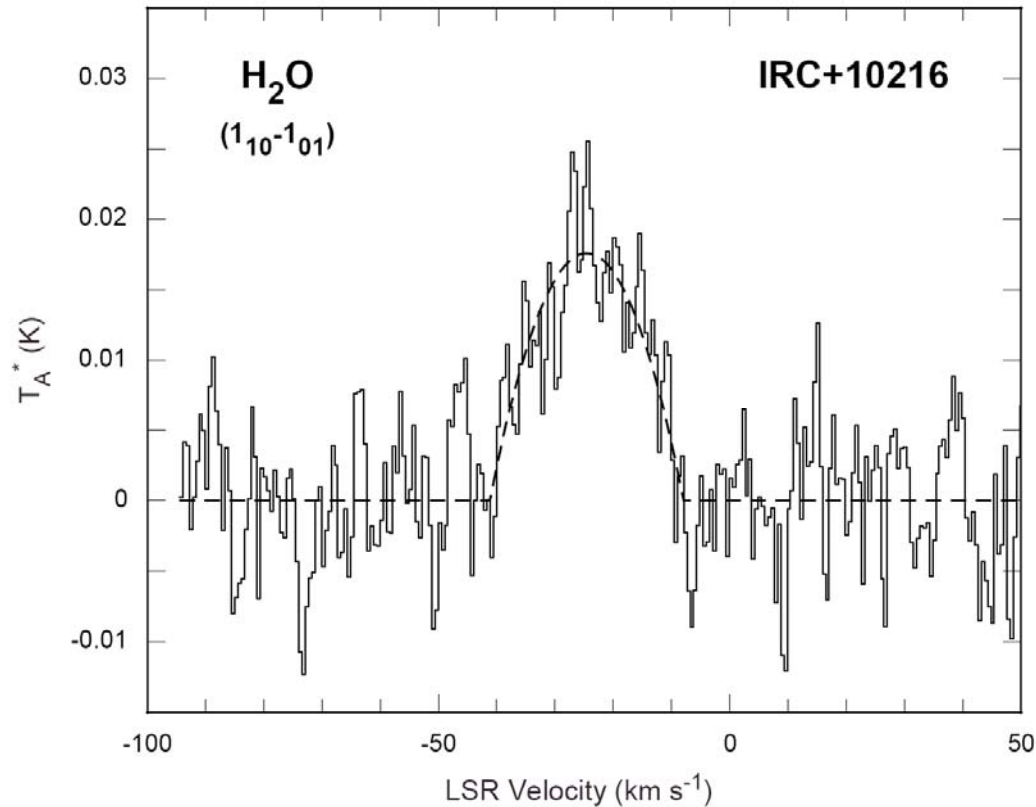


FIG. 13a

# **WATER in C-rich AGB STARS**

## WATER in Carbon rich AGBs stars



Melnick et al., 2001, Nature, 412, 160

Source of water : Comet evaporation

# OXYGEN CHEMISTRY IN THE CIRCUMSTELLAR ENVELOPE OF THE CARBON-RICH STAR IRC +

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Received 2006 March 21; accepted 2006 May 25

## ABSTRACT

In this paper we study the oxygen chemistry in the C-rich circumstellar shells of IRC +10216. The recent discoveries of oxygen-bearing species (water, hydroxyl radical, and formaldehyde) toward this source challenge our current understanding of the chemistry in C-rich circumstellar envelopes. The presence of icy comets surrounding the star or catalysis on iron grain surfaces have been invoked to explain the presence of such unexpected species. This detailed study aims at evaluating the chances of producing O-bearing species in the C-rich circumstellar envelope only by gas-phase chemical reactions. For the hot inner envelope we show that although most of the oxygen is locked in CO near the photosphere (as expected for a C/O ratio greater than 1), for radial distances larger than  $\sim 15$  stellar radii, species such as  $\text{H}_2\text{O}$  and  $\text{CO}_2$  have a large abundance under the assumption of thermochemical equilibrium. It is also shown how non-LTE chemistry makes the  $\text{CO} \rightarrow \text{H}_2\text{O}$ ,  $\text{CO}_2$  transformation predicted in LTE very difficult. Concerning the chemistry in the colder, outer envelope, we show that formaldehyde can be formed through gas-phase reactions. However, in order to form water vapor, it is necessary to include a radiative association between atomic oxygen and molecular hydrogen with quite a high rate constant. The chemical models explain the presence of  $\text{HCO}^+$  and predict the existence of SO and  $\text{H}_2\text{CS}$  (which has been detected in a 3 mm line survey to be published). We have modeled the line profiles of  $\text{H}_2\text{CO}$ ,  $\text{H}_2\text{O}$ ,  $\text{HCO}^+$ , SO, and  $\text{H}_2\text{CS}$  using a nonlocal radiative transfer model and the abundance profiles predicted by our chemical model. The results have been compared to the observations and discussed.

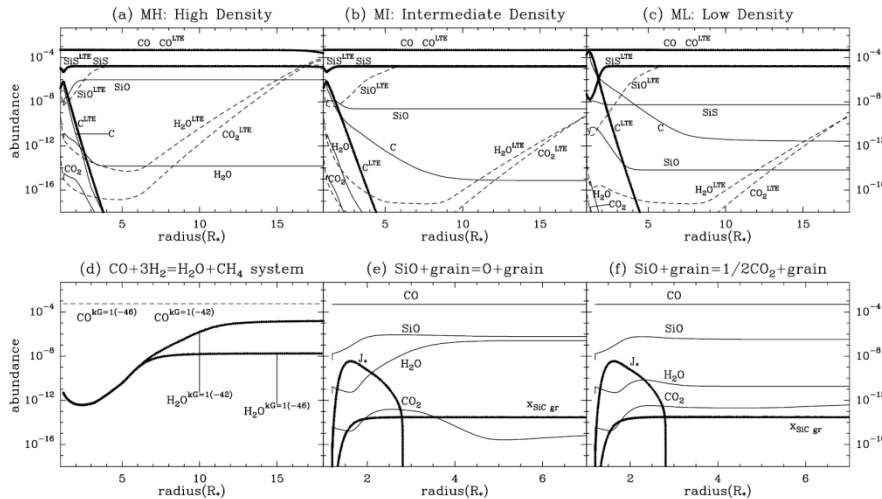


FIG. 3.—Abundances, relative to total number of hydrogen nuclei, of some O-bearing molecules. (a), (b), and (c) Abundances given by LTE (dotted and dashed lines) and by chemical kinetics (solid lines) assuming a constant velocity expansion of  $1 \text{ km s}^{-1}$  and the density profiles (a) MH, (b) MI, and (c) ML. (d) Abundances of CO and  $\text{H}_2\text{O}$  for a CO,  $\text{H}_2$ ,  $\text{CH}_4$ , and  $\text{H}_2\text{O}$  system with  $k_2$  (see text) equal to  $10^{-42}$  and  $10^{-46} \text{ cm}^3 \text{ s}^{-1}$ . (e) and (f) Abundances given by chemical kinetics with the density profile MH when we consider that 90% of SiO gas is deposited onto SiC grains and that the corresponding oxygen is released to the gas phase either as (e) atomic oxygen or as (f)  $\text{CO}_2$ . The nucleation rate of SiC grains  $J_{\text{SiC}}$  (with units of nuclei  $\text{cm}^{-3} \text{ s}^{-1}$ ) and its abundance  $x_{\text{SiC,gr}}$ , relative to total number of hydrogen nuclei, are also plotted as dotted lines. [See the electronic edition of the Journal for a color version of this figure.]

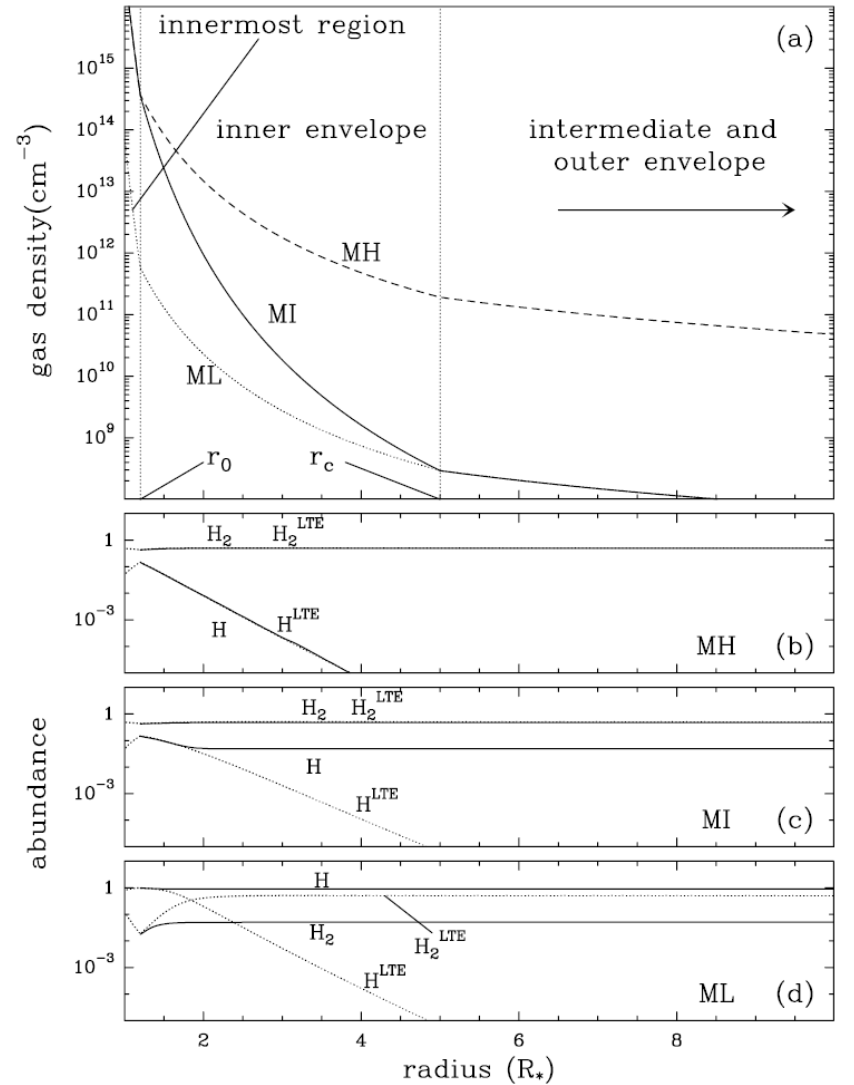


FIG. 1.—(a) Three different density profiles considered for the inner and innermost envelope: “MH” [ $K_n = 1$ ,  $n(r_0) = 3.7 \times 10^{14} \text{ cm}^{-3}$ ], “MI” [ $K_n = 1.8565$ ,  $n(r_0) = 3.7 \times 10^{14} \text{ cm}^{-3}$ ] and “ML” [ $K_n = 1$ ,  $n(r_0) = 1.2 \times 10^{12} \text{ cm}^{-3}$ ]. All profiles use a  $r^{-2}$  law for  $r > r_c$ . Panels (b), (c), and (d) show the abundances of H and  $\text{H}_2$ , relative to total number of hydrogen nuclei [ $n(\text{H}) + 2n(\text{H}_2)$ ], as given by LTE (dotted lines) and by chemical kinetics (solid lines) for the three density profiles, (b) MH, (c) MI, and (d) ML. Note the strong dependence of the  $\text{H}/\text{H}_2$  ratio on the densities considered.

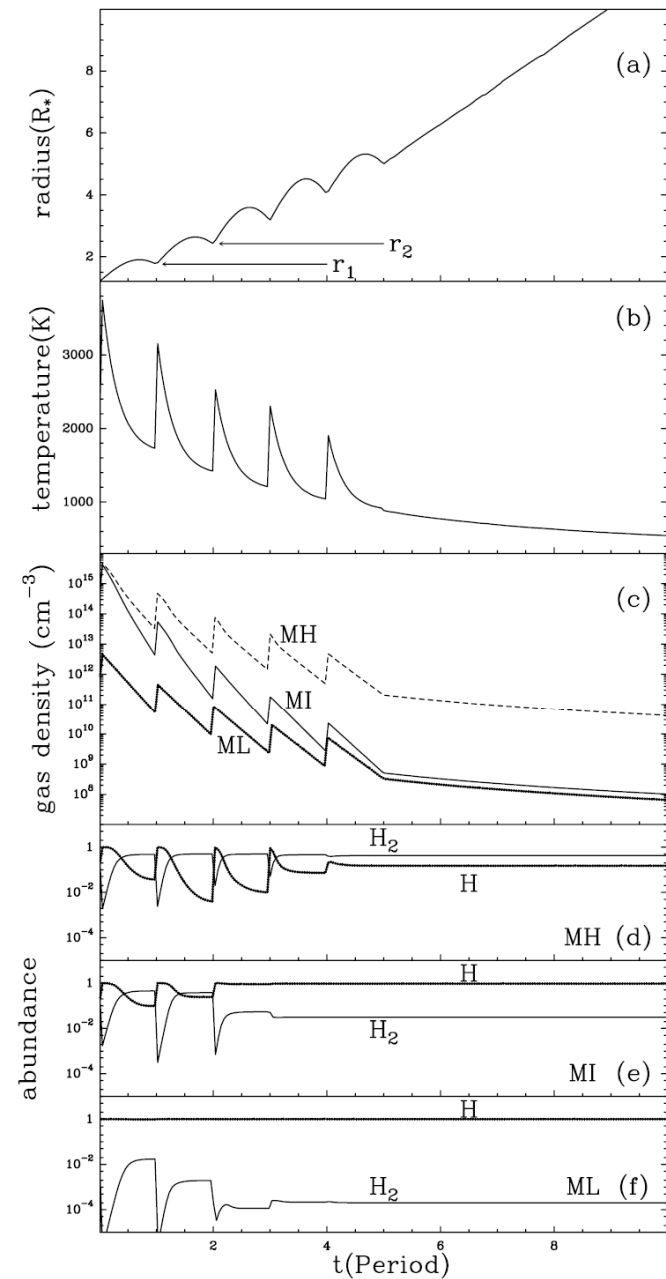


FIG. 2.—Parameters for a volume element of gas that suffers a history of five shocks: (a) trajectory; (b) temperature history; (c) density histories for models MH, MI, and ML. Panels (d), (e), and (f) show the H and  $\text{H}_2$  abundances, relative to total number of hydrogen nuclei, calculated by chemical kinetics for the three density laws shown in (c).

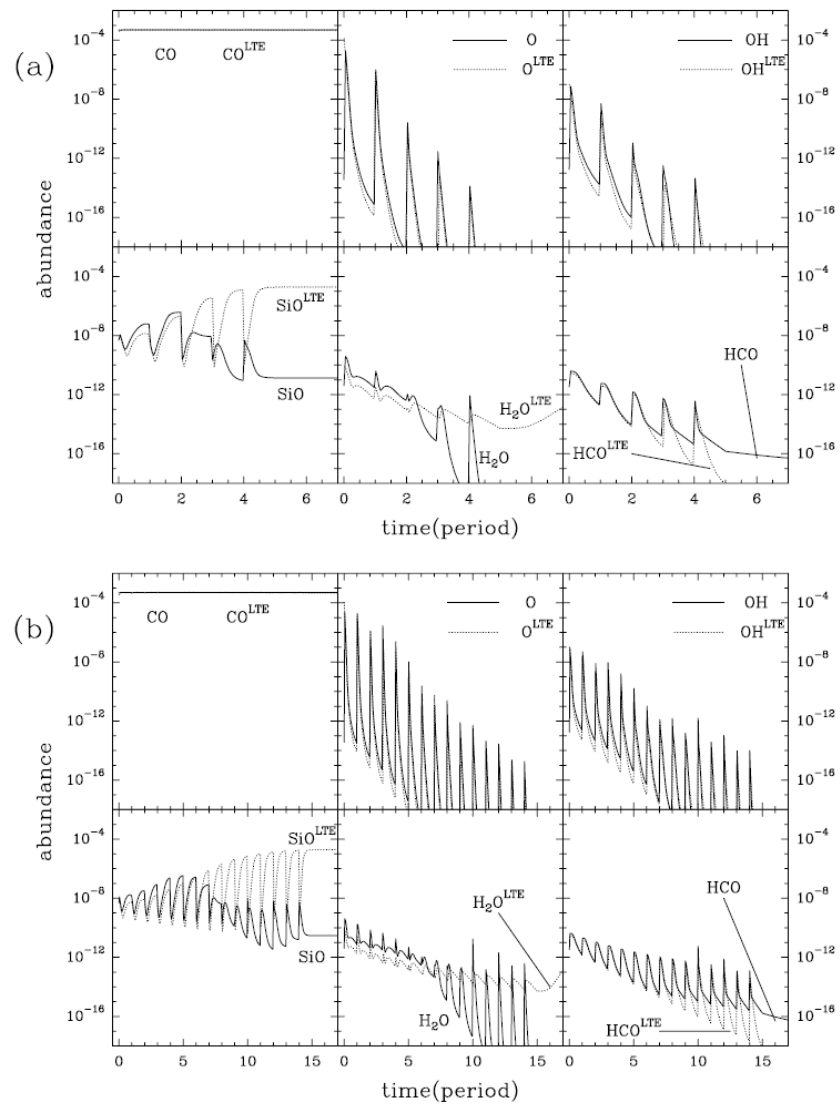


FIG. 4.—Evolution of the abundances, relative to total number of hydrogen nuclei, of some O-bearing molecules for a (a) 5 shock history ( $N_{\text{shocks}} = 5$ ) and (b) 15 shock history ( $N_{\text{shocks}} = 15$ ). The history starts at  $r_0$  ( $t = 0$ ) and reaches  $r_c$  at time  $N_{\text{shocks}} P$ . The LTE abundances for the physical conditions of the gas at each instant are shown as dotted lines.

**Results in agreement with previous models  
by Cherchneff**

## The role of infrared pumping

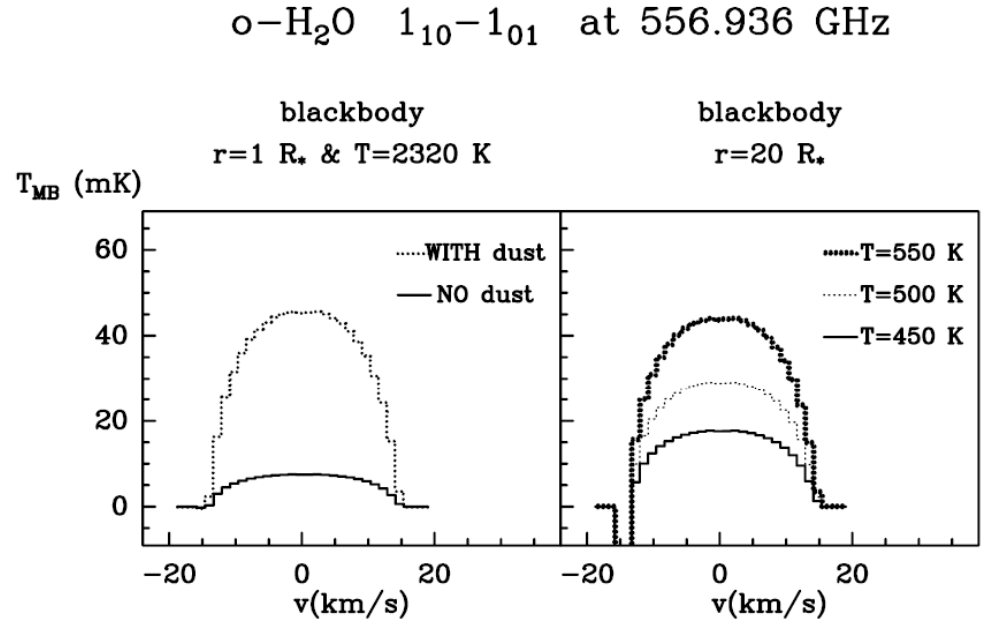
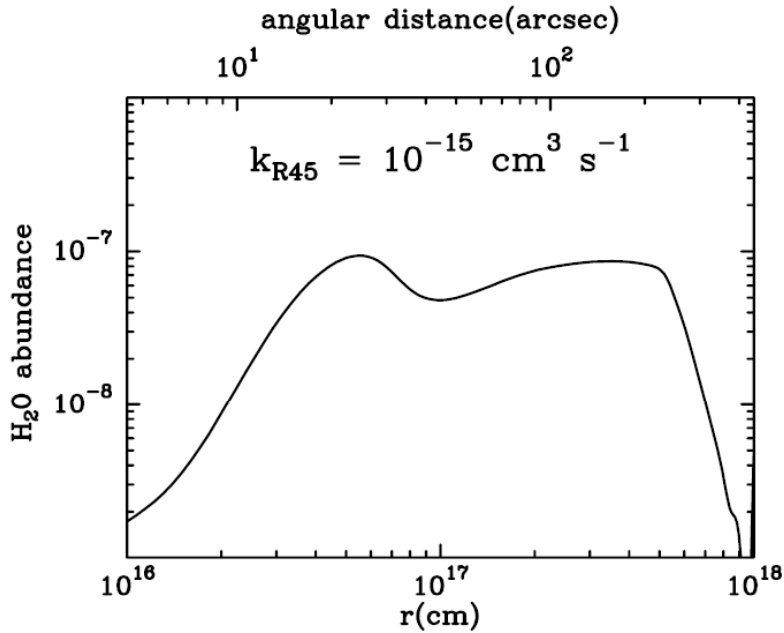


FIG. 7.—*Left*: Water vapor abundance, relative to total number of hydrogen nuclei, when assuming a rate constant for (R45) of  $10^{-15} \text{ cm}^3 \text{ s}^{-1}$ . *Right*: Calculated line intensities and profiles for the  $1_{1,0}-1_{0,1}$  transition of ortho- $\text{H}_2\text{O}$  for different IR fluxes. The *Odin* beam size ( $\sim 2'$ ) has been considered, and the lines have been smoothed to a resolution of  $1.3 \text{ km s}^{-1}$ .

LETTER TO THE EDITOR

# Water in IRC+10216: a genuine formation process by shock-induced chemistry in the inner wind<sup>★</sup>

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Received 31 October 2010 / Accepted 22 December 2010

## ABSTRACT

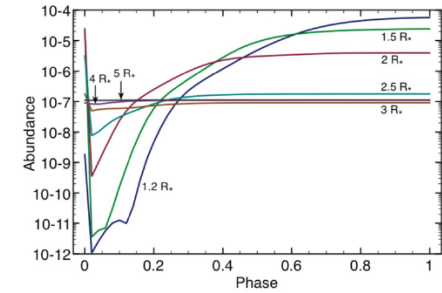
**Context.** The presence of water in the wind of the extreme carbon star IRC+10216 has been confirmed by the *Herschel* telescope. The regions where the high- $J$  H<sub>2</sub>O lines have been detected are close to the star at radii  $r \leq 15 R_*$ .

**Aims.** We investigate the formation of water and related molecules in the periodically-shocked inner layers of IRC+10216 where dust also forms and accelerates the wind.

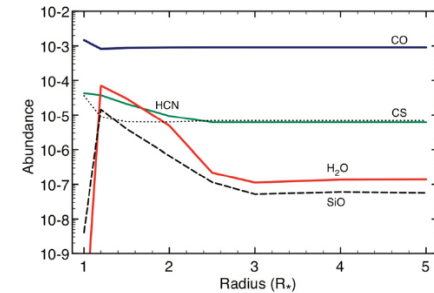
**Methods.** We describe the molecular formation by a chemical kinetic network involving carbon- and oxygen-based molecules. We then apply this network to the physical conditions pertaining to the dust-formation zone which experiences the passage of pulsation-driven shocks between 1 and  $5 R_*$ . We solve for a system of stiff, coupled, ordinary, and differential equations.

**Results.** Non-equilibrium chemistry prevails in the dust-formation zone. H<sub>2</sub>O forms quickly above the photosphere from the synthesis of hydroxyl OH induced by the thermal fragmentation of CO in the hot post-shock gas. The derived abundance with respect to H<sub>2</sub> at  $5 R_*$  is  $1.4 \times 10^{-7}$ , which excellently agrees with the values derived from *Herschel* observations. The non-equilibrium formation process of water will be active whatever the stellar C/O ratio, and H<sub>2</sub>O should then be present in the wind acceleration zone of all stars on the Asymptotic Giant Branch.

**Key words.** astrochemistry – molecular processes – stars: low-mass – stars: carbon – stars: AGB and post-AGB

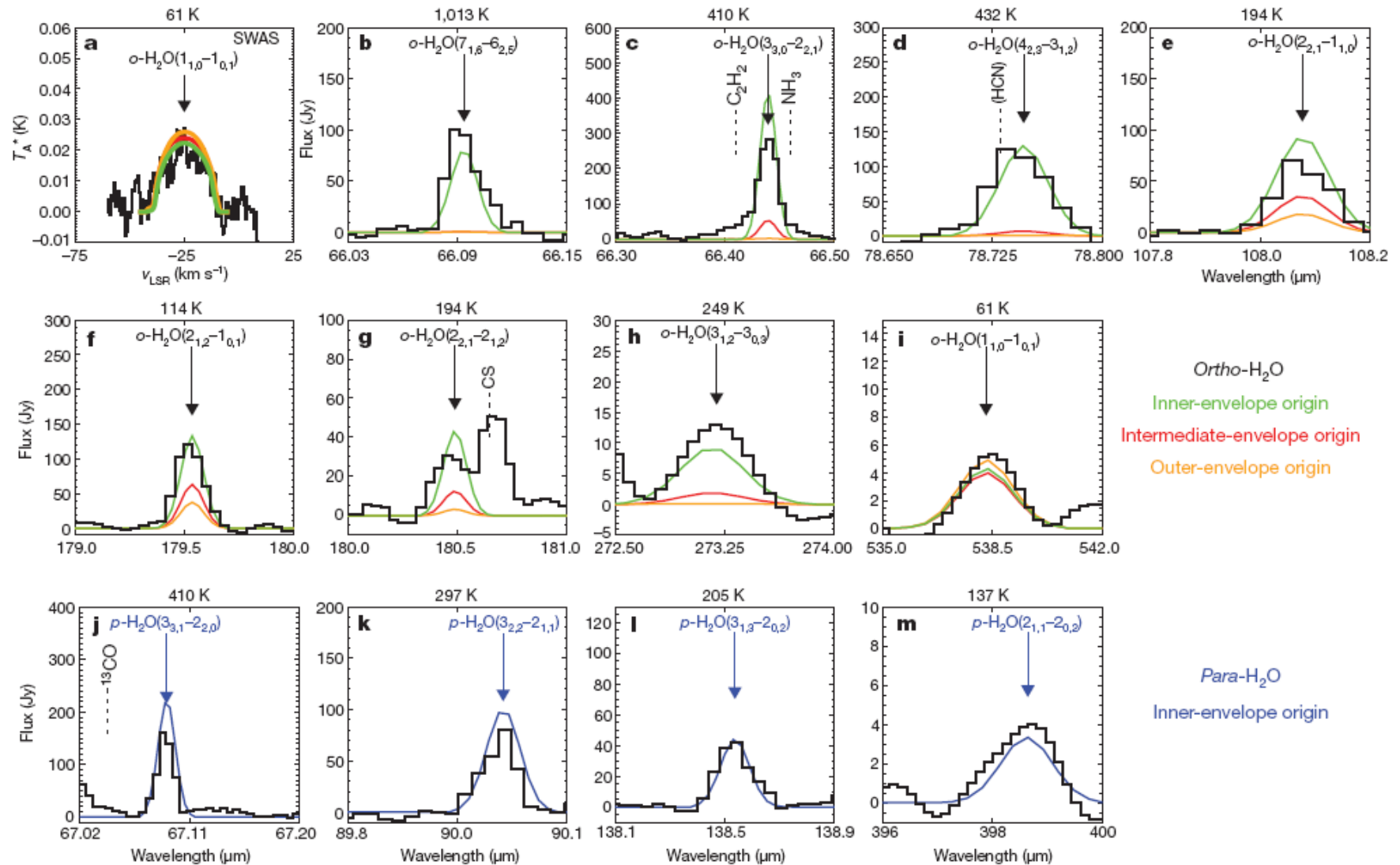


**Fig. 1.** H<sub>2</sub>O abundances with respect to total gas as a function of pulsation phase  $\theta$  and radius in the inner wind (shocks form at  $\theta = 0$  and  $\theta = 1$ ).



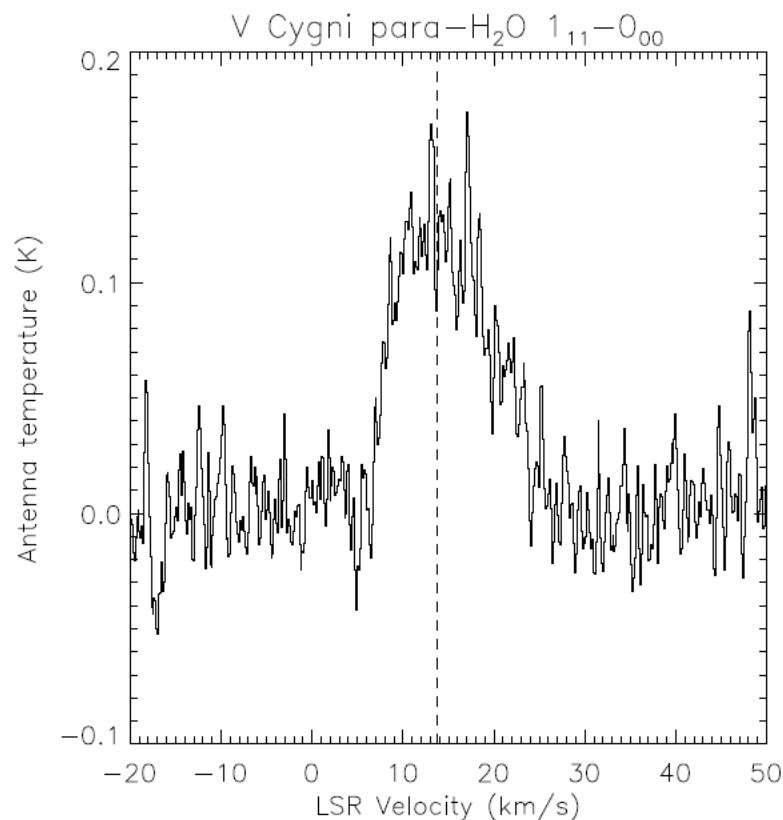
**Fig. 2.** Abundances with respect to H<sub>2</sub> of key molecules including H<sub>2</sub>O taken at  $\theta = 1$ , as a function of radius. Values at  $r = 1 R_*$  are derived from thermodynamical equilibrium calculations.

# Far-IR data from PACS/Herschel



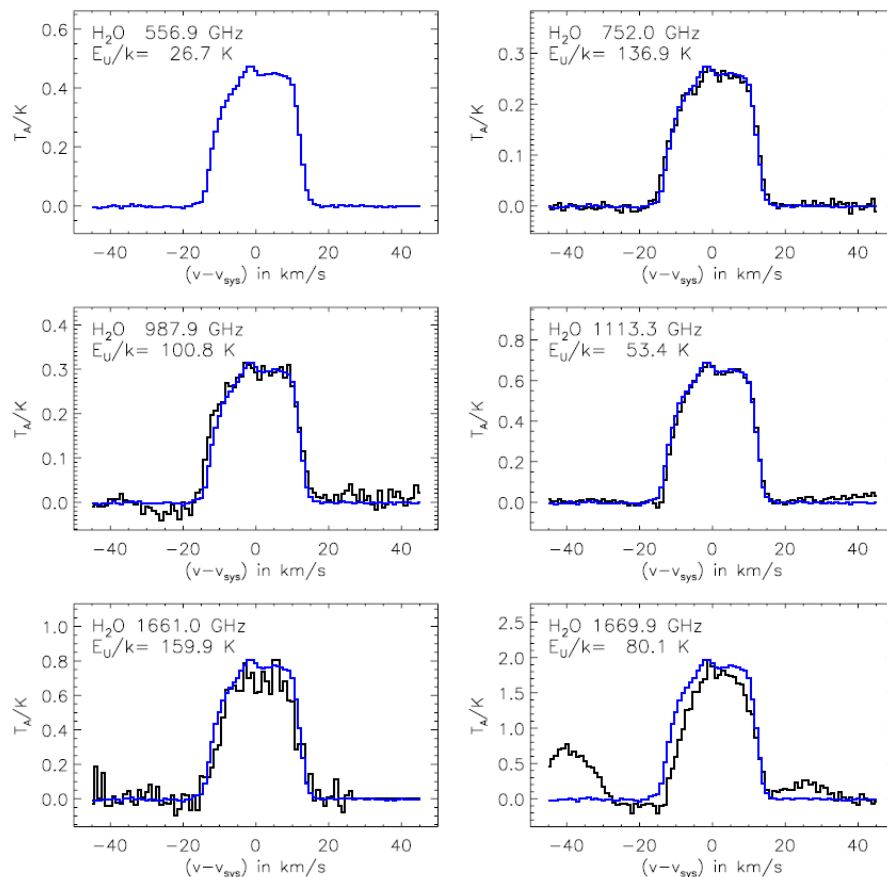
From Decin et al., 2010, Nature, 467, 64

# Herschel/HIFI A&A, 521, L5



**Fig. 1.** Continuum-subtracted spectrum of para-H<sub>2</sub>O 1<sub>11</sub>-0<sub>00</sub> obtained toward V Cygni. The vertical dashed line indicates the LSR velocity of the source, as determined by BW01 from observations of the CO  $J = 2-1$  line.

## Neufeld et al., Water in IRC+10216 ApJ, 2011, 727, L28



# IRC+10216 at a distance of 100 pc

0.05"

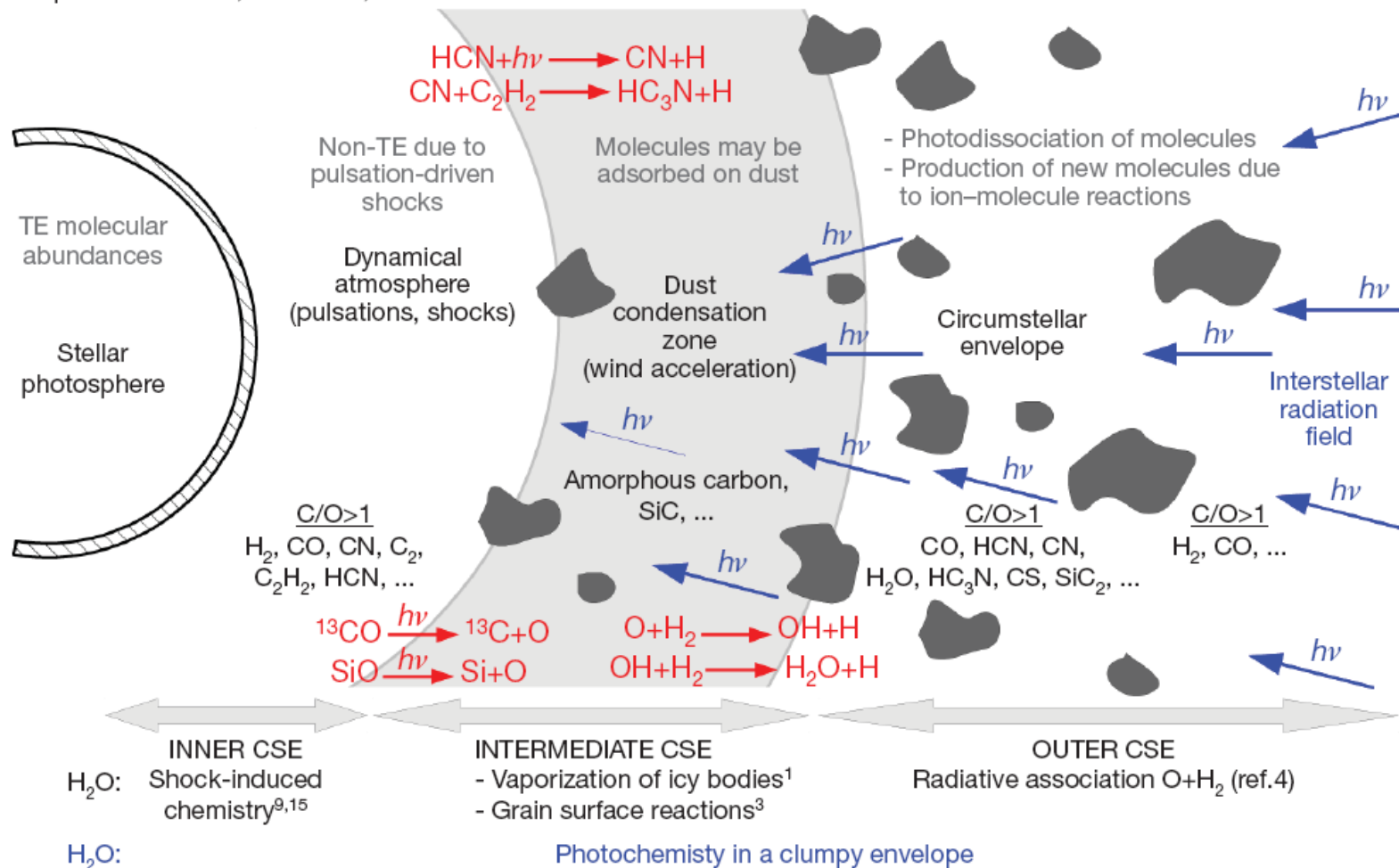
0.25"

1.5"

5.0"

400"

Weak CO  
emission  
detected up to  
R=300"



- I) Water produced in inner shocks (I. Cherchneff, 2011, A&A, 526, L11)
- II) Oxygen atoms available from the photodissociation of  $^{13}\text{CO}$  and SiO

**Agúndez & Cernicharo 2006**

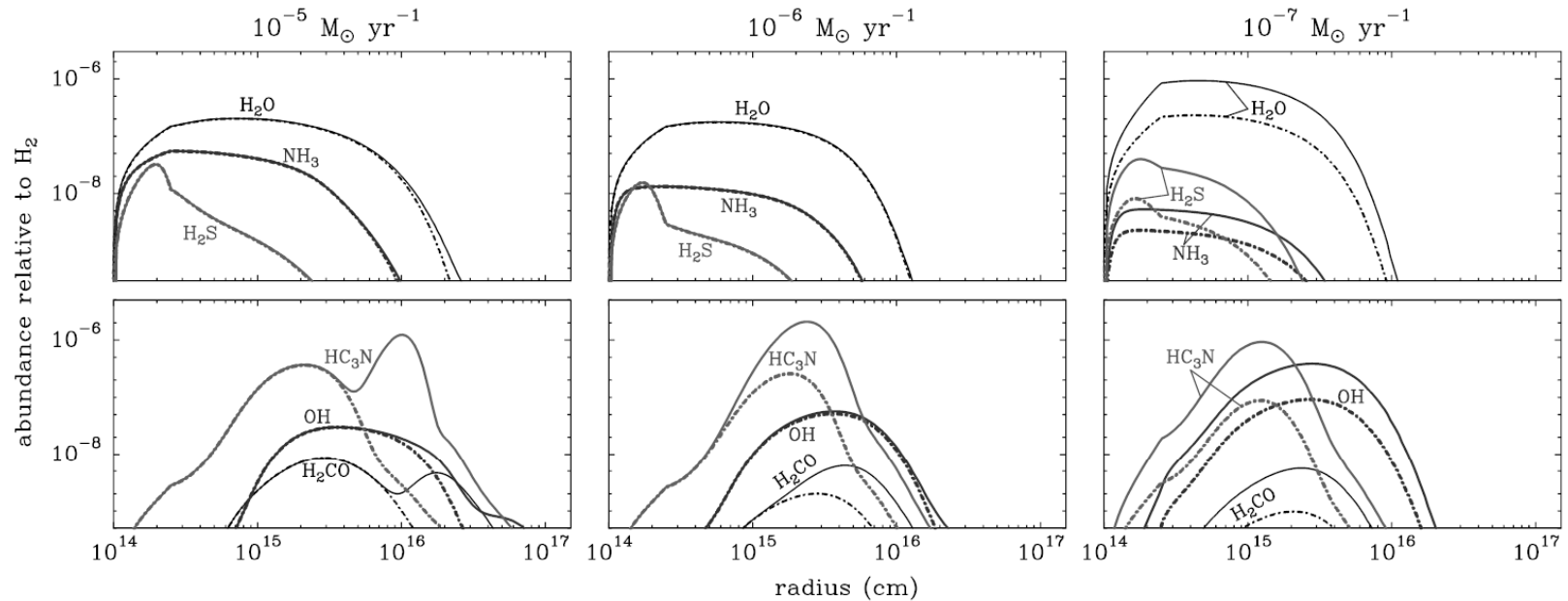


Fig. 2.— Calculated abundance of several molecules as a function of radius for carbon-rich CSEs with mass loss rates of  $10^{-5}$ ,  $10^{-6}$ , and  $10^{-7} \text{ M}_{\odot} \text{ yr}^{-1}$ . Dashed-dotted lines correspond to the abundance of the minor UV exposed component and continuous lines to the abundance weighted-averaged over the minor and major components.

**Detailed chemical modeling of clumpy envelopes :  $\text{H}_2\text{O}$ , OH and  $\text{H}_2\text{CO}$  could be abundant in C-rich AGBs.  $\text{HC}_3\text{N}$  abundance enhanced in the inner envelope.**

**Decin et al., 2010, Nature; Agúndez, Cernicharo & Guélin 2010, ApJ, 724, L133,**

**CRL618**

# **Carbon Rich Protoplanetary Nebula**

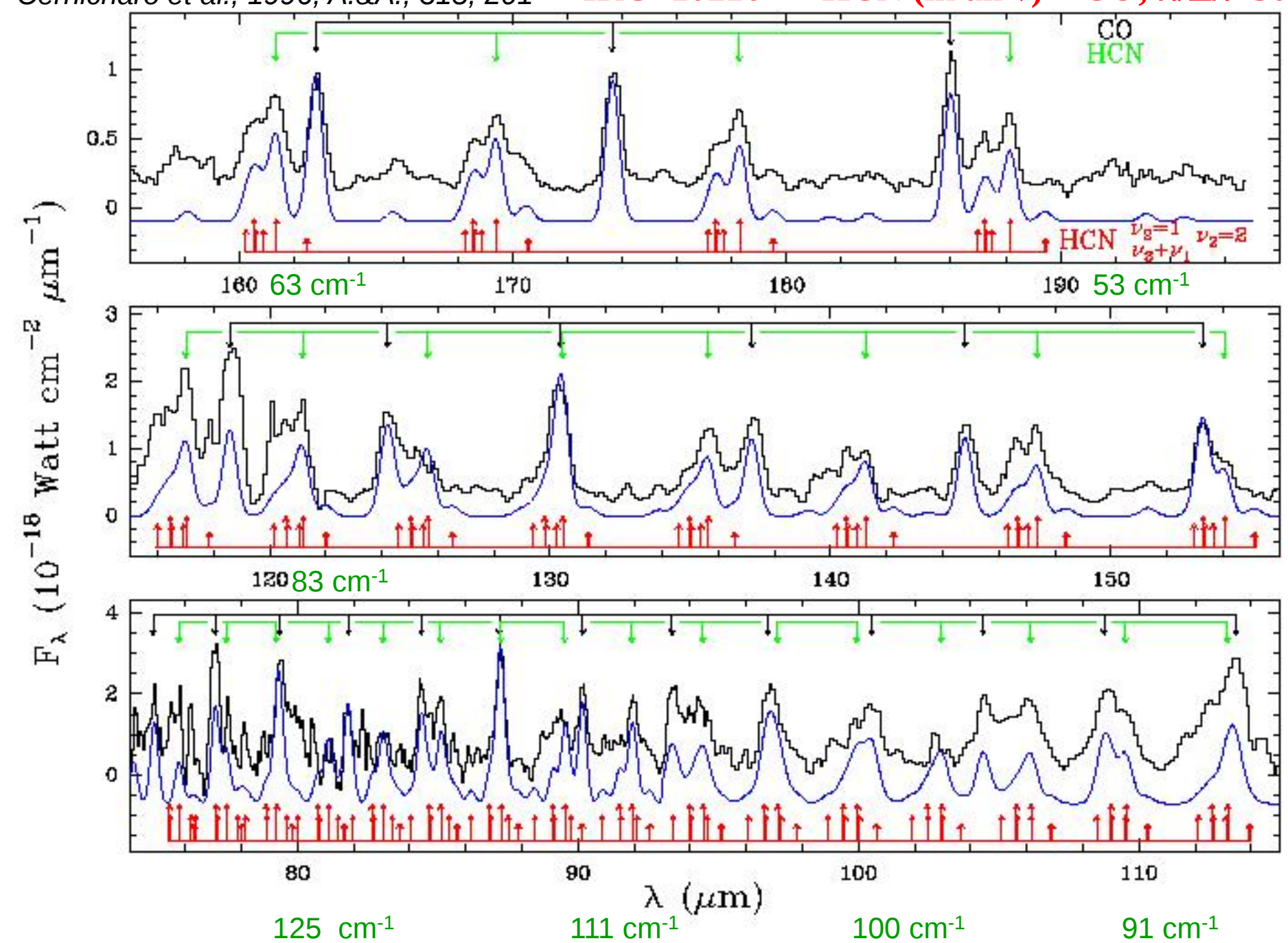
**$H_2O$ ,  $HCO^+$ ,  $OH$  and  $H_2CO$**

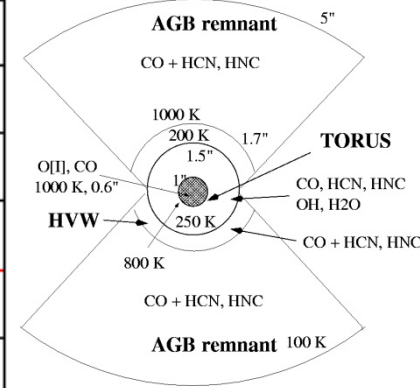
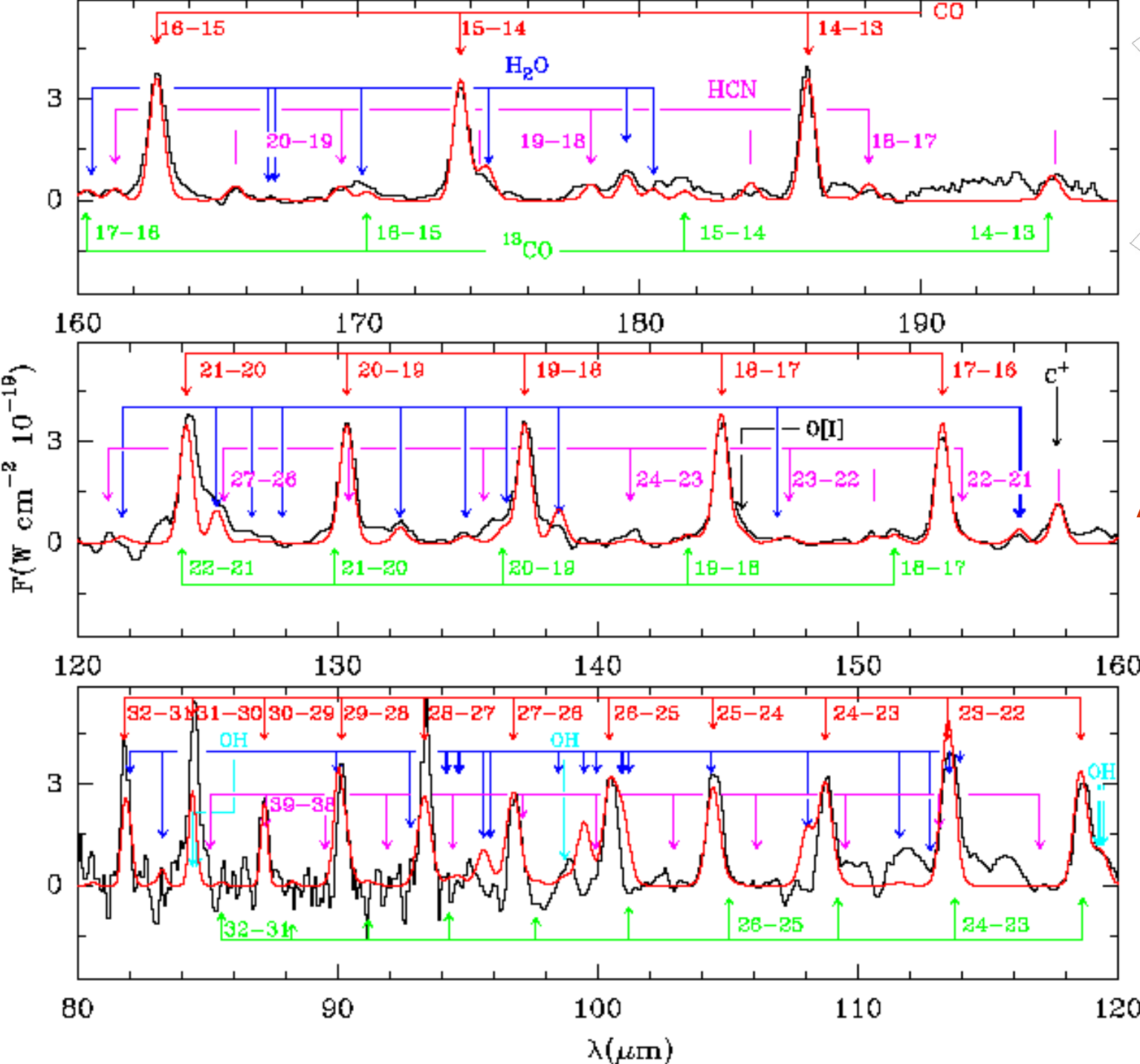
***Herpin & Cernicharo 2000, 2001***

***(why only these O-bearing species ?)***

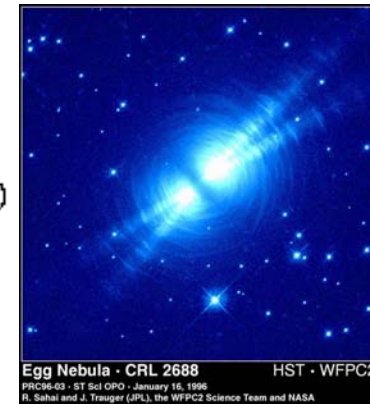
***How to interpret a line survey close to  
the line confusion limit ?***

**WORK DONE AT DAMIR (CSIC; MADRID) by Juan Ramón PARDO  
and collaborators (J. Cernicharo, J.R. Goicoechea, M. Guélin, A. Asensio)**





## In C-rich AGB stars

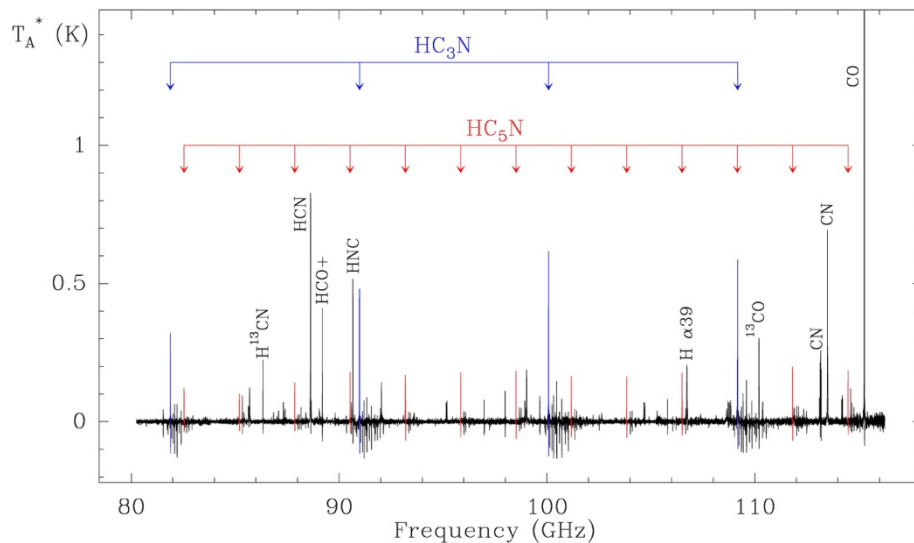


Herpin & Cernicharo,  
2000, *ApJ Letters*, 530,  
L129

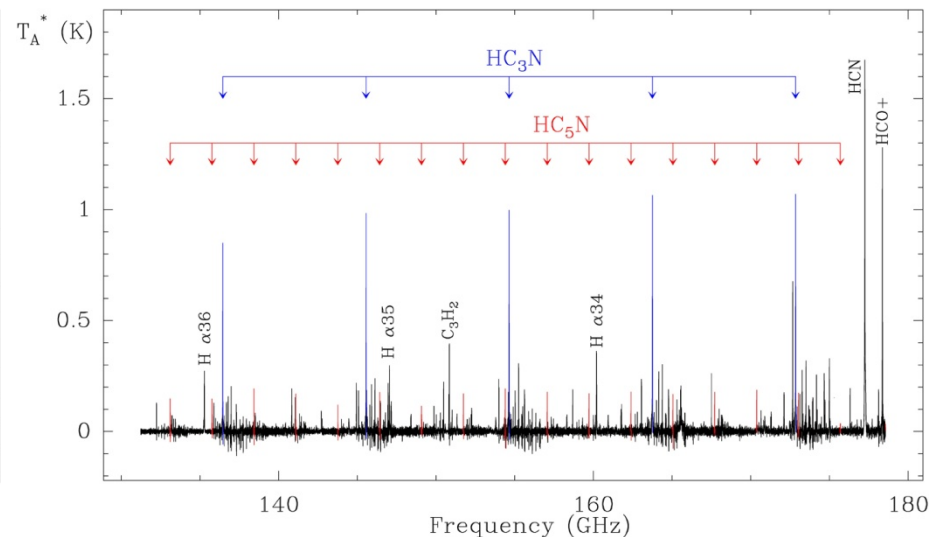
See Chemical Modelling by  
Cernicharo 2004, *ApJ Lett.*

# CRL618 Spectral Line Survey

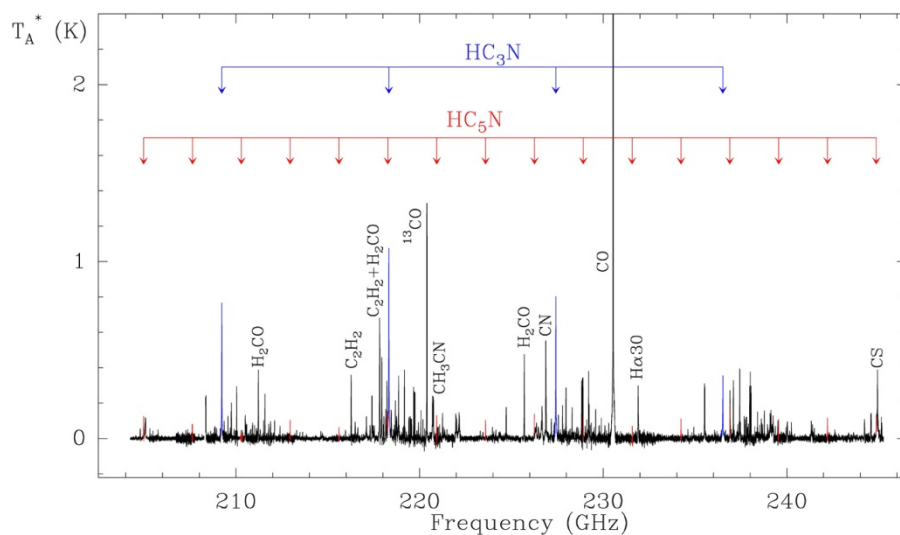
~ 3 mm



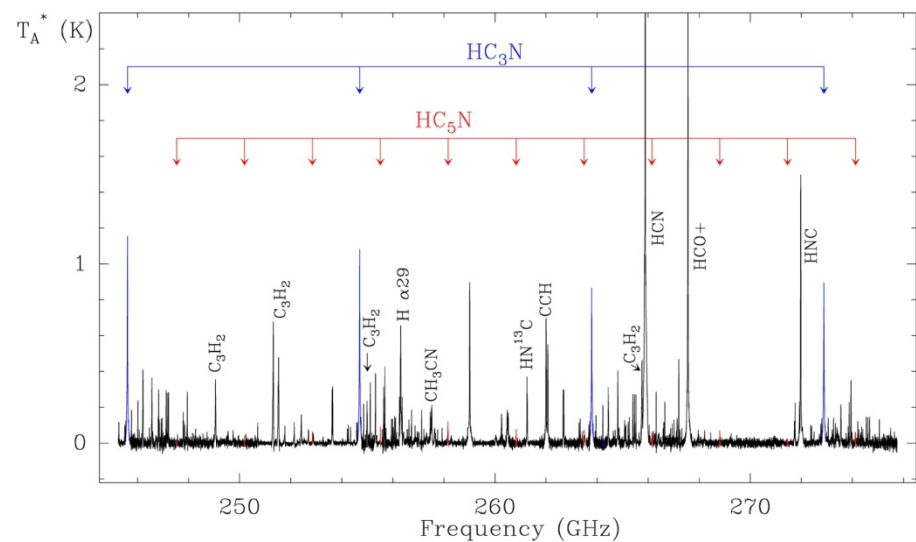
~ 2 mm



~ 1.3 mm



~ 1.0 mm



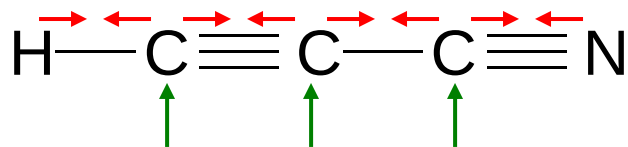
# The line survey

- Around 2500 líneas from 24 molecular species + recombination lines of H y He.
- Initially more than 80% of lines were U. Very soon we realized that the spectrum is dominated by  $\text{HC}_3\text{N}$  &  $\text{HC}_5\text{N}$  which are responsible for 85% of the total number of detected lines.
- Most lines have P-Cygni profiles. They transform into pure emission at high frequency. Some molecules do not show any absorption components, even at low freq.
- Abundant species show broad linewidths ( $\text{CO}$ ,  $\text{HCN}$ ,  $\text{HC}_3\text{N}$ ,  $\text{HCO}^+$ )

The large number of transitions for each molecule allows to reconstruct the morphology of the object:

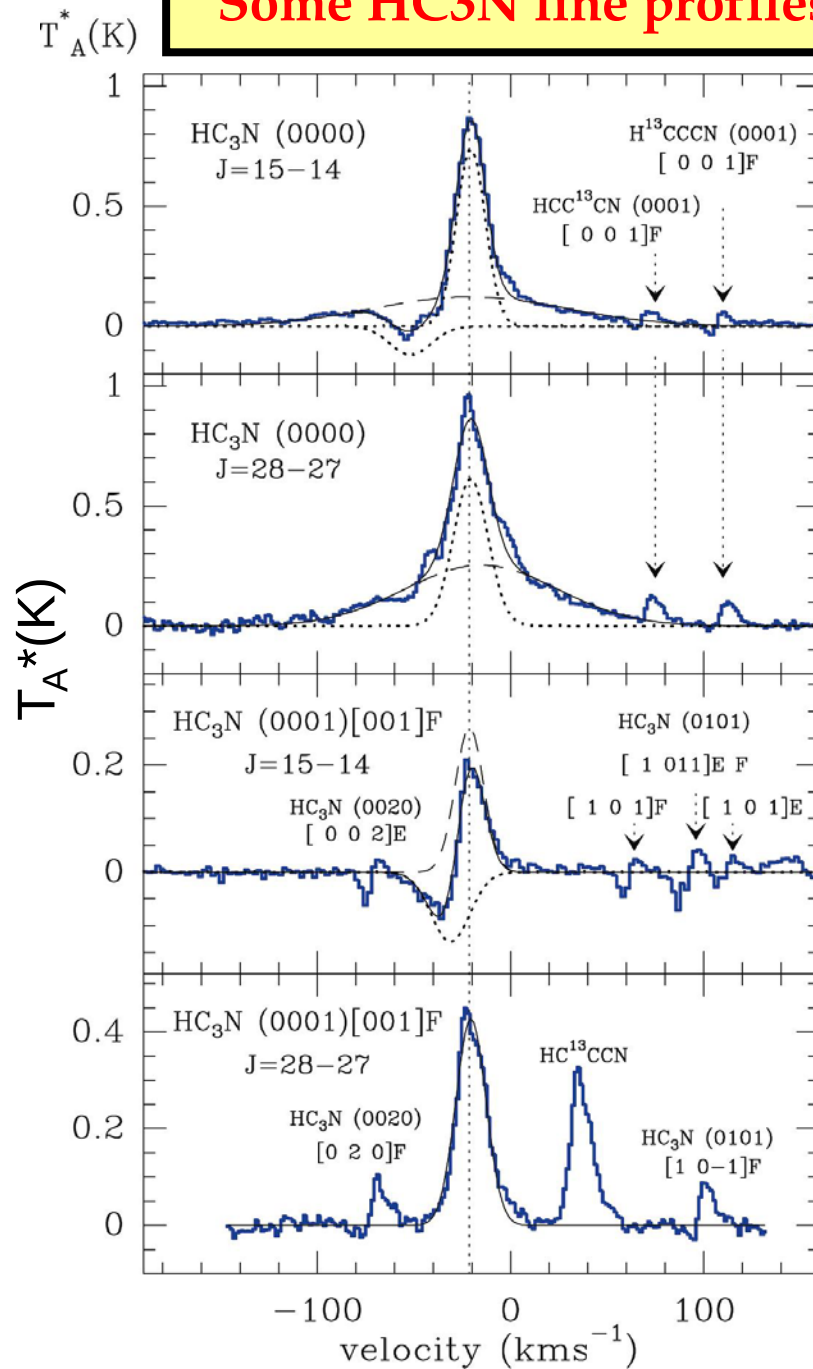
**MORPHOLOGY – PHYSICAL CONDITIONS – CHEMICAL ABUNDANCES.**

# Spectroscopy of HC<sub>3</sub>N



- 3 bending modes and four stretching modes.
- Bending modes: ( $\nu_7, \nu_6, \nu_5$ ): Energies below 663 cm<sup>-1</sup>
- Stretching mode  $\nu_4$  at 881 cm<sup>-1</sup>. The other are above 2000 cm<sup>-1</sup> (and not detected in CRL618).
- 14 vib states detected in CRL618 with energies up to 1100 cm<sup>-1</sup>.
- The line profiles show : high velocity wind, low velocity expanding envelope, P-Cygni profiles decreasing when increases.

## Some HC<sub>3</sub>N line profiles



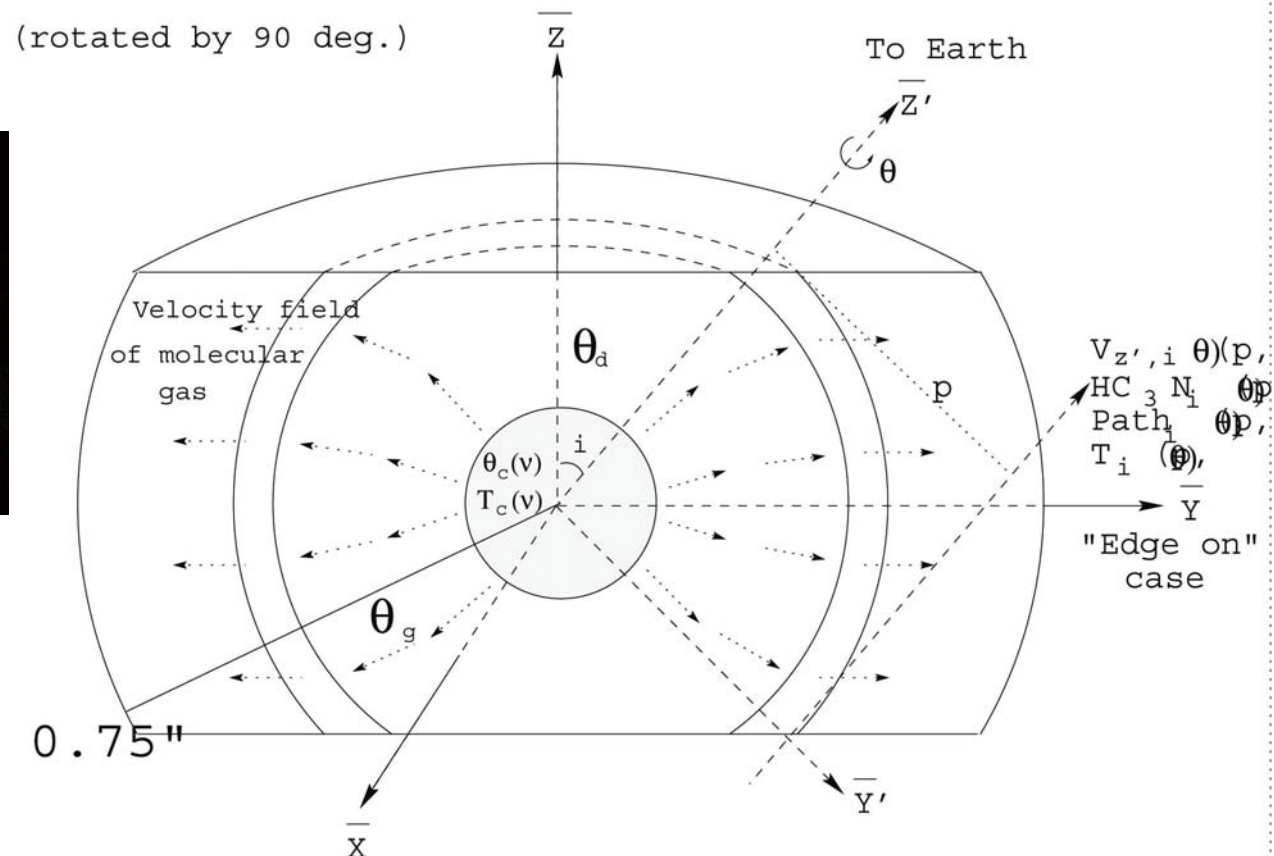
# Reconstructing CRL618 step by step

## Region responsible for the P-Cygni (inner envelope)

**Tracer : code for HC<sub>3</sub>N in vib excited states**

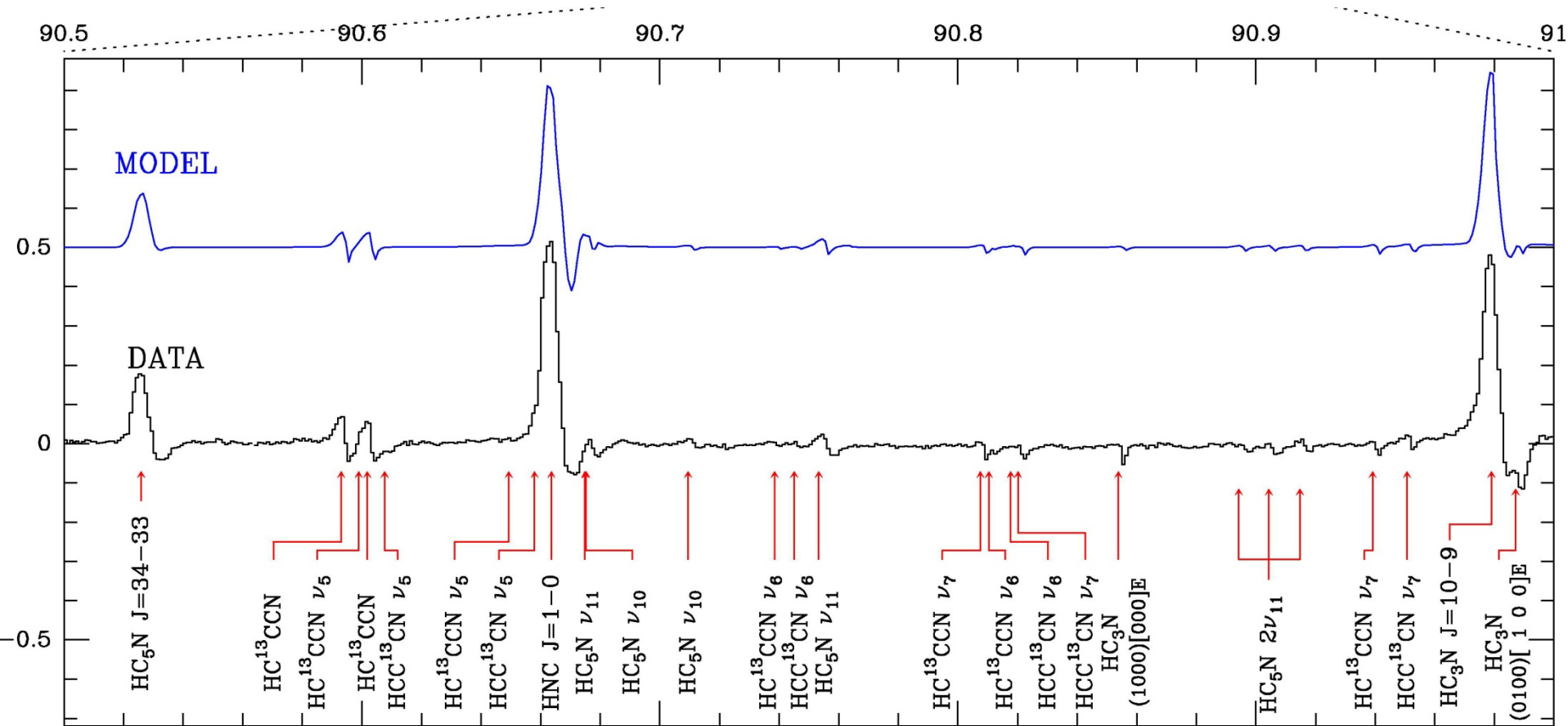


**Help: Previous studies, continuum and lines**

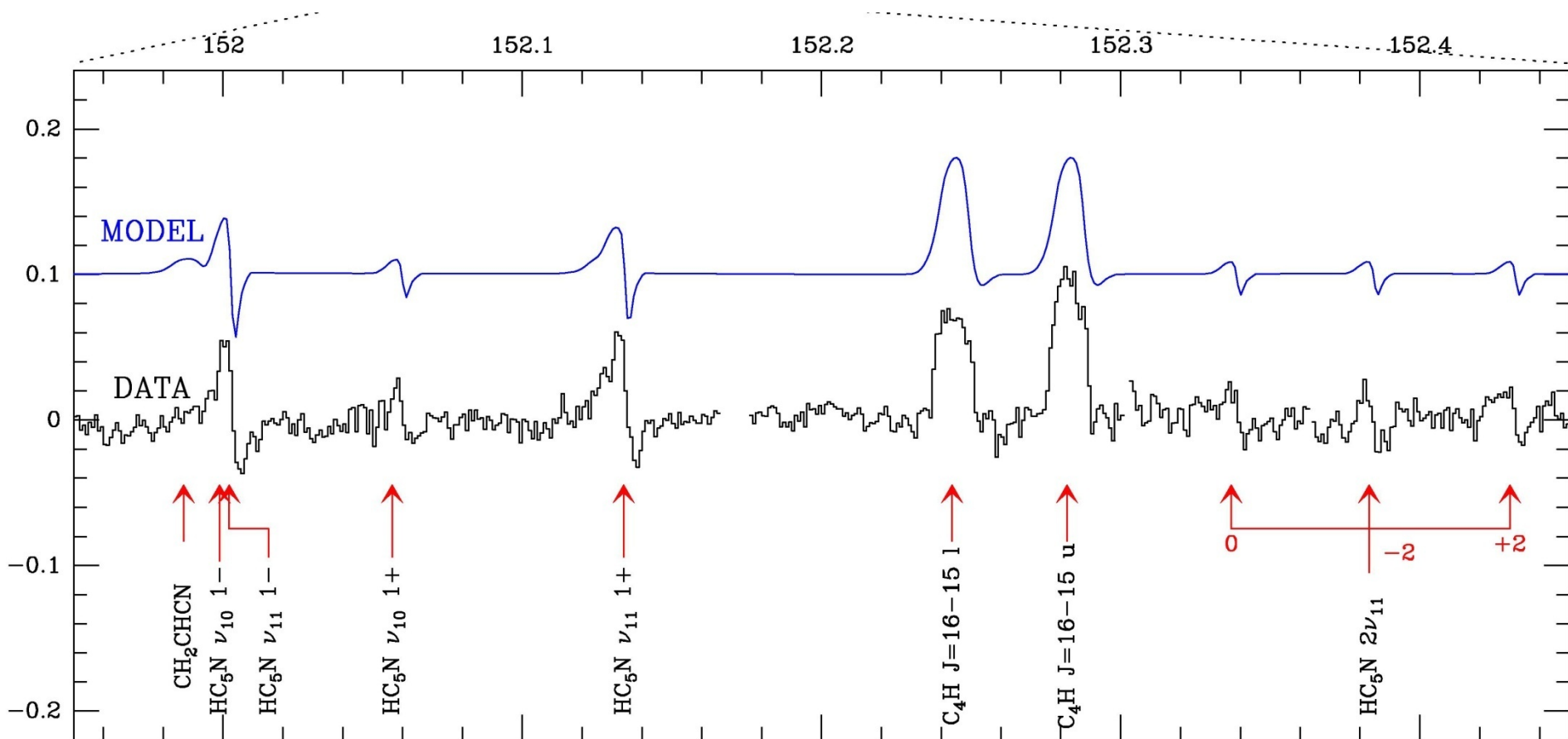


**Pardo & Cernicharo 2007, ApJ 654; Pardo et al., 2004, ApJ, 615, 495 & 2005, ApJ, 628, 275;**

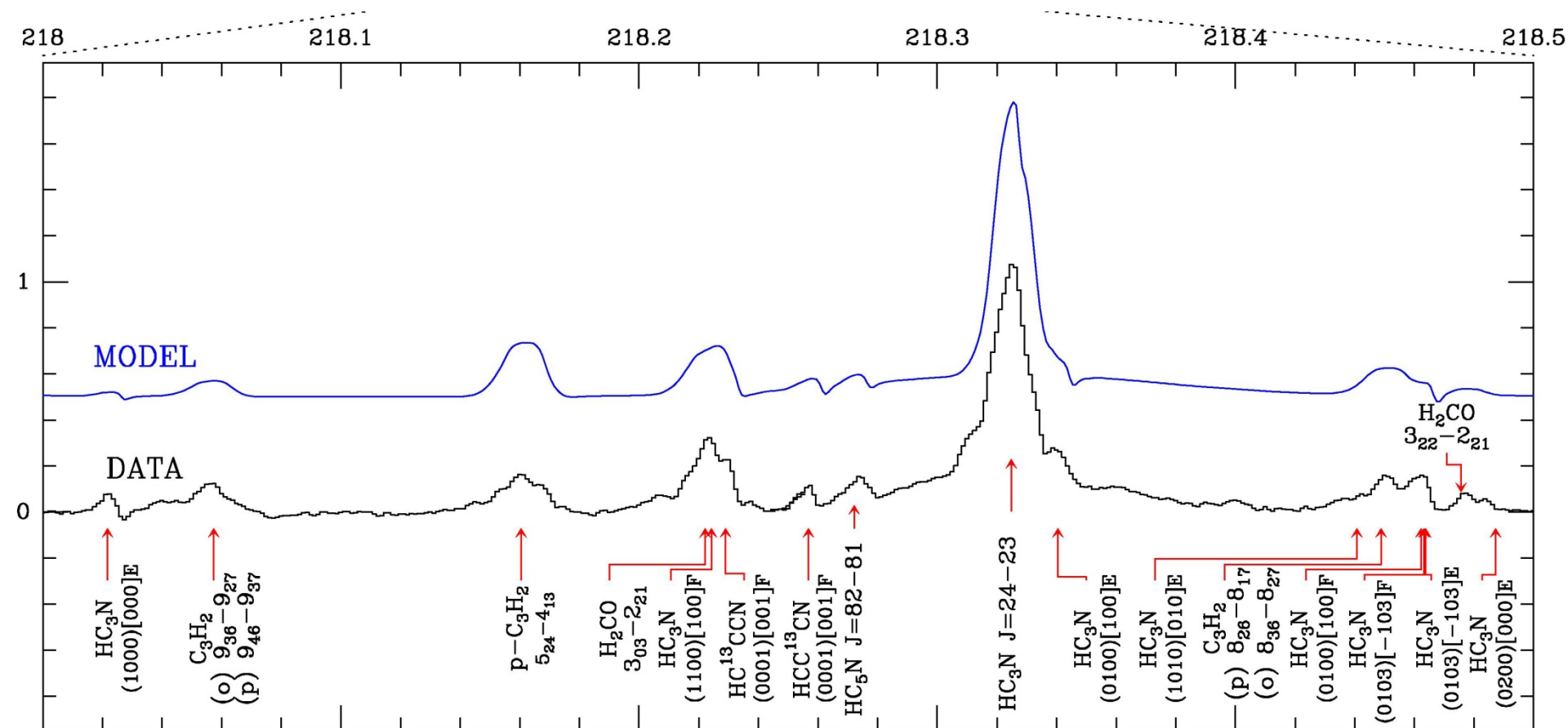
# 3 mm window : Data and final model



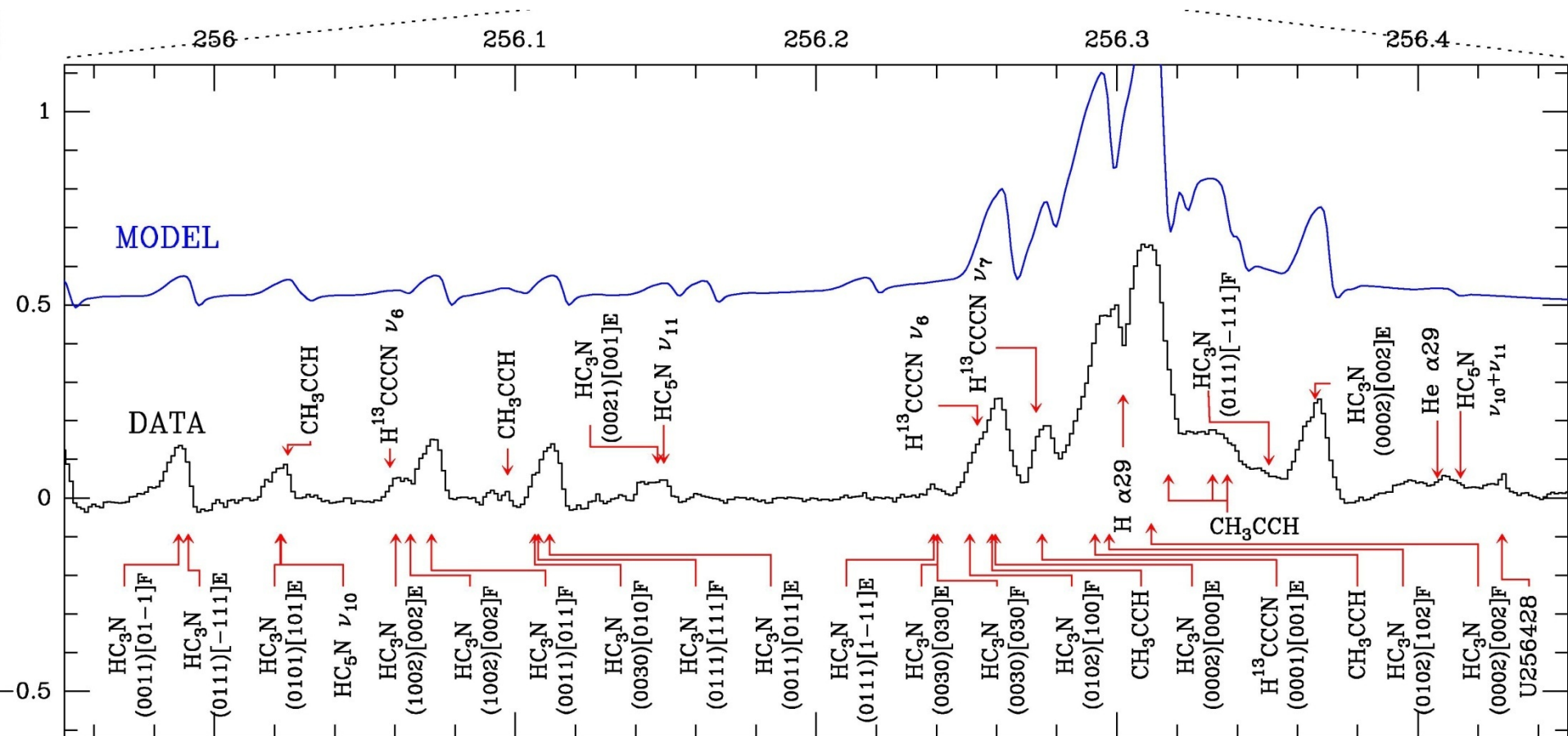
## 2 mm window : Data and final model



# 1.3 mm window : Data & final model

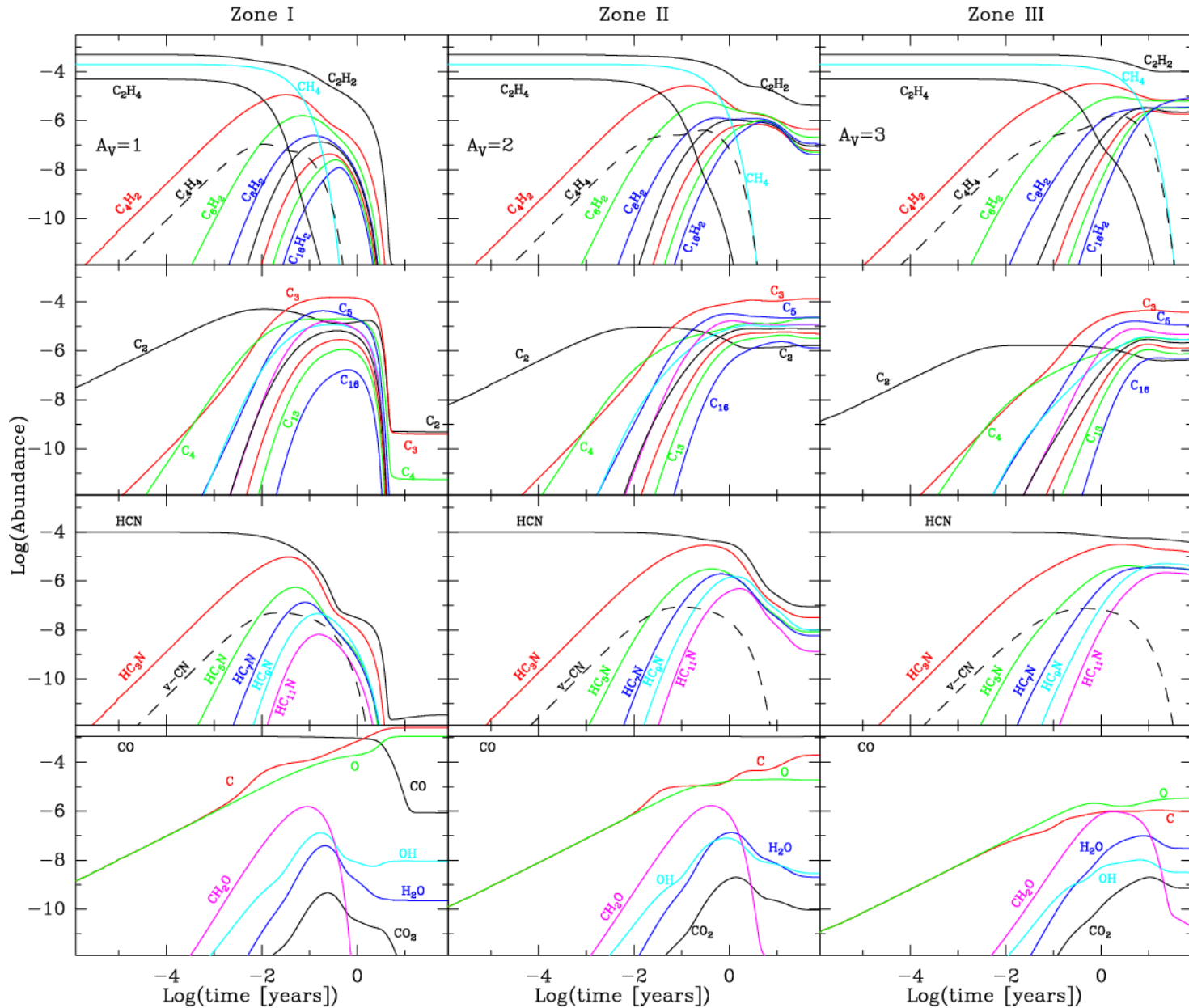


## 1 mm window : Data & final model

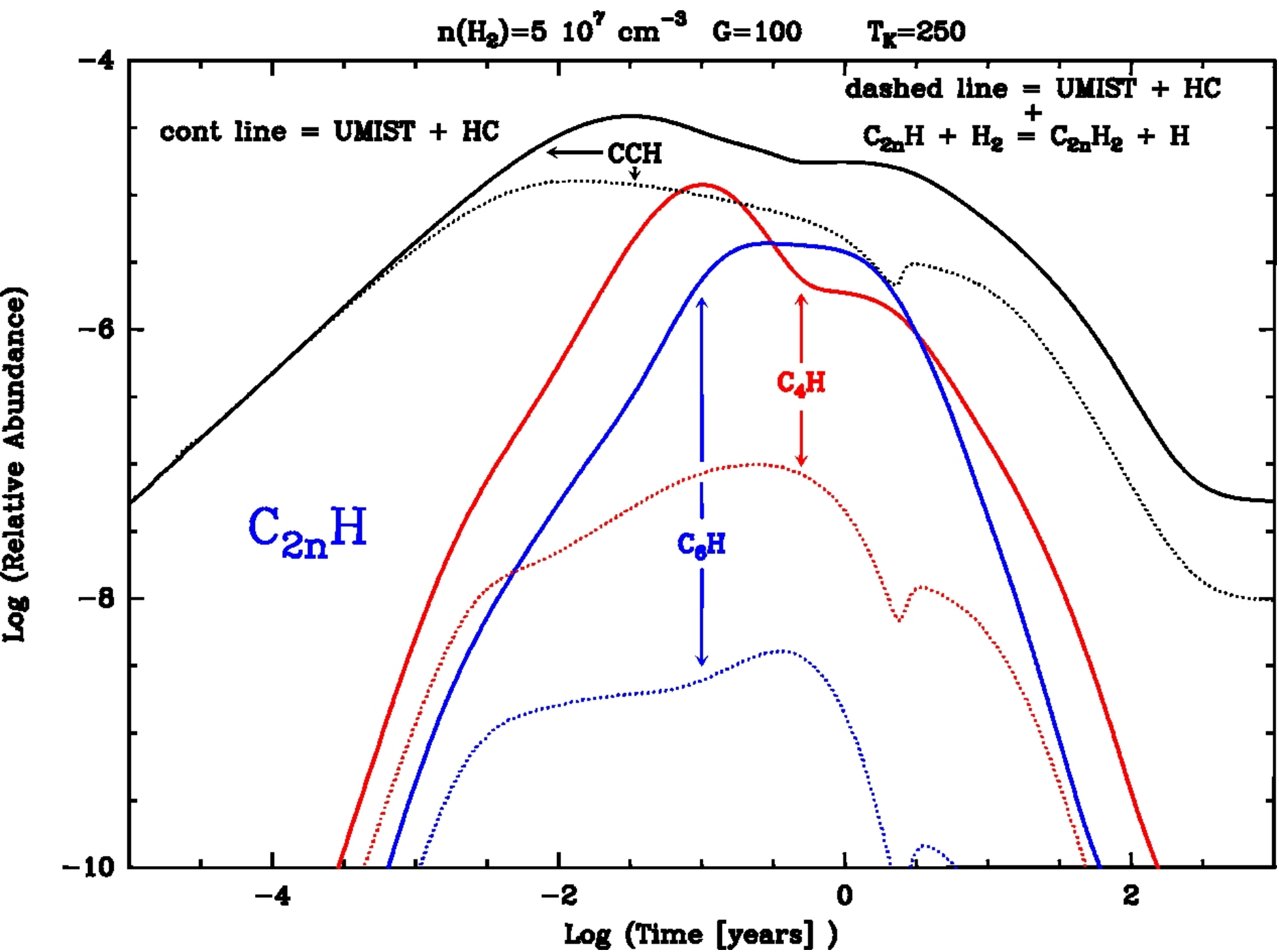


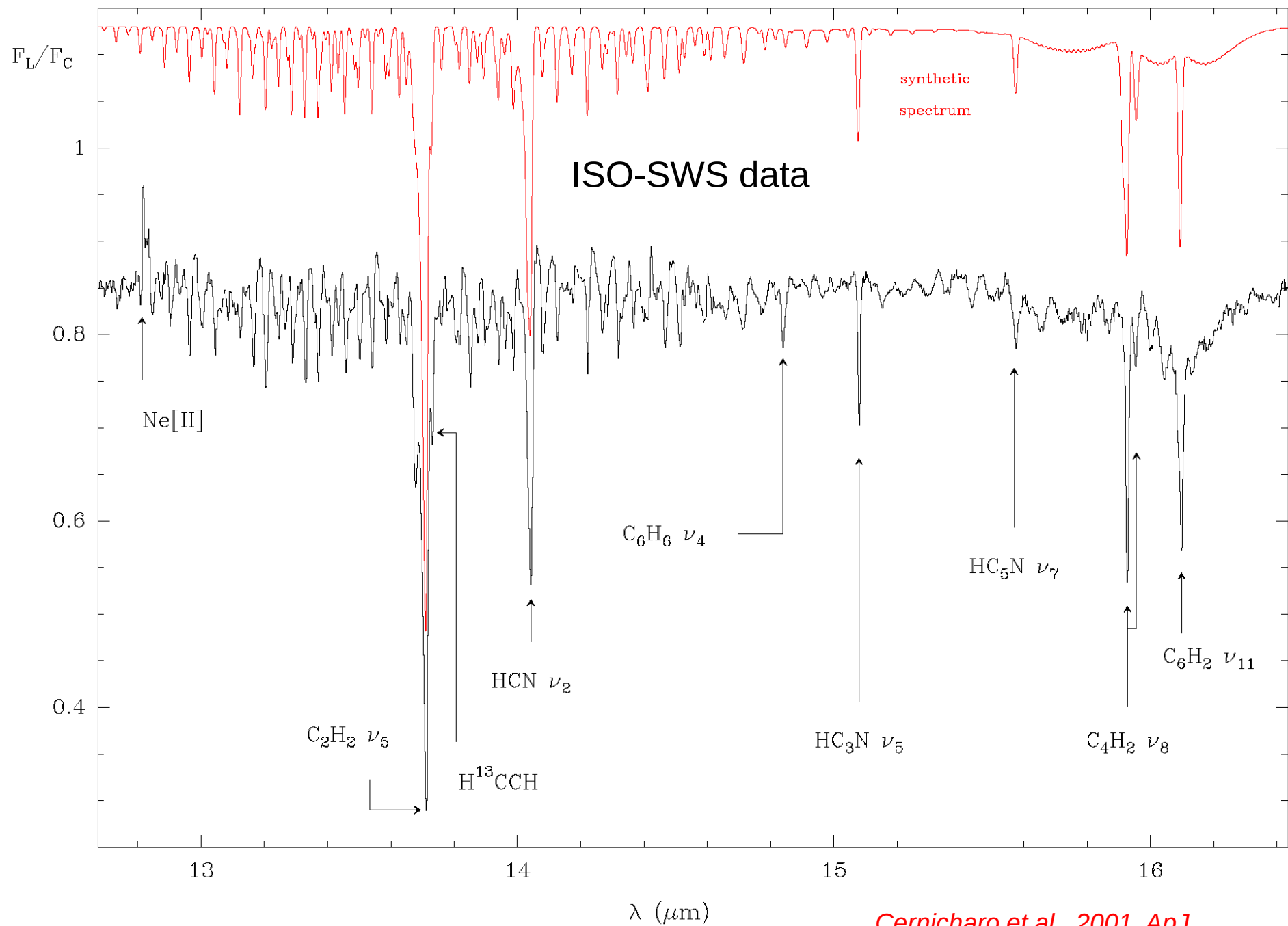
# Looking for new molecules : Modelling

Chemical  
modelling  
specific  
to a C-rich  
PDR



**Cernicharo**  
2004,  
*ApJ*, 608, L41





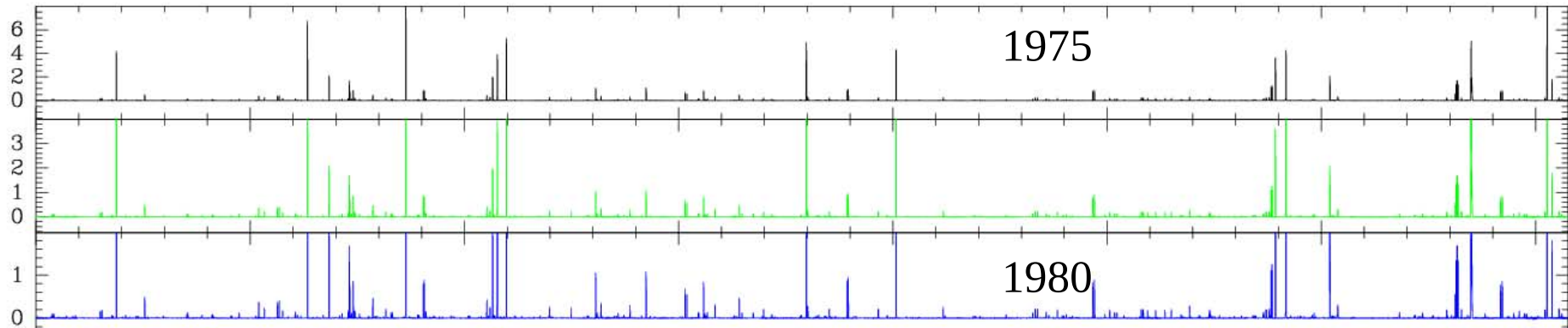
# LINE SURVEYS : ASTROCHEMISTRY WITH ALMA

- The need to know essential molecular parameters to interpret observations appear as a mandatory step in any observation of the Interstellar and Circumstellar media.
- Many molecules detected in space are exotic species difficult to be produced and observed in the Earth laboratories ( $C_nH$ ,  $C_nN$ ,  $C_nH^-$ ,  $C_nN^-$ ,...). Special physical conditions have to be reproduced. **Spectroscopy is needed**
- In most cases astrophysical environments are out of equilibrium (physical and chemical). **Collisional rates between molecules and  $H_2/He/H/e^-$  are needed.**
- Needs to be identified before we can progress in the study of chemical complexity and improve our chemical models. **Reaction rates are needed**

# THE SPECTROSCOPY PROBLEM

- Weeds
- How to deal with future ALMA data ?
- What we need from laboratory groups ?
- Which direction have we to follow ?
  - => high frequency (ALMA) => Physical processes
  - => Low frequency (GBT, VLA, SKA) => Heavy species ?

## 3mm line survey of IRC+10216 -30m IRAM telescope-



- What we could expect from line surveys ?
- Why we want to carry out line surveys ?
- What we need to interpret ALMA & Herschel line surveys ?

2007

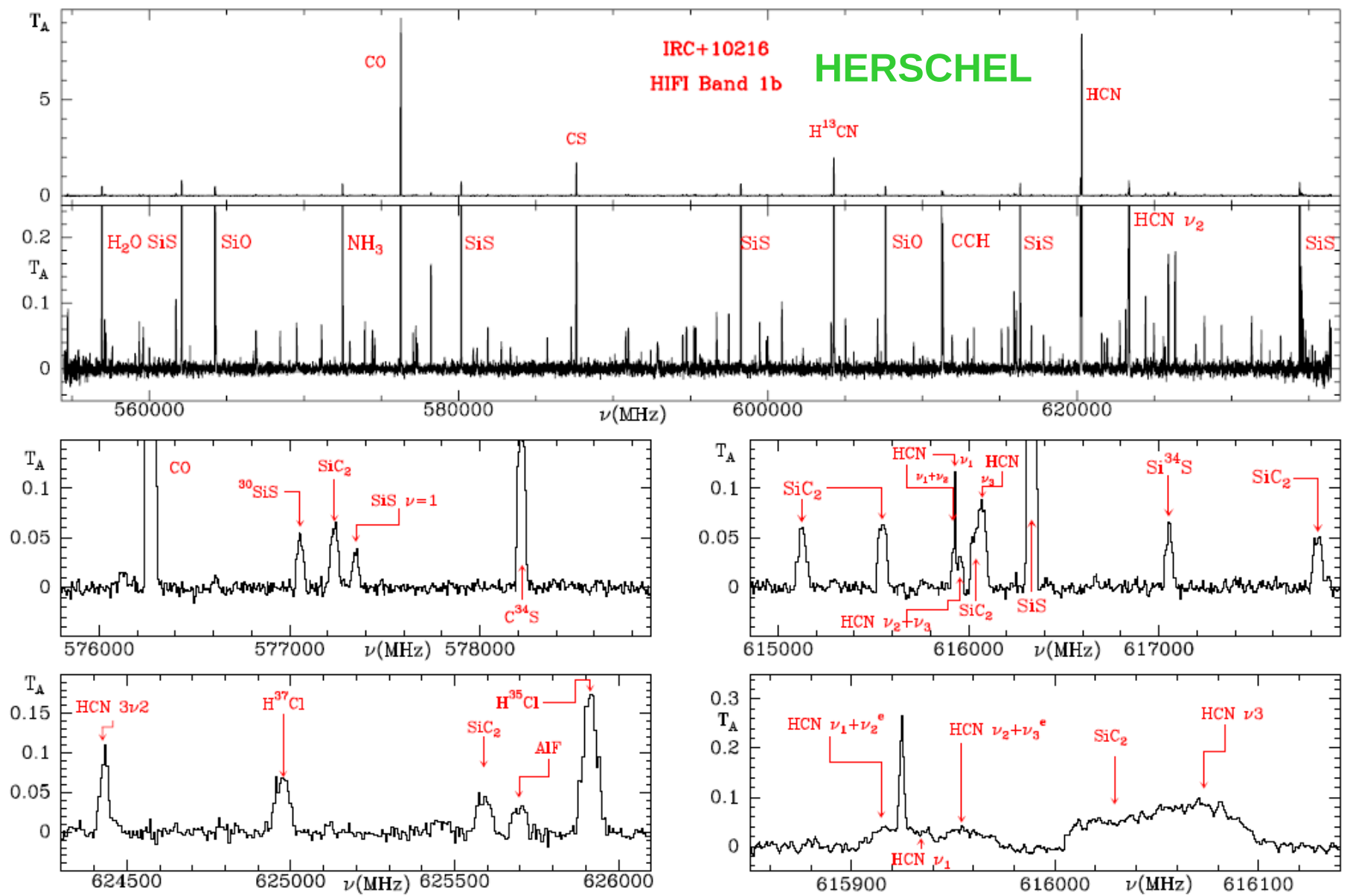


Fig. 1. Spectra of IRC +10216 observed with HIFI band 1b. The two upper panels present the complete spectrum on two different intensity scales. The panels below show different 3 GHz wide ranges of the survey. All data have been smoothed to a spectral resolution of  $2.8 \text{ km s}^{-1}$  except for the right bottom panel, which shows the spectrum around several vibrational lines of HCN with the nominal WBS resolution ( $1.1 \text{ MHz}$ ,  $\approx 0.5 \text{ km s}^{-1}$ ).

**Cernicharo et al., 2010, A&A, 521, L8. SiC<sub>2</sub> spectroscopy from space.**

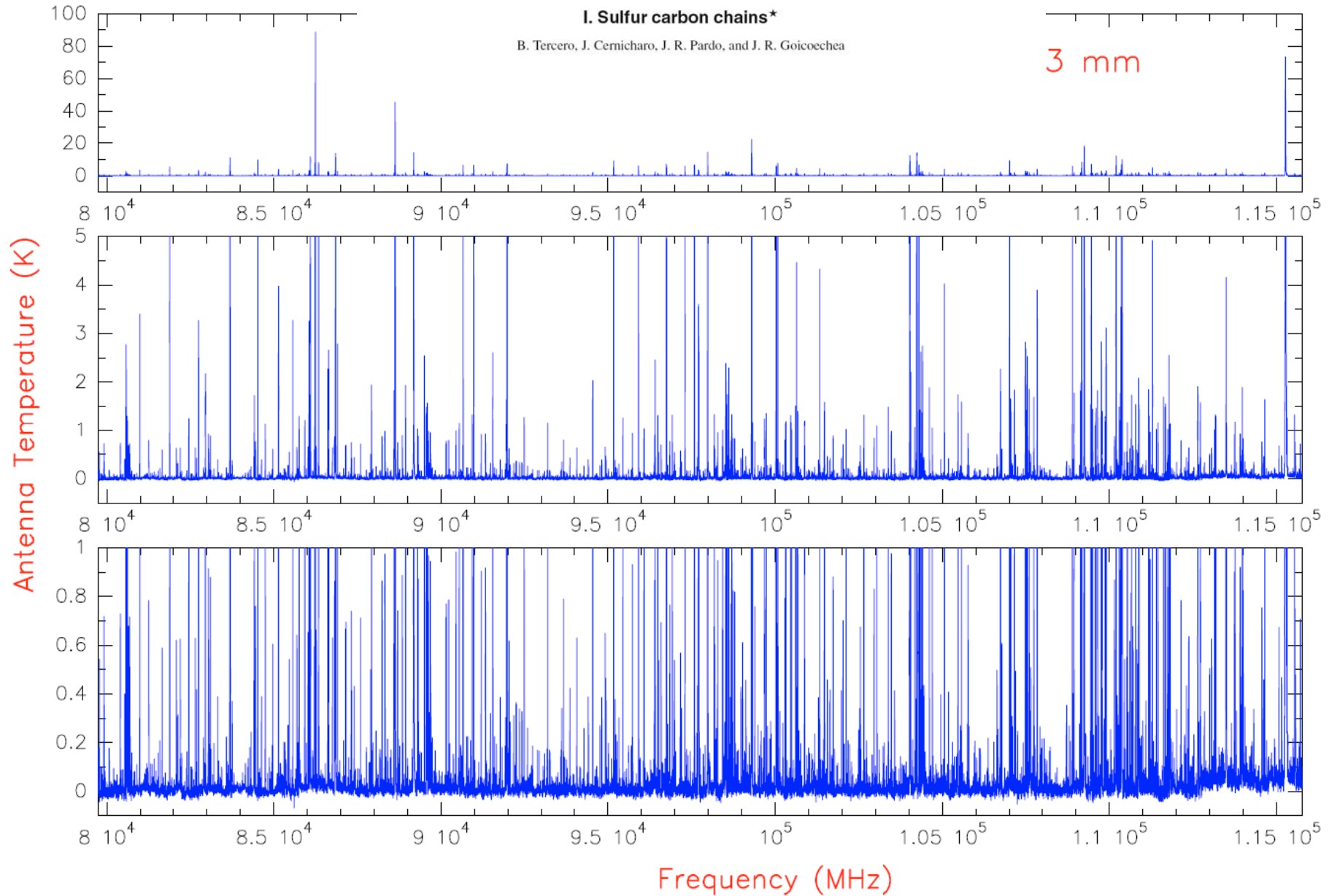
**HCl has been detected with PACS/SPIRE (Cernicharo et al., 2010, A&A, 518, L136)**

## A line confusion limited millimeter survey of Orion KL

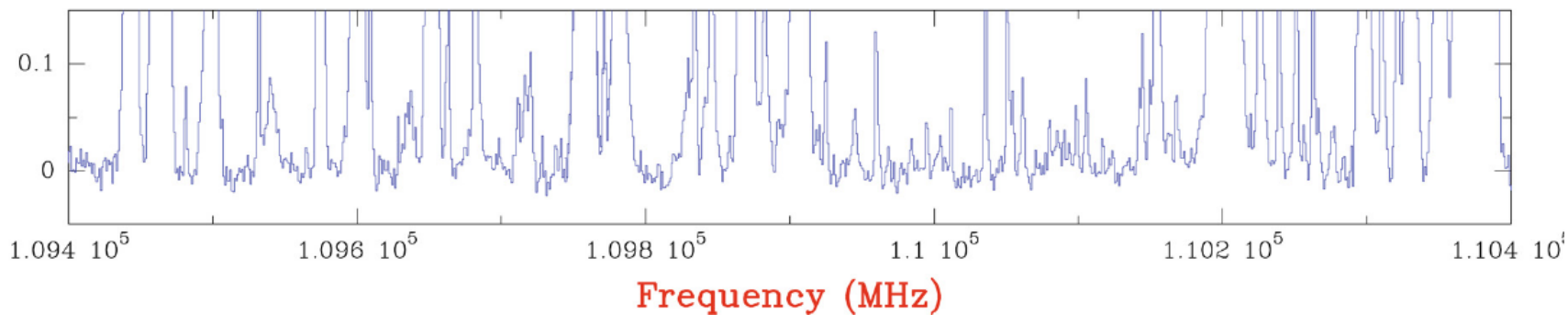
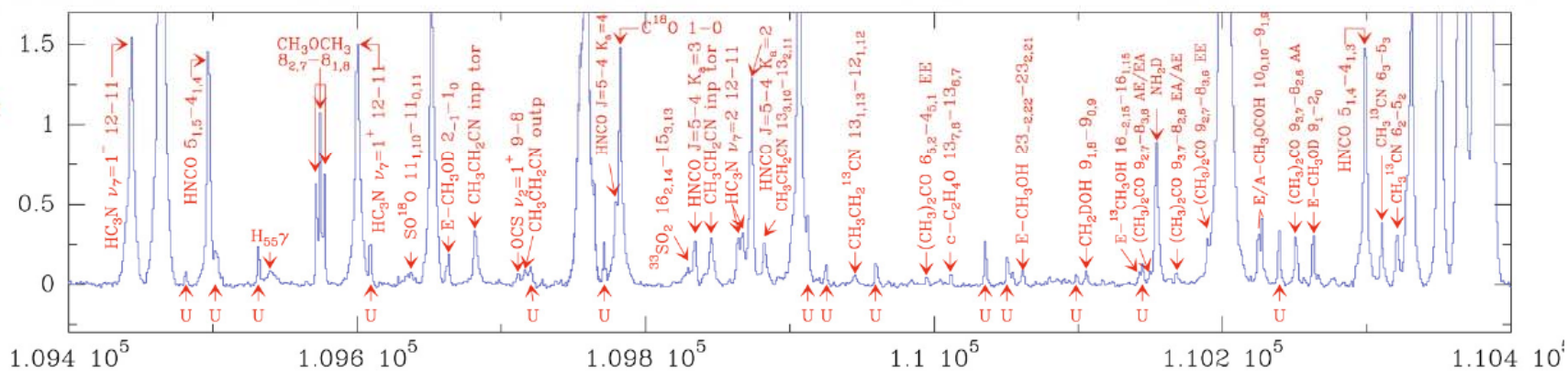
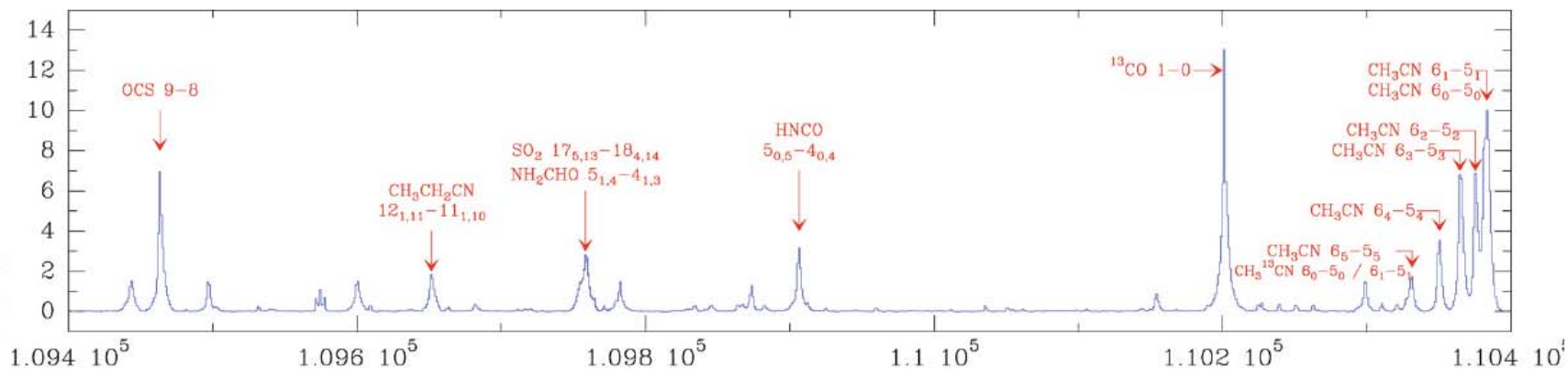
### I. Sulfur carbon chains\*

B. Tercero, J. Cernicharo, J. R. Pardo, and J. R. Goicoechea

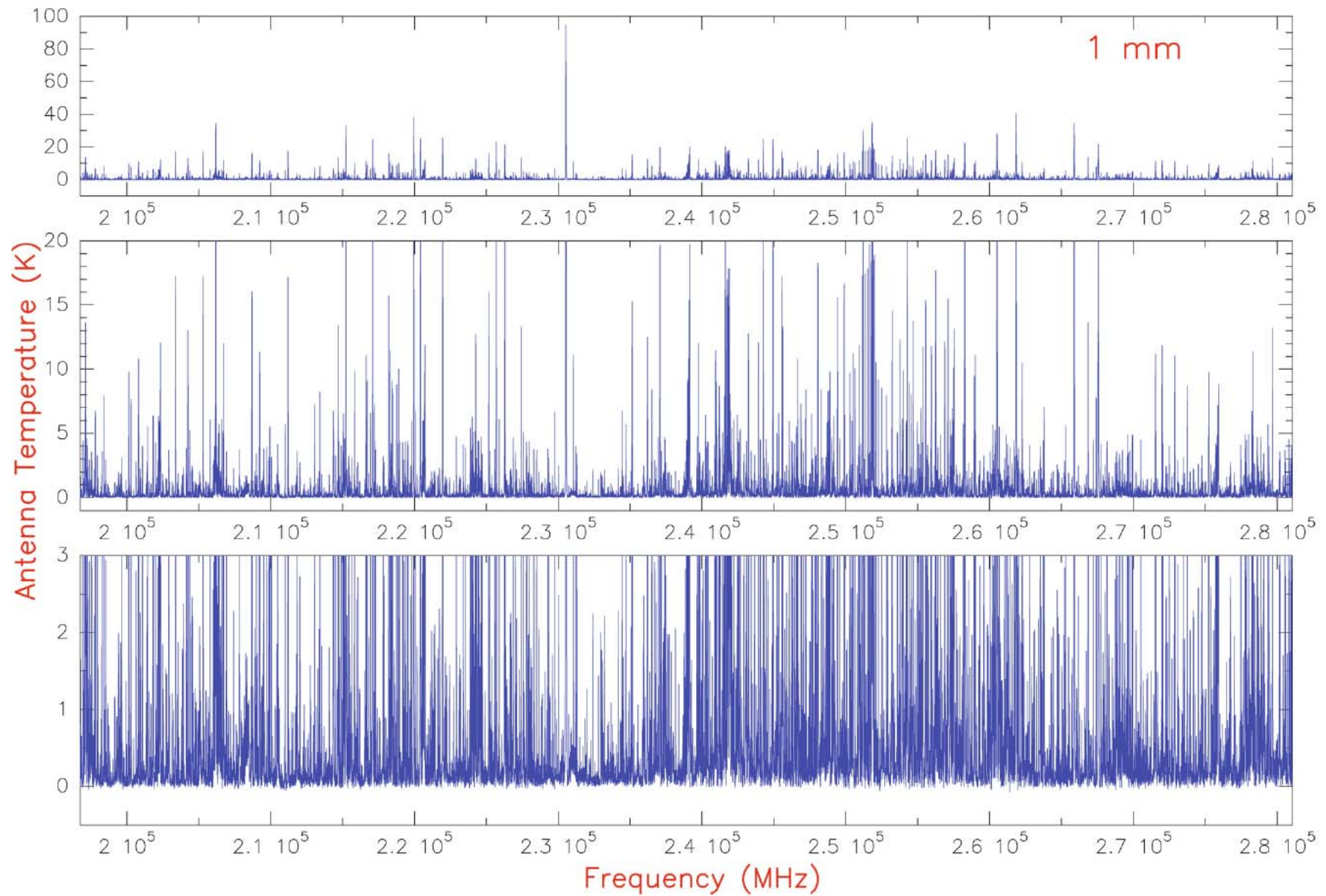
3 mm



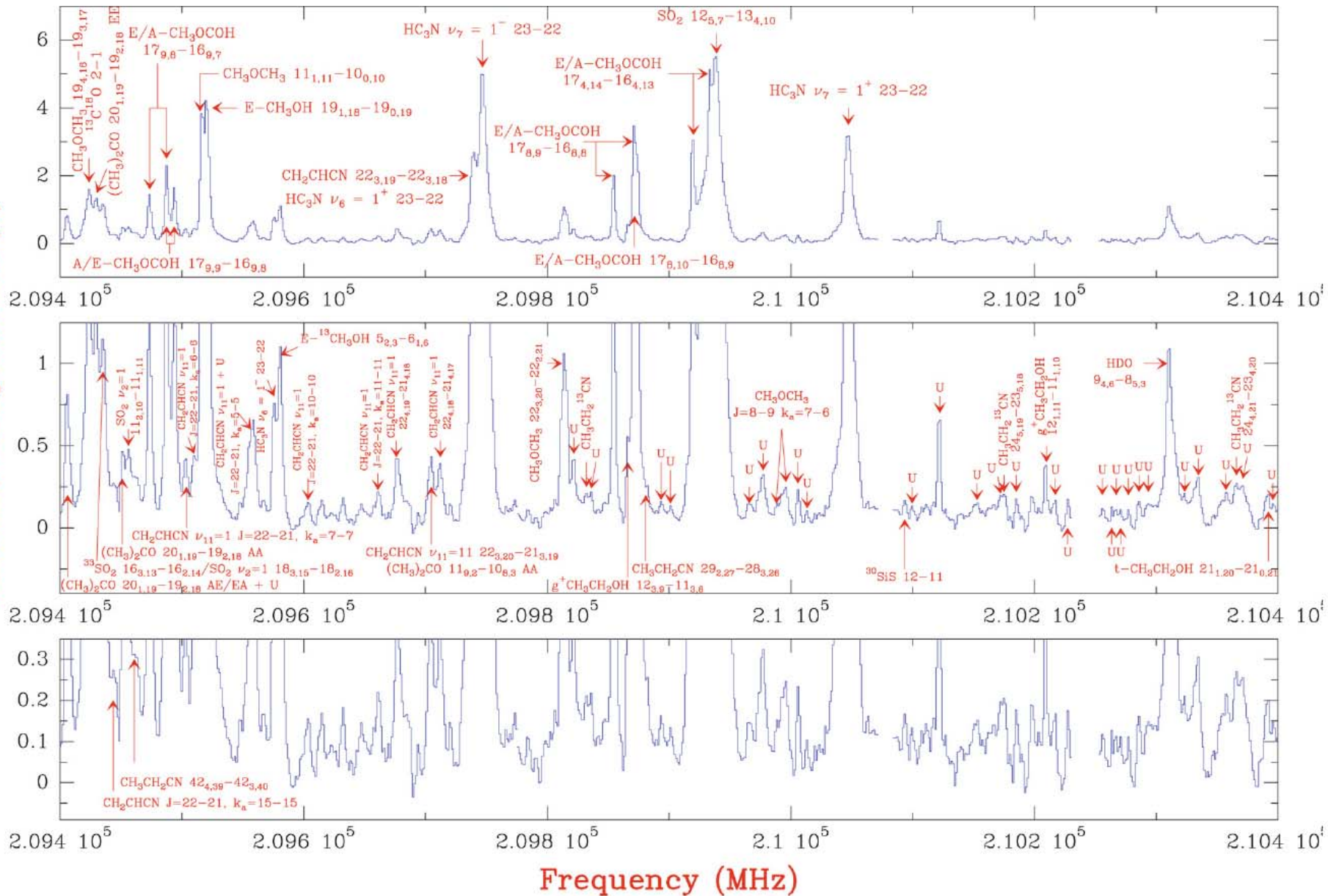
Antenna Temperature (K)



Frequency (MHz)



Antenna Temperature (K)



**LINE CONFUSION LIMIT REACHED. WHAT TO DO ?**

**ALMA WILL BE 7 times more sensitive than the 30-m telescope !!!**

Orion as seen with 30-m IRAM Telescope.  
10 min observing time/GHz

*ALMA will see hundreds of Orions*

*How to identify new species ?  
(Chemical Complexity)*

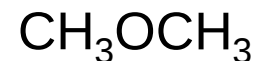
*Isotopes for most known species will be more  
abundant than unknown species !!*

*A real challenge !!!!*

*We need to maintain a very fruitful and close collaboration  
with chemists and physicists  
(laboratory and ab initio)*

Vibrational states

Isotopes



New species :

Alcohols, X-CN,

Ethers, (X-O-Y)

Cetones (X-CO-Y)

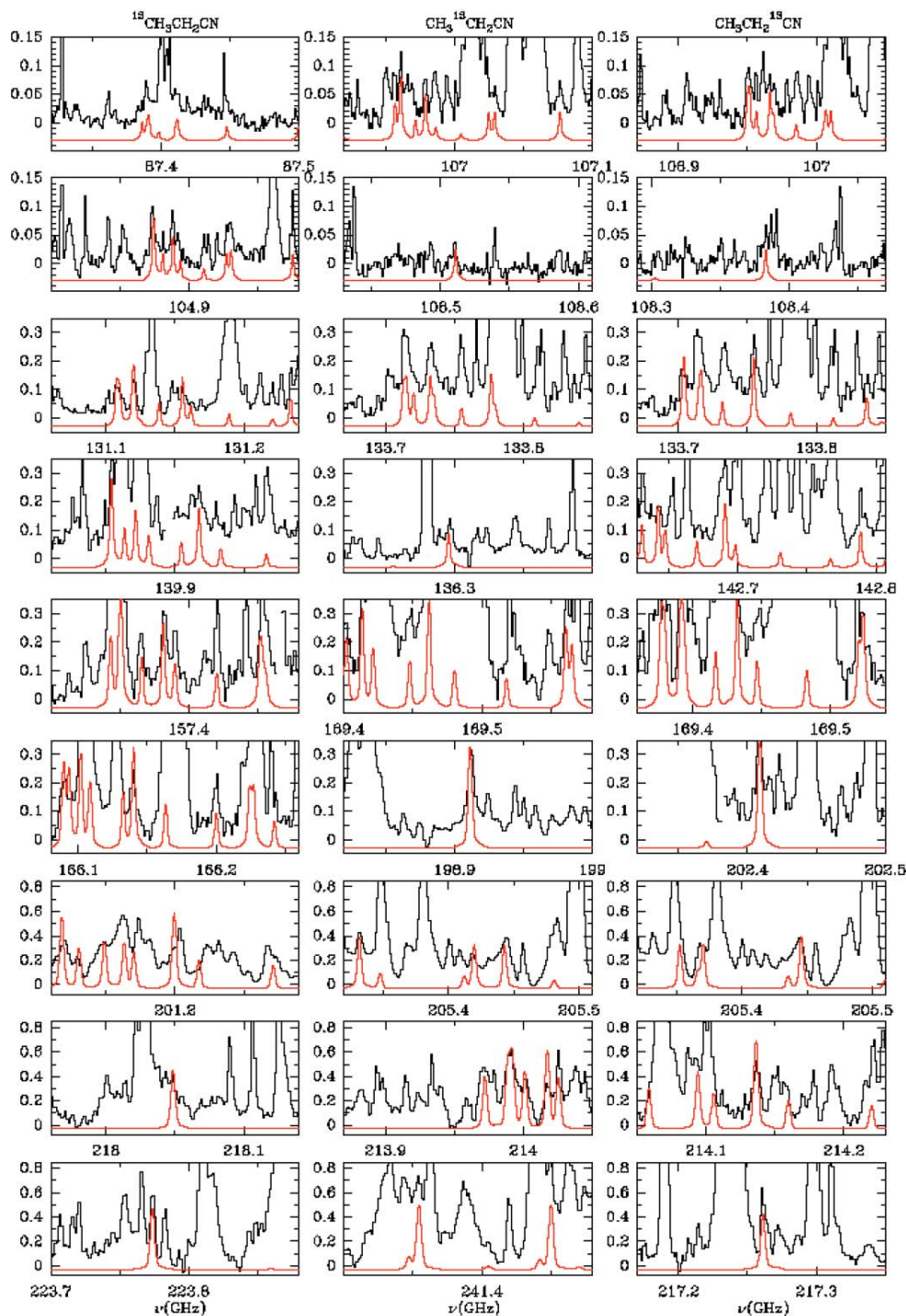
Isomers, ...

# Isotopic ethyl cyanide $^{13}\text{CH}_3\text{CH}_2\text{CN}$ , $\text{CH}_3^{13}\text{CH}_2\text{CN}$ , and $\text{CH}_3\text{CH}_2^{13}\text{CN}$ : laboratory rotational spectrum and detection in Orion<sup>★,★★</sup>

K. Demyk<sup>1</sup>, H. Mäder<sup>2</sup>, B. Tercero<sup>3</sup>, J. Cernicharo<sup>3</sup>, J. Demaison<sup>1</sup>, L. Margulès<sup>1</sup>, M. Wegner<sup>2</sup>,  
 S. Keipert<sup>2</sup>, and M. Sheng<sup>2</sup>

**Table 2.** Spectroscopic constants of the ground vibrational state of  $^{13}\text{C}-\text{CH}_3\text{CH}_2\text{CN}$ , A-reduction.

|                          | $^{13}\text{CH}_3\text{CH}_2\text{CN}^{(a)}$ |             | $\text{CH}_3^{13}\text{CH}_2\text{CN}^{(b)}$ |             | $\text{CH}_3\text{CH}_2^{13}\text{CN}^{(c)}$ |             |
|--------------------------|----------------------------------------------|-------------|----------------------------------------------|-------------|----------------------------------------------|-------------|
|                          | Value                                        | Uncertainty | Value                                        | Uncertainty | Value                                        | Uncertainty |
| $A/\text{MHz}$           | 27 342.6503                                  | (20)        | 27 045.8630                                  | (18)        | 27 635.4303                                  | (19)        |
| $B/\text{MHz}$           | 4598.06 735                                  | (39)        | 4697.96 236                                  | (40)        | 4689.91 341                                  | (31)        |
| $C/\text{MHz}$           | 4133.74 505                                  | (37)        | 4207.10 003                                  | (35)        | 4214.77 921                                  | (29)        |
| $\Delta_J/\text{kHz}$    | 2.99 463                                     | (31)        | 2.99 599                                     | (45)        | 3.03 369                                     | (25)        |
| $\Delta_{JK}/\text{kHz}$ | −48.3778                                     | (23)        | −45.0189                                     | (17)        | −47.8322                                     | (20)        |
| $\Delta_K/\text{kHz}$    | 546.335                                      | (13)        | 525.801                                      | (17)        | 553.930                                      | (45)        |
| $\delta_J/\text{kHz}$    | 0.66 2503                                    | (56)        | 0.68 4149                                    | (84)        | 0.67 6665                                    | (36)        |
| $\delta_K/\text{kHz}$    | 12.097                                       | (14)        | 12.749                                       | (14)        | 12.5104                                      | (59)        |
| $\Phi_J/\text{Hz}$       | 0.010157                                     | (90)        | 0.01169                                      | (18)        | 0.009812                                     | (68)        |
| $\Phi_{JK}/\text{Hz}$    | −0.06061                                     | (89)        | 0.03828                                      | (13)        | −0.0852                                      | (35)        |
| $\Phi_{KJ}/\text{Hz}$    | −1.7894                                      | (28)        | −1.827                                       | (39)        | −1.637                                       | (12)        |
| $\Phi_K/\text{Hz}$       | 30.558                                       | (44)        | 27.880                                       | (64)        | 29.95                                        | (30)        |
| $\phi_J/\text{Hz}$       | 0.003662                                     | (18)        | 0.003886                                     | (52)        | 0.003647                                     | (17)        |
| $\phi_{JK}/\text{Hz}$    | 0.0798                                       | (83)        | 0.143                                        | (10)        | 0.1329                                       | (36)        |
| $\phi_K/\text{Hz}$       | 4.7551                                       | (fixed)     | 4.7551                                       | (fixed)     | 2.77                                         | (23)        |



**More than 800 lines from the isotopes  
of  $\text{CH}_3\text{CH}_2\text{CN}$**

**Around 600 lines from the vibrational  
excited states of ethyl cyanide**

**More than 400 lines from those of  
 $\text{CH}_3\text{OCOH}$**

**Around 800-1000 lines identified every  
2 years in Orion. All lines above  
confusion limit could be identified  
around 2020 !!!**

**Belen started her PhD based on  
this line survey in 2006.**

**When combined with HEXOS data=>  
Work for a long period  
(several Nigel's units)**

**ALMA ?**

## **Astrochemistry : the problem of collisional rates**

**Getting physical and chemical conditions from data requires a detailed study of the radiative transfer through the observed source.**

**Only a few molecules (see, e.g., BASECOL) have been studied in detail and in most cases in collisions with He.**

**When collisional rates are not available astronomers do a lot of poor assumptions : This molecule is isoelectronic to this one, then let us use the same collisional rates !!**

**For example HNC and HCN. A rather simple case but  $\text{HNC}/\text{HCN} > 1$  in cold dark clouds. Chemical models have problems to explain this result. However,...**

**Detailed comparisons between species are only possible if collisional rates are available for both species**

**Even for an isotopologue and its mother molecule the rates can be very different ( $\text{H}_2\text{O}$  and  $\text{HDO}$ ;  $\text{CH}_3\text{OH}$ ,  $\text{CH}_2\text{DOH}$  and  $\text{CH}_3\text{OD}$ ,...)**

**Source structure and isotopic shifts in the frequencies of vibrational and electronic transitions can affect the excitation of the molecule and even its chemistry (UV selfshielding).**

**The calculation of a the collisional rates for a complex molecule can take a long time, even with the faster and cheaper computers we have nowadays**

**We rely 100% on chemical-physics groups accepting to do the job in close collaboration with astronomers.**

**(When quoting databases, please, quote also the sources !!)**